





Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

OWNERSHIP STATEMENT

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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Version history

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Date	Data points containing amendments or additions ¹ and	Document identifie and
	brief description	^v version number
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CA 7 FATE AND BEHAVIOUR IN THE ENVIRONMENT

Information on the updated dossier for the Annex I Renewal

Data on the fate and behaviour of isoxaflutole (RPA 201772) in soil, water and air were submitted within the EU Basic Dossier for Annex I inclusion in the year 1995. This pocument therefore the on those environmental fate studies which were not submitted within in the EU Basic Dossier

For a better overview existing data and their evaluation resulting from the process of Annex inclusion are summarized shortly amended by new data generated in Order of current fulfil requirements. The numbering and the headlines correspond to latest EV requirements

were performed with the fan The studies investigating into the environment following position of ¹⁴C-radiolabel:

diketoniurile metabolite RPA 202248 labelled in the In addition, studies cyclopropyl ring

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CA 7.1 Fate and behaviour in soil

CA 7.1.1 Route of degradation in soil

CA 7.1.1.1 Aerobic degradation Documents given in grey text were submitted within the EU Basic Dossier for Annex Linclusion 1995.

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Report:	X;	;	;199@M-158#35-0
Title:	RPA201772: Aerobic soil metabo	lism (
Report No:	R000347		A A O A
Document No:	M-158435-01-1	~ . Ŭ [*]	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
Guidelines:	USEPA (=EPA): N, 162-1, (198	2); Deviation not specified	
GLP/GEP:	yes 🔗		S .

Report:	•; (1995; M-213) 10-01 A
Title:	Herbicides: RPA 201772 Route of degradation (acrobic metabolism) in one soil interim
	report) Q V X X X X X Y
Report No:	C022444
Document No:	M-213110-001 6 6 6 6
Guidelines:	EU (=EE6): annex I, section 7.1 (1995); Deviation not pecified
GLP/GEP:	yes the way of the of t

Report:	d;
Title:	Herbicides Isoxaflutole Route Degradation (Aerobic Metabolism) in one Soil
Report No:	
Document No:	M-240821-0 1 3 3 0
Guidelines:	S Deviation wot specified > S & A S
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Title:	, &PA 20	1)772: Rate of D	egrada Öðn unde	ř Aerobic Con	ditions in Three	Soil Types
Report No:	B00382	7 🛫 Ö	47 4	Å		
Document No:	N_2408	22@1-1	<u>, 7</u> 0'	~		
Guidelines:		EC): 71/31/VI/9	4-EN rey 1; FA	O: Draft Rev	.3, Part 1,1.1;D	eviation not
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GLP/GEP: ^{AQ}	yes	N N	6 G			

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Report:	ö; ;2012;M-427583-01
Title:	Cyclopropy - Y-14C APA 202248:: Aerobic degradation/metabolism in one European soil
Repért No:	ME\$211/486 Q ~~~
Document No:	M-427589-01-1
Guidelines:	ØECD307; BSEPA4QPPTS 835.4100 and OPPTS 835.4200; EU 95/36/EC amending
	91/4 9/EEC; not specified
GLP/GEP:	yeð v v



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The <u>route of degradation in aerobic soil</u> had been investigated under laboratory conditions in three studies:

- two soils at 21°C and moisture at 75% of the field capacity at 0.33 bar following application of phenyl-UL-¹⁴C-labeled active substance (KCA 7.1.1.1/01);
- one soil under standard conditions at 20°C and moisture at 45% maximum water holding capacity (MWHC) after application of phenyl-UL-¹⁴C-labeled active substance (KCA 7.1.1.1 /02) and KCA 7.1.1.1 /03);
- three soils under standard conditions at 20°C and mosture at 40 % MWHC following applications of phenyl-UL-¹⁴C-labeled active substance (KCA 7.1.1.1/04).

The data requirement was addressed under Point (1.1.1.1 of the Dossier submitted and evaluated within the process of evaluation for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments (February

The evaluation revealed that the degradation of isoxaflutole predominantly proceeded via ring opening at the isoxazole heterocyclic ring system and rearrangement in the formation of the major (>10% AR) and predominant metabolite diketomicile RPA 202248. The step can be induced by chemical and biotical processes. The biotic degradation of RPA 202248 resulted in the formation of berzoic acid RPA 203328 as a major metabolite. Finally, metabolite RPA 205894 was observed at trace level in aerobic soil degradation tests. The degradation in apobic soil was accompanied by formation of nonextractable residues (NER) while the extent of mineralization was low under the conditions of the laboratory.

At the time of review for Annex I inclusion metabolites PPA 202248 and RPA 203328 occurred as major compounds in tests on route of degradation in aerobic soil. The two compounds were thus considered as residues in the environmental risk assessments for soil, ground water and surface water.

For current risk assessments no further metabolites were considered when following the introduction of new data requirements including new trigger values as laid out in Commission Regulation 283/2013 amending Regulation 1107/2009

2

The metabolic pathway in aerobic soil under conditions of the laboratory is summarized in Figure 7.1.1.1.

In order to demonstrate the degradability of potential residues originating from the cyclopropane ring in aerobic sol, a study was performed starting from the rapidly formed metabolite diketonitrile RPA 202248 rather than from the correspondingly lakeled active substance (KCA 7.1.1.1 /05).

Invalidation of study KCA 7.1 1 /010

During the review in preparation for Annex I Renewal it was found that extraction of soil samples was performed by applying several successive extraction steps in the course of the study. These included ambient extraction and up to two further extraction steps, i.e. extraction with acidified solvent and further extraction methods like Soxhlet extraction. The latter steps of extraction started for both soils either after 14 days of incubation (sandy loam) or straight after application at day zero (clay) till study end. For example, the levels of radioactivity amounted to 8.5% of applied radioactivity for the sandy loam soil by day 14 or 9.1% for the clay soil by day zero to be 16.0% (sandy loam) and 10.5% (clay) after 12 months of incubation.

However, the radioactivity in these additional extracts was not subject to analysis. Therefore their contribution to the overall distribution of components in the extractable portion of radioactivity formed must be unknown. Consequently, there is no adequate description of the total pattern of metabolites possible for this study.

Study KCA 7.1.1.1 /01 is therefore regarded as invalid with no consideration within the process of





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Report:	d; ;2012;M-427583-01		Õ
Title:	[Cyclopropyl-1-14C]RPA 202248:: Aerobic degradat	ion/metabolism in one European	sø
Report No:	MEF-11/486		
Document No:	M-427583-01-1		ク
Guidelines:	OECD 307; US EPA OPPTS 835.4100 and OPPTS 91/414/EEC up t specified	835.4200; EU 95/36/EC amend	ling
	91/414/LEC;not specified		4
GLP/GEP:	yes (冷	L L L	

Executive Summary

The degradation of [cyclopropyl-1-¹⁴C]-RPA 202248 was investigated in one European soil under aerobic conditions by incubation in the dark at 20,9°C and a soil moisture of 55% of MWHC for 91 days in maximum. The test was performed at west concentration of 0.533 mg test substance/kg soil, equivalent to a field application rate of 200 g a.s./ha

Recovery of radioactivity was 96.6% 3.7% Total extractable radioactivity dereased from 901.5% on day zero to 5.9% on day 91. The decrease of extractable radioactivity was accompanied by the formation of non-extractable residues (NER) to account for 38.9% AR after 91 days. Mineralisation was significant to account for 47.5 % of AR petermined as C-cathon doxide by day 91. Formation of other organic volatile components was insignificant ($\leq 61\%$ ABY.

As a result of rapid biotransformation in soil no other formation of restabolites than 14C-carbon dioxide was observed at a significant level in the course of the study.

The biotic character of degradation of [46]-RPA 202248 in crobic soil was clearly underlined by the formation of 14C-carbon Choxide as the major product of conversion along with the formation of nonextractable (bound) residues.

The degradation of cyclopropyl 4-14Cl-RPA 202248 and aerobic soft was fast to result in a half-life of 14 days based on an SFO t Ot kingst model.

. Material an Methods

- A. Materials
- 1. Test Material: [Cyclopropyl_1] X 2022 48 (AE 0540092, BCS-AB59005) C]RR
 - Specific radioactivity: 4.08 MBq/@g (110.26 µCi/mg)
- 2. Soil:

Ine soil bad been freshov collected from the field followed by sieving The physico-chemical characteristics are summarized in Table 7.1.1.1-1. The soil had been freshow collected from the field followed by sieving to 2 mm.

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B. Study design

1. Experimental conditions: Samples of 100 g dry weight of soil each were filled into glass incubation flasks and pre-equilibrated prior to treatment at approximate study conditions (darkness, 20 °C, moisture content of 55% MWHC) for 4 days. At start, each sample received 0.552 mg test substance/kg soil, a dose representing a field rate of 200 g a.s./ha. Following application the samples were attached to 'static' systems with traps to collect ¹⁴C-carbon dioxide and other volatile components. Samples were incubated at 20 ± 1 °C and a moisture content of 55% MWHC in the dark for 91 days in maximum. In addition, untreated soil samples were incubated under the same conditions of or determination of soil microbial activity.

2. Sampling: Duplicate samples were removed for work-up after 0, 4, 3, 7, 14, 30, 58 and 91 days of incubation. Samples for determination of soil microbial biomass were investigated after 0 and 91 days of incubation. The complete samples were immediately processed by extraction and HPLC analysis was usually performed the same day. Therefore no additional investigations of storage stability were necessary.

3. Analytical procedures: The entire soft sample in each test vessel was processed by a stepwise extraction procedure. The initial step was performed with 80 mL aqueous acetonitrile solution (4:1, v/v) three times successively by shaking the soil/softvent thixture at ambient temperature for 30 min. After separation by centrifugation the soft was extracted again with the same volume of the same solvent mixture twice successively by heating in a microwave extractor at 70°C for 10 min with stirring, followed by centrifugation. Microwave and ambient extracts, were combined and an aliquot (10 mL) concentrated to a small volume (2mL) prior to analysis.

The ¹⁴C-material balance was established for each sample by extraction analysis of volatiles and combustion of con-extractable residues. Following quantitation of radioactivity in extracts by LSC, analysis was performed by reversed phase HPCC and ¹⁴C-flow-through detection techniques. The determination of non-extractable residues (NER) was performed by combustion/LSC of aliquots of the air-dried extracted soil.

The LOQ of the HPLO analytical method was estimated to be 9.9% AR on the basis of the LOD of the radio-detector and based on the smallest peaks observed in various chromatograms in the course of the study.

C. Determination of degradation kinetics: Degradation data were kinetically evaluated by use of the software KinGui, version 1.1. Following calculations of fits with kinetic models SFO, FOMC and DFOR, the best fit was evaluated to be a minimum in the significance test and the error of chi-square (χ^2) , followed by visual assessment.

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II. Results and Discussion

A. Data: The results of aerobic biotransformation of [cyclopropyl-1-¹⁴C]RPA 202248 after inculation in one European soil are summarized in Table 7.1.1.1-2.

 Table 7.1.1.1-2: Degradation of [cyclopropyl-1-14C]RPA 202248 in silt loam soil conditions (mean ± SD)

							<u>^</u>	0 0	
				ÖSam	pling int	érval (da	ys) 🔊		Ĵ ⁴ .
Component		0	1	3	70%	14	×\$0	58	ç ⁹ 1 0
RPA 202248	Mean*	100.9	950	91.7	\$6.8	57.3 ×	17.8	[*] 3.30°	2.j
	SD	±1.2	&≇0.6	±1.2	$\neq \pm 0.5$	±0.0	É.ر	±0.0	+0.2
Total other unidentified**	Mean*	0.5 Қ	0.0	0,60	0.0	×Q_10	3 .4	∕≫4.2 `	[©] 3.8
	SD	±0.0	±0,0	£0 .0	@ 0.0 (00°±0.0	0.1%	±0,4	±0 _f 1°
Total extractable	Mean*	A01.5	~95.0 <i>_</i>	91.7	76.8	57	21.2	A .7	.9
radioactivity	SD @	$2 \pm 0 7 $	$\pm 0 \mathscr{O}$	±1.2	±0.5	× 0.0	⊊£0.6	0.2^{4}	10.0
Non-extractable	Mean	04.7	\$ 3	\$8.3	୍ଲ୍ 17.8 ୍କ୍	C27.0	42 A	40,9	38.9
radioactivity	SDO	±0.0	±0.1	±0.13	±0,00	±QQ	±0.4	∘ <u></u> ≜Ø.3	±0.3
1400	⊘ ean*≪	0.00	0.3	00)	3.17	©1.6 ≈	S30.2%	43.2	47.5
	SD SD	±Q ≈0	±0.0	£0.0	©±0.0 ¢	±0.4	±0.\$	±0.4	± 0.0
Other veletiles	Mon*	ي 0.0	Ø<0.1	©″<0.1	<0 <u>.</u>	ૢૼ૾૽ૢૢ૽ૠૼ	9 .1	<0.1	< 0.1
Ouler volatiles	,≴D	±0.05	±000	±0°0	¢±0.0	₩0.0	$S \pm 0.0$	±0.0	± 0.0
Total radioactivity (0	Mean*	102.1	\$ 9 8.5	00.9 ⁽	0°97.30°	95,0	93.8	91.8	92.3
	SP X	±0.7			±\$%.1	\$0 .4	±0.5	±0.5	±0.3

Values given as prcentages of initially applied radioactivity

SD = standard deviation, * Mean values of two replicates; ** Distributed intersix components in maximum with none of the individual components observed at more than 12% AB in the source of the study

 \bigcirc

B. Mass balance: The total material balances of radioactivity showed a complete recovery to range from 91.8 – 102 % AB. The results we summarised in more detail in Table 7.1.1.1-3. Conclusively there were no signs for losses of radioactivity from sample work-up and processing.

Table 7.1.1.1-3: Total material balances of radioactivity of 14C-202248 in one EU soil

4		o
<i>V</i>	Soil	
, to	Total Recovery (%AR) &	91.8 - 102.1
\sim	Mean (% AR)	96.6
	Rel. standard deviation	3.7

Values given as percentages of initially applied radioactivity

۸Ô

C. Bound and extractable residues: Values of extractable radioactivity decreased rapidly with time accompanied by foundation of non-extractable residues to finally undergo ultimate degradation as summarized in Table 7.4 1.1-4. Starting from a complete extractability given by day zero (101.5%) values decreased to 5.9% after the maximum incubation period of 91 days.

In turn, values for non-extractable radioactivity (NER) were low by day zero starting from 0.7% to increase to 42.4% by day 30 and to decrease to 38.9% at the last sampling interval (day 91).

Soil	Extractable re	sidues (%)	Non-extractable residues (%)		
	Day 0	Day 91	Day 0	Day 🎾 🖓	
	101.5 ±0.7	5.9 ±0.0	0.7 ± 0.0	3809 €0.3 5 5	

Values given as percentages of initially applied radioactivity.

D. Volatile radioactivity: ¹⁴C-RPA 202248 was extensively mineralised to ¹⁴C carbon dioxide to account for 47.5% AR at study end (day 91). Formation of other volatile radioactivity was insignificant at any sampling interval ($\leq 0.1\%$ AR)

E. Transformation of test substance: ¹⁴C-carbon disxide s as formed as the major and predominant transformation product observed in the course of the study. No single component was observed at a significant level while other unidentified components occurred at trace level of the course of the study. The biotic character of RPA 202248 degradation in acrobic soil is underlined by the formation of non-extractable (bound) residues that showed a decline but could not be converted fully to carbon dioxide during the runtime of the study.

F. Degradation kinetics: The degradation kinetics was evaluated by fitting of data to the three kinetic models SFO, FOMC (Gustafson-Holden) and DFOP¹ for the test substance only with the quality of fits assessed according to FOCUS kinetic guidance. The mitial concentration at time zero was included in the parameter optimisation. All data points were weighted equally. For optimal goodness of fit, the initial value was also allowed to be estimated by the model. The best-fit kinetic model was selected by applying the criteria for chi-square (χ^2) scaled-error to be a minimum and on the basis of visual assessment. The results of the kinetic evaluation are provided in Table 7.1.1.1-5.

The fits describing degradation of RPA 202248 in the soil resulted in low chi-square (χ^2) errors for all models applied with overall ranges of χ^2 errors being marginal with the SFO kinetic model to result also in a visually acceptable fit.

The degradation half-life of RKA 202248 was 14 days associated with a DT₉₀-value of 47 days.

0	· O	<u>à</u> V				
Soil 🦓		Kinetic	DTQ	DT90	Chi ² Err	Visual
, Ku		nodel	U (days)	(days)	(%)	assessment
	, ô	ŠFO (°¶14	47	5.1	+
	e se	FOM	<u>14</u>	48	5.6	+
	A A	DFQP	14	47	5.9	+

Table 7.1.14,5: Kinetics of aerobic degradation of RPA 202248 in one soil at 20°C

Best fits according to the criteria set are marked bold.

Visual assessment good methin; - bad

¹ SFO = Single first order; FOMC = First order multi compartment; DFOP = Double first order in parallel

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III. Conclusion

Following application of cyclopropyl-1-14C-RPA 202248 to one aerobic soil the formation of other metabolites than ¹⁴C-carbon dioxide was insignificant when considering its contribution investigations into the route of degradation of isoxaflutole residues in aerobic Soil.

CA 7.1.1.2	Anaerobic	degradation
------------	-----------	-------------

The degradation of c	yclopropyl-1-14C-RPA 202248 in one aerobic soil was fast to result in a half-the
of 14 days following	SFO as the best fit kinetic model.
Populta of kinatia ava	Justion of this study wars also read in report CA 21 2 DV /06 in order to derive
Results of kinetic eva	futuring of this study were also used in report KCA24.1.2.21 /0000 of the
modeling input paran	neters.
CA 7.1.1.2 Ana	erobic degradation \mathcal{N} \mathcal{N} \mathcal{L} \mathcal{N} \mathcal{N} \mathcal{N} \mathcal{N}
Report:	k; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;
Title:	RPA201772 Anaerobic Aquatic Metabolism
Report No:	R016760 ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~
Document No:	M-192288-0141 2 4 a 4 2 a
Guidelines:	UŞEPA (=EPA): Nº162-3 (1982); Deviation not specified
GLP/GEP:	Ver A 4 2 A A A A A A
oll/oll/	

1 had been investigated in one study under The route of degradation laboratory conditions in

one water-logged soil under standard conditions (20°C) following application of UL-¹⁴C-phenyl-labeled active substance ($K \neq 2, 1, 4, 2, 0$ labeled active substance (KCA 7.1 + 2 /01)

The data requirement was addressed under Point 7.1,1.1.2.1 of the Dossier submitted and evaluated within the process of evaluation for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its agrendments.

Consequently there is no detailed description of this existing data in this update.

The evaluation revealed that soxafluole degraded slowly under anaerobic conditions of the test to form the same pattern of metabolites as observed under aerobic conditions. The degradation of the active substance isoxaflutole therefore actuded the formation of the diketonitrile RPA 202248 as a major metabolite. Metabolite & PA 209834 was observed as a major component in addition, while this compound had been observed as a minor compound in aerobic soil. Its transient character was demonstrated by & decrease in the course of the test. Finally, the benzoic acid RPA 203328 was observed as a menor component.

No other methodites were observed at levels requiring further assessment as laid out in the actual data requirements according to Commission Regulation 283/2013 amending Regulation 1107/2009.

Based on the results it has been concluded that the route of anaerobic soil degradation is identical to that observed in aerobic soil.

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CA 7.1.1.3 Soil photolysis

Report:	§; ;1994;M-158351-(01 🔊	
Title:	RPA201772 - Soil photolysis	Ĺ,	
Report No:	R000311	O,	
Document No:	M-158351-01-1	2	
Guidelines:	USEPA: 161-3 (1982); not specified 🔊		
GLP/GEP:	yes Arr	Ĩ	\$ <u>\$</u> \$
	¥	20	

The <u>route of degradation on irradiated soil surfaces</u> had been investigated in one study under laboratory conditions in:

• one US soil under standard conditions (25°C, 75% of field capacity at 0.33 bar) following application of UL-¹⁴C-phenyl-labeled active substance and irradiation with artificial surplicity (xenon burner, cutoff of wave lengths <290 µm) for 31 days in maximum (KCA 7.1.1.3 /01)

The data requirement was addressed under Point 7.1.1 2.2 of the Dossier submitted and evaluated within the process of evaluation for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments with no detailed description here.

The evaluation revealed that isoxaflutole was rapidly degraded in irradiated samples and in dark controls under the conditions of the test. However the qualitative pattern (i.e. foute) of metabolites formed as well as their quantitative occurrence (i.e. tate) was practically the same for irradiated samples and for dark control samples included in parallel. Metabolites RPA 202248 and RPA 203328 were observed as the only major (>10% AR) components.

Conclusively, the degradation observed in irradiated samples was mainly due to the contribution of biotic degradation. Degradation, fates were similar for irradiated samples and dark controls to result in values for the half-life of 22.8 hours (0.95 days) for irradiated samples and 19.7 hours (0.82 days) for dark controls under the conditions of the test.

Consequently, photolytic processes to not complete significantly to the overall elimination of isoxaflutole residues from the soil environment.

No other metabolites were observed at levels requiring further assessment as laid out in the actual data requirements according to commission Regulation 283/2013 amending Regulation 1107/2009.

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- Rate of degradation in soil CA 7.1.2
- Laboratory studies CA 7.1.2.1
- CA 7.1.2.1.1 Aerobic degradation of the active substance

CA 7.1.2	Rate of degradation in soil
CA 7.1.2.1	Laboratory studies
CA 7.1.2.1.1	Aerobic degradation of the active substance
Report:	n; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;
Title:	RPA201772: Aerobic soil metabolism
Report No:	R000347
Document No:	M-158435-01-1
Guidelines:	USEPA (=EPA): N, 162-1, (1982); Deviation not specified
GLP/GEP:	yes yes

Report:	5; 995;M-213110-01 ~ ~ ~
Title:	Herbicides: RPA 201772 Route of degradation aerobic metabolism) in one son (interim
	report)
Report No:	C022444
Document No:	M-213110-01-1 (Petering report)
Guidelines:	EU (=EEC): amex I, section 41 (1995); Deviation not specified
GLP/GEP:	yes of o y y y y y y

Report:	6; 1996 (1996) 1-240 (1-01)
Title:	Herbicides: Isoxaflutore Route of Degradation (Aerobic Metabolism) in one Soil
Report No:	B003826 O S S S S S S
Document No:	M-240821-01-1 (Final report)
Guidelines:	Deviation not specified
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Title:	<u> </u>	RPA 201772	; Rate of Degra	dation under	Aerobic Cor	ditions in Three	e Soil Types
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Title: A RPA 20122: Rate of degradation under aerobic conditions in one soil at 10 degrees C
Report No. CO2244S
Document No: $\sqrt{M-213113-01}$
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Environmental, revision I; Deviation not specified
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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Report:	g; ;2013;M-464601-01	0
Title:	Isoxaflutole: Kinetic modelling evaluation of aerobic soil degradation study data to a	rive 📎
	trigger endpoints	Ş
Report No:	VC/13/007C	0
Document No:	M-464601-01-1	Ŋ
Guidelines:	EU Council Directive 91/414/EEC, as amended by Commission Directive 95/36/E	C of
	July 1995, Section 5, Point 7 and Commission Regulation (EC) No 1107/2009 of 2	1 🖏
	October 2009; not applicable	ď,
GLP/GEP:	no 🖉 û û ô	
	v Q. a X V	Š

Report:	n; 2013;M-464596-01 O Q O
Title:	Isoxaflutole: Kinetic modelling evaluation of aerobic soil degradation study data to derive
	modelling endpoints Q^{U} γ Q^{U} γ Q^{U}
Report No:	VC/13/007B
Document No:	M-464596-01-1
Guidelines:	EU Council Directive 91/414 CEC, as amended by Commission Directive 95/36/LC of
	July 1995, Section & Point 7 and Commission Regulation (EC) No 11072009 of 21
	October 2009; not applicable of L L A A A L
GLP/GEP:	no Q V V V V V Q Q Q

For the active substance isoxaflutole aerobic soil can be derived from studies performed under the following conditions:

- two soils at 21°C and moisture at 75% of the field capacity a 0.33 bar after application of phenyl-UL-14C-labelled active substance (KGA 7.1.2.1.1 /0);
- one soil under standard conditions at 20°C and mosture at 45 % maximum water holding capacity, (MWHC) after application of prenyl 4/L-14C-Tabelled active substance (KCA 7.1.2.1.1 /02 and KCA 7.12.1.1/93);
- three soils under standard conditions at 20% and poisture at 40% MWHC following application of phenyl-UL-14C-labelled active substance (KGA 7.1-271.1/04);
- one soil at 10°C and 40 20 MWHC and application of pheny UL-14C-labelled active substance (KCA 7.1.2.1.1005).

This data requirement had been addressed inder Bint 70.1.2 of the Dossier submitted and evaluated within the process of evaluation for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Following latest gridance on kinetic evaluation the data from existing studies have been re-evaluated therefore superseding the existing kinetic evaluations.

The evaluations were performed separately in order to derive values for the half-lives and the DT₉₀ for comparison with trigger endpoints and half-lives for modeling endpoints for input into environmental risk assessments.

As explained under KCA 7.1.1.1 in more detail, study KCA 7.1.2.1.1 /01 had been invalidated and thus not included into the current evaluations.

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Report:	h; ;2013;M-464601-01
Title:	Isoxaflutole: Kinetic modelling evaluation of aerobic soil degradation study data to derive
	trigger endpoints
Report No:	VC/13/007C
Document No:	M-464601-01-1
Guidelines:	EU Council Directive 91/414/EEC, as amended by Commission Directive 95/36/ÉC of
	July 1995, Section 5, Point 7 and Commission Regulation (EC) No 1107/2009 of 21 🖉
	October 2009; not applicable
GLP/GEP:	no a v v v

Executive Summary

For the active substance isoxaflutole degradation data as referenced under KCA 7.1.2,1.1 402 to KCA 7.1.2.1.1 /05 were kinetically evaluated according to FOCUS Guidance to derive values for the half-life and the DT₉₀ in aerobic soil for comparison with frigger endpoints². For metabolites RPA 202248 and RPA 203328 the kinetic analysis was performed in combination with parent compound data, amended by derobic soil degradation data from separate tests with RPA 202248. The results are summanized under Point 7.1 2.1.2

For the evaluation degradation data of the active substance isosaflutole in Soil under aerobic conditions of the laboratory were considered from three studies following application of UL-¹⁴C-phenyl-labelled active substance to five different soils.

Non-normalised values of the DT_{50} and the DT_{90} were derived from SFO best fits in three soils and DFOP best fits in two soils with results summarized in Table 7. D2.1.1 4.

From tests at 20°C non-normalised half hves of isoxatiutole ranged from 0.3 days for clay loam soil 95/05 to 3.9 days for loam soil 95/02 while values for the DT_{90} ranged from 1.0 day for soil 95/05 to 18.6 days for soil 95/02.

At 10°C, the DT_{50} was 6.1 days and the DT_{90} was 35.6 days in soil 95/02.

Table 7.1.2.1.2-1: Summary of results of kinetic evaluation of degradation for isoxaflutole in aerobic soil in the aboratory for companison against for triggers

	.0'
Compound & 5 6 5	Isoxaflutole
200, Normormalised DIS, range (days)	0.3 – 3.9
Worst case DT (days)	3.9
20°C, Non-normalised DT ₉₀ , ange (days)	1.0 – 18.6
Worst case DT90 (days) 🔬 🖉 🔊	18.6
10° C, Non-normalised Γ_{50} (days)	6.1
10°C Non-normalised DT ₂₀ (days)	35.6
ĉ	

² Please note that KCA 7.1.2.1.1 /01 has been invalidated and thus excluded. For details, see CA 7.1.1.1 /01.

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

I. Material and Methods

For the parent compound isoxaflutole details on study conduct and its results have been summarized with the software KinGUI, version 2.

The measured values were taken into account as reported and thus treated as individual replicates. All sets with their data points were weighted equally. The concentration at this zero was included in the parameter optimisation with the initial value being allowed to be estimated by the model. In cases where the radioactive residues in soil were below the limit of detection (LOD) the respective values were set to 0.5 LOD for the evaluation for time points before or after which a value above LOP was determined. For some studies no LOD was given in the original deport. In these cases no values were added.

In some cases degradation products of the applied substance were already detected at time zero. In such cases the respective percentages were added to the parent values and the values for the metabolite were set to zero.

All radioactive residues in soil were used for the kinetic evaluation. For some of the stadies performed for very long periods of up to one year the evaluations for deriving modelling indposits used only data measured up to day 120 days which is the maximum recommended duration for laboratory studies according to OECD Guideline 307 (2002)

For fits of compounds under evaluation, SFQ kinetles was tested first due to its simplicity and its nearly exclusive use intervironmental exposure models.

In general, also the use of other sinetic model approaches is possible as proposed by FOCUS. The evaluation thus considered also the model approaches first order multiple compartment (FOMC), dual first order in parallel (DFOP) and Hockey Stick (HS), in principle, following the scheme for identification of the propriate kinetic model as proposed by FOCUS!

To check the parameters for their significance a single sided test was used. The probability of t should be low or equal to zero as this probability can be assumed to be higher the more uncertain a parameter is. In general, a value of 0.05 for the probability of t is considered as appropriate with degradation parameters being regarded as monificant at this level.

H. Results and Discussion

Following application of the parent substance isoxaflutole to aerobic soil the nearly exclusive and successive formation of metabolites RPA 202248 and RPA 203328 was observed thus supporting a 'linear' pathway under aerobic conditions. Formation fractions for metabolites were thus set to one regarded as being suitable for this assessment.

Calculation of pon-normalised DT₅₀-values based on best fits:

For isoxaflutole the kinetic evaluation of soil degradation tests using the SFO approach did not result in acceptable fits to the experimental data. For two data sets the evaluation according to FOMC resulted in a better fit than SFO to describe the degradation data. Finally the DFOP model was chosen as optimal fit for the two data sets that had shown better fits already by use of the FOMC model. The



resulting non-normalised values for the DT_{50} and the DT_{90} derived are summarized in Table 7.1.2.1.1-2.

Table 7.1.2.1.1-2: Non-normalised DT50-values for isoxaflutole in aerobic soils under laboratory conditions for trigger evaluation

-			4/17	· · · · · · ·
Soil (Origin)	Label position	DT ₅₀	_% DT ₉₀	Model
		(days)	(days)	
95/02, loam, 20°C (Study 1)	phenyl	<u>گ</u> 3.9	18.6	DEOP S
95/04, sandy loam, 20°C (Study 2)	phenyl	§ 2.4 Q	8.1	SFO SFO
95/05, clay loam, 20°C (Study 2)	phenyl 🖌	0.3	1.0 🖉	SFQ SFQ
95/06, loamy sand, 20°C (Study 2)	phenyl 🔬	0.6.Q	° 1.8	SÊŎ ÇŰ
95/02, loam, 10°C (Study 3)	phenyl	ál í	, 3₽,6 , C	PFOP OF
	~~~			K × ~ ~ ~

Study 1: KCA 7.1.2.1.1 /02 and KCA 7.1.2.1.1 /03 Study KCA 7.1.2.1.1 /04; Study 3:0xCA 7.1.2.1.1 /05

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The evaluation according to FOCUS functic guidance resulted in values for half-lives and the  $DT_{90}$  of the parent compound isoxaflutole for comparison with EO trigger endpoints. The approach for fitting with experimental data resulted in the use of the single-phase SFO and the biphasic kinetic model DFOP for calculation. Values of non-normalised half-lives and  $DT_{90}$  from best fits to measured data ranged from 0.3 to

3.9 days. The corresponding DT₉₀ ranged from 7.0 day to 1866 days.

At 10°C, the corresponding values for the  $D\Phi_{50}$  and  $DT_{90}$  were 6 4 days and 35 O days.

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Report:	J.	by	<b>\$201</b> .	3;10-464596-01		
Title:	/ Asoxaflut	de: Kinetic mo	delling evaluat	ion of acrobic s	of degradation st	udy data to derive
	modelOn	g endpoints &				
Report No:	© VC/13/0	07 <b>B</b>				
Document No:	MK46459	<b>6-</b> 01-1 ~	0, 0, 0			
Guidelines:	EU Coð	acil Directive 9	1/414/£EC, as	amended by C	Commission Dire	ctive 95/36/EC of
**	July 199	5, Section 5, P	oint 🄊 and Con	nmission Regul	lation (EC) No 11	107/2009 of 21
	🔊 October	2009;not appl	icable 🔿	~ ~		
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#### Executive Summary

For isoxatiutole degradation data as referenced under KCA 7.1.2.1.1 /02 to KCA 7.1.2.1.1 /05 were kinetically evaluated according to FOCUS Guidance to derive modelling endpoints for use as input parameters in environmental risk assessments³.

For metabolites RPA 202248 and RPA 203928 the kinetic analysis was performed in combination with parent compound date amended by aerobic soil degradation data from separate tests with RPA 202248 and summarized under Point CA 7.1.2.1.2.

For the evaluation degradation data of isoxaflutole in soil under aerobic conditions of the laboratory were considered from three studies following application of UL-¹⁴C-phenyl-labelled active substance to five different soils.

³ Please note that KCA 7.1.2.1.1 /01 has been invalidated and thus excluded. For details, see KCA 7.1.1.1 /01.



A stepwise approach was made for the calculation of normalised half-lives in soil. The initial step consisted of best fits to the measured data following the SFO kinetic model. In case of unacceptable fits according to the criteria set bi-phasic models, i.e. FOMC or DFOP were applied. The SFO model adequately described the measured data for all data sets. In a next step, values were normalised by comparison of study incubation conditions to reference conditions (20°C, pF2 moisture) with results summarized in Table 7.1.2.1.1 -3.

For use as modelling endpoint, the overall mean normalised half-life of isoxaflutole was 0 days. The formation fractions were one for RPA 202248 and 0.96 for RPA 203328.

Table 7.1.2.1.2-3: Normalised laboratory DT50-values for isoxafutole in aerobic soil in the laboratory for use as modelling input parameters in environmental exposure assessments

use as mouening	5 mput parameters		intar caposula	assessingines	a. 2	
	/			ĵ –		L,
Compound			🔍 🌂 Isowaf	lutole		Ŋ
Normalised (20°C, p	oF2) DT ₅₀ , range (d	ays)	JO 49.3 -	- 3.9 🔊		
Mean (geometric)			y ₁ 0,	\$ <u>`</u>	ý O	
		». ~~	ST O	Ŭ Â	ji v	
				~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		
					% <i>,</i>	

Material and Methods

For the parent substance isoxaflutible details or study conduct and its results have been summarized under Point CA 7.1.1.1. The degradation data were sinetically evaluated following FOCUS guidance with the software Kingdul, variance

The measured values were taken byto account as reported and thus treated as individual replicates. All sets with their data points were weighted equally. The concentration at tone zero was included in the parameter options ation with the initial value being allowed to be estimated by the model.

In cases where the fadioactive residues in soil were below the limit of detection (LOD) the respective values were set to 0.5 LOD for the evaluation for time points before or after which a value above LOD was determined. For some studies no LOD was given in the original report. In these cases no values were added.

In some cases degradation products of the applied substance were already detected at time zero. In such cases the respective percentages were added to the parent values and the values for the metabolite were set to zero.

All radioactive residues in soil were used for the kinetic evaluation. For some of the studies performed for very long periods of up to one year the evaluations for deriving modelling endpoints used only data measured up to day 120 days which in the maximum recommended duration for laboratory studies according to OECD Guideline 307 (2002).

For fits of compounds we der evaluation, SFO kinetics was tested first due to its simplicity and its nearly exclusive use in environmental exposure models.

In general, also the use of other kinetic model approaches is possible as proposed by FOCUS. The evaluation thus considered also the model approaches first order multiple-compartment (FOMC), dual first order in parallel (DFOP) and Hockey Stick (HS), in principle, following the scheme for identification of the appropriate kinetic model as proposed by FOCUS.

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

To check the parameters for their significance a single-sided t-test was used. The probability of t should be low or equal to zero as this probability can be assumed to be higher the more uncertain a parameter is. In general, a value of 0.05 for the probability of t is considered as appropriate with degradation parameters being regarded as significant at this level.

The DT₅₀-values derived were normalised to standard reference temperature 20 °C and soil monsture? 100 % field capacity in order to obtain standardised input parameters for predictions of environmental concentrations. This normalisation was conducted according to the standard approach by EQCUS

The degradation of isoxaflutole in aerobic soil resulted in the predominant and successive formation of metabolites RPA 202248 and RPA 203328. From observation on the course of the tests, it is reasonable to assume a 'linear' pathway for their formation, i.e. the components were formed successively.

IL Results and Discussion

Following application of the parent substance is a flutore to the robit soil the nearly exclusive and successive formation of metabolities RPA 202248 and RPA 203328 was observed thus supporting a 'linear' pathway under aerobic conditions. Formation fractions for metabolities were thus set to one regarded as being suitable for this assessment. The resulting formation fractions are summarized in Table 7.1.2.1.1-4.

Calculation of non-normalised DT 50-values based on SFQ Dest fiktor isoxaflutor

For the parent substance the kinetic evaluation of soil degradation tests using the SFO approach resulted in acceptable fills to the experimental data for all data sets. The results are summarized in Table 7.1.2.1.49.

Normalised DT50 values for isoxaflutole

For the ose in environmental modering the destadation half lives were normalised to reference conditions of 100 % field capacity regarding soil moisture and 20°C for the temperature. The parameters used in the laboratory tests and the espective correction factors calculated are summarized in Table 7.1.2.1.1-6. The half lives regulting from normalisation are summarized in Table 7.1.2.1.1-7.

Table	7.1,2,1.1	-4: Formation	fractions	of meta	bolites & RPA	202248	and RPA	203328	from	application	of
	J.	parent substa	nceisoxafl	utole to	aerobie soil	under la	boratory c	ondition	s		

Soil (Origin) 🗸 💭	Lapel position	Formation fraction	on for process
		Isoxaflutole	RPA 202248
	Q Q	to	to
	j [°] a, [°]	RPA 202248	RPA 203328
95/02, loan 20°C Study A	~Q~ 1	1.00	1.00
95/04, sandy loan, 20°CQStudy D	1	1.00	0.868
95/05, Qay loadr, 20°G (Study 2)	1	1.00	1.00
95/06, loamy sand, 20°C (Study 2)	1	1.00	0.966
95,62, loan, 10°C (Study 3)	1	1.00	1.00
Mean (mithmetic)		1.00 *	0.96

Label position: 1 = phenyl

Study 1: KCA 7.1.2.1.1 /02 and KCA 7.1.2.1.1 /03; Study 2: KCA 7.1.2.1.1 /04; Study 3: KCA 7.1.2.1.1 /05

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Table 7.1.2.1.1-5: Non-normalised DT₅₀-values for parent substance isoxaflutole in aerobic soils under laboratory conditions Ø

-			
Soil (Origin)	Label position	DT50	Model
		(days)	Ŭ <u> </u>
95/02, loam, 20°C (Study 1)	1	4.2	SFØ " "
95/04, sandy loam, 20°C (Study 2)	1	2.4	SFO N
95/05, clay loam, 20°C (Study 2)	1	0.3	SFQ Q 4
95/06, loamy sand, 20°C (Study 2)	1	<u>گ</u> 0.6	SFQ SFQ
95/02, loam, 10°C (Study 3)	1	6.7 ₂	Stor stor
Label position: $1 = phenyl$	Â	(, ⁰	

Label position: 1 = phenyl

Study 1: KCA 7.1.2.1.1 /02 and KCA 7.1.2.1.1 /03; Studa 2: KCA 7.1.2.4

Table 7.1.2.1.1-6: Correction factors for soil temperature and moisture conten

		0	4 j C	r 🔊 '0	d A.
	Soil (Origin)	Incubation	Incubation Q	pF 2	Correction ?
		temperature	moisture	moist@e **	factor
		(°C), `~	@(% w/\$)	(% W/W)	
	95/02, loam, 20°C (Study 1)	~~~~ 20€ [~] ~~	~~~~~ 2 6C96 ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Q25.52	Q 1,000
	95/04, sandy loam, 20°C (Study 2)	< <u>√</u> 20°´```	27.13	8 23. 19 5	<u>س (</u> 1.000
	95/05, clay loam, 20°C (Study 2)	j 620 (b)	35.01	O 6248 Û	*≫0.667
	95/06, loamy sand, 20°C (Study 2)	<u></u> 20 0°	17 40	§× 28 .28 ∞	% 0.714
	95/02, loam, 10°C (Study 3)		26.96	ي <u>کې 25.52</u>	O° 0.387
1					

* Values given in study report

** Calculated values according to FOCUS, 2000 Study 1: KCA 7.1.2.1.1 /02 and KCA ЌСА 7.1.2.1.1 /05 1 /04: Study &

slues in aerobic soil for parent compound isoxaflutole for Table 7.1.2.1.1-7: Normalised use as modelling input parameters in environmental exposure assessments

Soil (Origin) 🔍		DT ₅₀	, 🖗 DT _{so} y	©hi ² - error	t-test	VA
, Q	, ·	(days)	(days)	, (Š)		
95/02, loam, 20°C	(Study) 🕺	× .2 .	A3.9 S	9.3	<2e-16	Good
95/04, sandy loam,	, 20°C (Study 2)	2.4	8.1	3.8	<2e-16	Very good
95/05, clay loam, 2	20% O (Study 2)	0°20	≪″0 <i>≸</i> %	3.4	<2e-16	Very good
95/06, loamy sand,	20°C (Study 2)	¢ øa _s	1.3	• 4.9	<2e-16	Very good
95/02, loam, 10°C	(Study 🕄 🔊	2.6 ×	\$8.6	10.4	<2e-16	Good
Mean (geometric)	0	^ 0.9				

2 /03 Study & KCA 7.1.2.1.2 /04; Study 3: KCA 7.1.2.1.2 /05 Study 1: KCA 7.1.2.1.2

IIP Conclusion

The evaluation according to FOCUS kinetic guidance resulted in values for half-lives of the parent compound soxaftutole is aerobic soil for use as modeling input parameters in environmental risk assessments.

As a result of fitting to the measured data the use of SFO was regarded as the appropriate kinetic model to derive non-normalised values for the DT₅₀ in a first step. The values were then normalised for moistore (pF 2) and temperature (20°C).

The values derived from laboratory tests in five EU soils are regarded as suitable and reliable for use in environmental exposure assessments.



For the active substance isoxaflutole, a normalised half-life of 0.9 days was calculated as modelling

For the formation fraction of active substance to metabolite RPA 202248 a mean value of 100 was calculated while for the formation of RPA 203328 from RPA 202248 a mean value of 0.96 was derived. in the second se

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Title:	Herbicides: Isoxaflutole Route of Degradation (Aerobic Metabolism) in one Soil
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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Report:	t; ;2012;M-427583-01
Title:	[Cyclopropyl-1-14C]RPA 202248:: Aerobic degradation/metabolism in one European soil
Report No:	MEF-11/486
Document No:	M-427583-01-1
Guidelines:	OECD 307; US EPA OPPTS 835.4100 and OPPTS 835.4200; EU 95/36/EC amending
	91/414/EEC;not specified
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Report:	8; ;2017;M-464601-00 ²
Title:	Isoxaflutole: Kinetic modelling evaluation of aerobic soil degradation study data to derive
	trigger endpoints
Report No:	VC/13/007C
Document No:	M-464601-01-1
Guidelines:	EU Council Directive 91/414/EEG, as an ended by Commission Directive 95/36/EC of
	July 1995, Section 5, Point 7 and Commission Regulation (EC) No 1107/2009 of 21
	October 2009;not applicable
GLP/GEP:	

Report:	9; ;20(\$7;M-464596-@)
Title:	Isoxaflutole: Kinetic modelling evaluation of aerobic soil degradation study data to derive
	modelling endpoints
Report No:	VC/13/007B & 0 A A A A
Document No:	M-464596-01-1 ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~
Guidelines:	EU Council Directive 91/414/EEC, as amended by Commission Directive 95/36/EC of
	July 2995, Section 7, Point 7 and Commission Regulation (EC) No 1107/2009 of 21
	October 2009; not applicable 2 5 2 2 2
GLP/GEP:	no star star star star star star star star

For isoxaflutole metabolites laboratory data on the rate of degradation in zerobic soil had been derived from studies performed under the following conditions: ų. K

- two soils at 21°C and a mosture of 75% of the field capacity at 0.33 bar following application of phonyl-UL-14C-labelled active substance (KCO 7.1.24) 2 /019
- one soil under standard conditions at 20°C and mossture at 45 % maximum water holding capacity (MWHC) after application of phonyl-ULO4C-labelled active substance (KCA 7.1.2.1.2 /02 and KCA 7.1.2.1.2 /03
- three soils under standard conditions at 20 °C and moisture at 40% MWHC following application of phenyl-UL-14C-labered active substance (KCA).1.2.1.2 /04);
- one soil at 10°C and 40 MWOC and application of phenyl-UL-14C-labelled active substance (KCA 7.1.2.1.2 /05)

This data requirement had beer addressed upper Point 7.1.1.2 of the Dossier submitted and evaluated within the process of evaluation for Annex inclusion as published in the corresponding Monograph of RMS The Netherlands (ebruary 1992) and its amendments.

Following latest guidance on kinetic evaluation the data from existing studies have been re-evaluated. The new Menetic evaluations considered new rate data from a separate test performed with cyclopropy-1-1-14C-labelled RPA 202248 (KCA 7.1.2.1.2 /06) in addition. The test was performed to investigate the degradation behavior of the cyclopropyl-ring system and to support the determination of a robust data set for this metabolite.



The kinetic evaluations were performed separately in order to derive values for the comparison with trigger endpoints (KCA 7.1.2.1.2 /07) and modeling endpoints (KCA 7.1.2.1.2 /08) for input into environmental risk assessments.

The re-evaluation following new guidance in the kinetic analysis of degradation data in activities so supersede the existing evaluations.

As explained under KCA 7.1.1.1 in more detail, study KCA 7.1.2.1.1 (b) had been invalidated and thus not included into the current evaluations.

Report:	9; 2012;M-4275\$3-01
Title:	[Cyclopropyl-1-14C]RPA 202278:: Aerobic degradation/metabolism in one European soil
Report No:	MEF-11/486
Document No:	M-427583-01-1 & 6° 5 5 5 6 5 5
Guidelines:	OECD 307; US EPA OPTS 833.4100 and OPTS 833.4200 EU 95/36/EC amending
	91/414/EEC;not specified
GLP/GEP:	yes which a do a do a do

The study has been summarised in prore defail under KCA 7.1 1/1 /05 of this document.

The kinetic evaluation of the degradation of [Cyclophopyl-b] C]-BPA 202248 in aerobic soil resulted in a non-normalised DC₅₀ of 14 days and a DT_{90} of 47 days based on an SFO best fit kinetic model. The values may be used for comparison with trigger endpoints and as modelling input in environmental risk assessments.

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Report:	<u></u>		L.	2013;M-4	64601-01		
Title:	J Isosat	lutole: Kinet	ie modelling e	aluation of a	erobic soil	degradation stu	udy data to derive
	d strigge	r endpoints	Nº 4	/ ~~ A			
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Document No:	© M-462	1 601- 0 -1	4 24		×		
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Ê.S	July 1	1995, Section	15, Point 7 and	d Commissio	Regulati	on (EC) No 11	107/2009 of 21
**	Octol	der 2009;not	applicable/	6 47	-		
GLP/GEP:	No 🔨		× ~				
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Executive Summary

For the active substance isoxathrole degradation data were kinetically evaluated under KCA 7.1.2.1.1 /06 according to FOCNS Guidance to derive endpoints for comparison with trigger endpoints. For metabolites RFA 202248 and RPA 203328 the kinetic analysis was performed in combination

with parent compound data as referenced under KCA 7.1.2.1.2 /02 to KCA 7.1.2.1.2 /054.

The degradation data for RPA 202248 and RPA 203328 was derived from tests with active substance isoxaflutole in aeropic soil under conditions of the laboratory in four studies after application of UL-¹⁴C-phenol labelled active substance to four different soils⁵. Additional information on the rate of degradation of RPA 202248 was available for one soil (see KCA 7.1.2.1.2 /06).

⁴ Please note that KCA 7.1.2.1.1 /01 has been invalidated and thus excluded. For details, see KCA 7.1.1.1 /01.

⁵ Please note that KCA 7.1.2.1.1 /01 has been invalidated and thus excluded. For details, see KCA 7.1.1.1 /01.

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Non-normalised values of the DT_{50} and the DT_{90} were derived from SFO best fits for three data sets and DFOP best fits for two data sets with results summarized in Table 7.1.2.1.2 -1. Non-normalised half-lives for metabolite RPA 202248 ranged from 10.4 days for sandy login soil 95/04 to 43.4 days for loam soil 95/02 while values for the DT_{90} ranged from 3.4 days for sandy login

soil 95/04 to 144 days for loam soil 95/02.

B

At 10°C, the corresponding values for the DT_{50} and DT_{90} were 44.8 days and 149 days in the loom soil? 95/02.

Non-normalised half-lives from tests performed at 20°C for metabolite RPA 203328 ranged from 3.5 days to more than 1000 days. Values for the DT₂₀ ranged from 51.6 days to more than 1000 days. 0° For tests performed at 10°C, the corresponding ralues for the DT₅₀ and D 0° were 112 days and 379 days.

Table 7.1.2.1.2-1: Non-normalised DT₅₀-values for metabolites RPA 202248 and RP 203328 in aerobic soil under laboratory conditions for trigger evaluation

	<u></u>	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	i Y 👒		A ON	
Compound		∑ Ba	A 2022	1 8 A	°°∕≫ RF	A 203328
20°C, Non-normalised DT ₅₀ , rang	Q(days)√		0.4 43.	4 🔊	õ 3 ₆	S->1000 ℃
Worst case DT50 (days			4 3.4	\sim		>1000
20°C, Non-normalised DT ₉₀ , and	e (days)		4.4 - 14	ď, ď		.6 * 1000 * .
Worst case DTm/(days	K U		144		ð,	0000 (j
10°C, Non-normalised Data (days)		100	4 4 8	Ĩ,	ČA (J112 0
10°C, Non-normalised DT ₉₀ (days) ()	L'	@ 1 49		Y Q	379
	2		~			K. 1

A. Material and Methods

Data sets were derived from studies with the parent compound and a separate study with RPA 202248. The details on study conduct and their results have been summarized under Point CA 7.1.1.1. The degradation data were kinemally valuated following COCUS guidance with the software KinGUI, version 2.

The measured values were taken into account as reported and thus treated as individual replicates. All sets with their data points were weighted equally. The concentration at time zero was included in the parameter optimisation with the initial value being allowed to be estimated by the model.

In cases where the radioactive residues in soil were below the limit of detection (LOD) the respective values were so to 0.5 LOD for the evaluation for time points before or after which a value above LOD was determined. For some studies no LOD was given in the original report. For these cases no values were added.

In some cases degradation products of the applied substance were already detected at time zero. In such cases the respective percentages were added to the parent values and the values for the metabolite were set to zero@

All radioaction residues in soil were used for the kinetic evaluation. For some of the studies performed for very long periods of up to one year the evaluations for deriving modelling endpoints used only data measured up to day 120 days which is the maximum recommended duration for laboratory studies according to OECD Guideline 307 (2002).

For fits of compounds under evaluation, SFO kinetics was tested first due to its simplicity and its nearly exclusive use in environmental exposure models.



In general, also the use of other kinetic model approaches is possible as proposed by FOCUS. The evaluation thus considered also the model approaches first order multiple-compartment (FOMC), double first order in parallel (DFOP) and Hockey Stick (HS), in principle, following the scheme for identification of the appropriate kinetic model as proposed by FOCUS. To check the parameters for their significance a single-sided t-test was used. The probability of t should be low or equal to zero as this probability can be assumed to be frigher the more uncertain a? parameter is. In general, a value of 0.05 for the probability of t is considered as appropriate with degradation parameters being regarded as significant at this level.

II. Results and Discussion successive formation of metabolites RPA 202248 and RPA 203928 was observed that supporting a 'linear' pathway under aerobic conditions Formation fractions for metabolites were thus set to one.

Calculation of non-normalised DT 50-values based on best fits

For the parent compound isoxaflutole the kinetic evaluation of soil degradation tests using the SFO approach did not result in acceptable with to the experimental data. For two data sets, the evaluation according to FOMC resulted in a better than SFO to describe the degradation Data. Finally the DFOP model was chosen as optimal fit for the two data sets that had shown better fits already by use of the FOMC model. From combination of best fits optained for the active substance with SFO fits for the metabolites the resulting non-pormalised values for the De 50 and the DT 90 are summarized in Table 7.1.2.1.1-3 for metabolite RPA 202248 and Table 7.1.2.1.1-4 for metabolite RPA 203328.

Table 7.1.2.1.2 ... Non normalised DE-values for metabolite RP 202248 (AE 0540092) in aerobic soils under aboratory conditions for trigger evaluation

	1		())/	(// n	1///2		
Soil (Origun)	Ŭ S	DT 50	DIG	^\$#fm* ^	Chi ² -error	t-test	VA
K~y ^v		(days)	(days)		(%)		
95/02, loam, 20°C	C (SQudy 1) 🖉 🖉	4 <u>3</u> .9	≪144 🖉	1.00	11.3	<2e-16	Good
95/04, sandy loam	£20°C (Study 2₽	40 .4 🔬	⁰ 34.4	00:00	14.7	<2e-16	Good
95/05, clay loam, 1	20°C Study 2	41.7 ×	189	S 1.00	5.6	<2e-16	Very good
95/06, loamy sand	l, 2000 (Study 2) ``	≫ 29,4	°~97.6	1.00	13.5	8.2e-15	Good
95/02, loam, 10°C	C (Study 3	4 4 8	2 149°°	1.00	14.7	<2e-16	Acceptable

Study 1: K&A 7.1.2.1.2 /02 and KOA 7.1 21.2 /03 Study 2: KCA 7.1.2.1.2 /04; Study 3: KCA 7.1.2.1.2 /05

VA = Visual Assessment, ffm = formation fraction: where ffm was optimised to 1 during evaluations, the data



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Table 7.1.2.1.2-3: Non-normalised DT ₅₀ -values for metabolite RPA 203328 (AE B197555) in	aerobic soils
under laboratory conditions for trigger evaluation	Ŵ

Soil (Origin)	DT_{50}	DT90	ffm	Chi ² -error	t-test	XA O
	(days)	(days)		(%)	ð	
95/02, loam, 20°C (Study 1)	204	678	1.00*	3.2	₹ 5.49e-11	Excellent
95/04, sandy loam, 20°C (Study 2)	>1000	>1000	0.868	4.8	0.180 🔍	Good
95/05, clay loam, 20°C (Study 2)	3.5	11.6	1.00*	16.9	<2e-16	Acceptable
95/06, loamy sand, 20°C (Study 2)	>1000	>1000	Ö0.966	19/4	0.5	Acceptable
95/02, loam, 10°C (Study 3)	112	379 "	¥ 1.00*	.06.2	<2,2,2,16	Séxcellent 🗸

Study 1: KCA 7.1.2.1.2 /02 and KCA 7.1.2.1.2 /03; Study 2 KCA 7.1.2.1.2 /04; Study 3: KCA 7.52.1.2 6° VA = Visual Assessment; ffm = formation fraction: where ffm was optimised to 1 during evaluations, the data was refitted with ffm = 1

III. Comelusion

The evaluation according to FOCUS kinetic guidance resulted in values for half-lives and the DT₀ for metabolites RPA 202248 and RPA 200328 for comparison with EU trigger endpoints. The fitting with experimental data resulted in a combined use of the Bi-phasic kinetic model DFOP and the SFO model for the parent compound is exafluted with SFO kinetics for the metabolites.

For metabolite RPA 202248 non-normalised half-lives from best fits to measured data ranged from 10.4 days to 43.4 days. The corresponding values for the DT_{50} ranged from 34.4 days to 144 days. At 10°C, the corresponding values for the DT_{50} and DT_{60} were 44.8 days and 149 days.

For metabolite RPA 203328 calcolable values of non-normalized half-lives and DT_{90} from best fits to measured data ranged from 3.5 days to 204 days while values for the DT_{50} ranged from 11.6 days to 678 days. For tests performed at 10°C, the corresponding values for the DT_{50} and DT_{90} were 112 days and 379 days.

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Report:	;2013;14,464596-01
Title:	Isoxaflutole Kinetic modefling evaluation of aerobic soil degradation study data to derive
Ô	modelling endpoints
Report No:	AC/13/0077B
Document No:	0 M-464 396-01 1 ~ ~ ~ ~
Guidelines:	EU Council Directive 91/414/EEC, as amended by Commission Directive 95/36/EC of
- And	July 1995, Section 5, Point 7 and Commission Regulation (EC) No 1107/2009 of 21
Į.	October 2002; not applicable
GLP/GEP:	no A O A
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Executive Summary

For metabolites RRA 202348 and RPA 03328 degradation data were kinetically evaluated according to FOCUS Guidance to derive modelling endpoints as input parameters for environmental risk assessments.

The kinetic analysis was performed in combination with parent compound data as referenced under KCA 7.12.1.2 /02 to KCA 7.1.2.1.2 /05⁶. The evaluation considered degradation data of the active substance isoxaflutole in soil under aerobic conditions of the laboratory from four studies following

⁶ Please note that KCA 7.1.2.1.1 /01 has been invalidated and thus excluded. For details, see KCA 7.1.1.1 /01.



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application of UL-¹⁴C-phenyl-labelled active substance to four different soils. For RPA 202248 these were amended by aerobic soil degradation data from a separate test (KCA 7.1.2.1.2 /06). A stepwise approach was made for the calculation of normalised half-lives in soil. The initial step consisted of best fits to the measured data following the SFO kinetic model. A case of unacceptable fits according to the criteria set bi-phasic models, i.e. FOMC or DFOP were applied. The evaluation resulted in the SFO model to adequately describe the fit to measured data for all data sets. In a next step, values were normalised by comparison of study increasion conditions to reference conditions (20°C, pF2 moisture) with results summarized in Table 7 12.1.1 -4.

For use as modelling endpoints, overall mean normalised half-lives were 18.9 days for RPA 202248 when including data from the separate study with RPA 202248 (see KCA 7, 1.2.12, 06), and 121 days for RPA 203328.

Mean values for formation fractions were one for metabolite RPA 202248 and 0.96 for RPA 203325/

Table 7.1.2.1.2-4: Normalised laborators DT56 values for RPA 202248 and RPA 203328 in acrobic soil in the laboratory for use as modelling input parameters in environmental exposure assessments

Compound			RPA 202248	34 >	RPA 203328	
Normalised (20°C, pF2	2) \mathbb{Q}_{50} , range	(days)	[™] 10.4 – 44.8		Q.3 - 1900	
Mean (geometric)	Ŵ K,		(, 💦 18.9* ⁽)	20	<u>\$ 121</u>	
* Including a DT ₅₀ -v	plue of $\overline{\mathbb{O}4.2}$ da	vs from K	CA 7.1.2.1.2 /06	2 4		

I. Material and Methods

Studies with the parent substance isoxatlutole are summarized under Point CA 7.1.1.1. The degradation data were kinetically evaluated following FOCUS guidance with the software KinGUI, version 2. The measured values were taken into account as reported and thus treated as individual replicates. All sets with their data points were weighted equally. The concentration at time zero was included in the parameter optimisation with the initial value was estimated by the model.

In cases where the adioactive residues is soil were below the limit of detection (LOD) the respective values were set to 0.5 LOD for the evaluation for time points before or after which a value above LOD was determined. For the studies no LOD was given in the original report. In these cases no values were added

In some cases degradation products of the applied substance were already detected at time zero. In such cases the respective percentages were added to the parent values and the values for the metabolite were set to zero.

All radioactive residues in soit were used for the kinetic evaluation. For some of the studies performed for very long periods of up to one year only data up to 120 days were used which is the maximum recommended duration for laboratory studies according to OECD Guideline 307 (2002).

SFO kinetics was tested that due to its simplicity and its nearly exclusive use in environmental exposure models.

In general, also the use of other kinetic model approaches is possible as proposed by FOCUS. The evaluation thus considered also the model approaches first order multiple-compartment (FOMC),

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

double first order in parallel (DFOP) and Hockey Stick (HS), in principle, following the scheme for identification of the appropriate kinetic model as proposed by FOCUS.

To check the parameters for their significance a single-sided t-test was used. The probability obt should be low or equal to zero as this probability can be assumed to be higher the more uncertain a parameter is. In general, a value of 0.05 for the probability of t is considered as appropriate with degradation parameters being regarded as significant at the level.

According to FOCUS, the DT_{50} -values were normalized to standard reference temperature 20 °C and soil moisture 100 % field capacity in order to obtain standardised input parameters for predictions of environmental concentrations.

The degradation of isoxaflutole in aerobic soib esulted in the predominant and successive formation of metabolites RPA 202248 and RPA 2033284 From observation in the course of the tests, it is teasonable to assume a 'linear' pathway for their formation, i.e. the components were formed successively.

VII. Results and Discussion

Following application of the parent substance isos allutole to aerobic soil the nearly exclusive and successive formation of metabolites RPA 202248 and RPA 203328 was observed thus supporting a 'linear' pathway under aerobic conditions. Formation fractions for metabolites were thus set to one regarded as being suitable for this assessment.

Calculation of non-normalised DT values based on SFO best fit:

For metabolites RPA 202248 and RPA 203328 the kinetic evaluation of soil degradation tests using the SFO approach resulted in acceptable first to the experimental data for all data sets. The results are summarised in Table 7.1.2.4.2-5 and Table 7.1.2.4.2-6

Normalised DT50-values for RPA 202248 and RPA 203328

For the use in environmental modeling the degradation half-lives were normalised to reference conditions of 100% field capacity degrading soil moisture and 20°C for the temperature as summarized under Table Table 7.1.2.4.2-7. The values of half-lives resulting from normalisation are summarized in Pable 7.1.2.1.2-8 and Table 7.1.2.1.2-9.

Table 7.42.1.2-5: Non normatised DT50-values for metabolite RPA 202248 in aerobic soil under laboratory conditions

Soil (Origin)	Label position	DT_{50}	Model
		(days)	
95/02, loam, 20°C (Study 1)	N N	44.9	SFO
95/04, sand Toam, 20°C (Study 2)		10.4	SFO
95/05, clas loam Q0°C (Study 2)	₩ 1	41.7	SFO
95/06, loamy sand, 20°C (Study 2)	1	29.4	SFO
95/02, toam, OPC (Study 3)	1	44.6	SFO

LabeDposition: 1 = phenyl

Study 1: KCA 7.1.2.1.2 /02 and KCA 7.1.2.1.2 /03; Study 2: KCA 7.1.2.1.2 /04; Study 3: KCA 7.1.2.1.2 /05

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Table 7.1.2.1.2-6: Non-normalised DT50-values for metabolite RPA 203328 in aerobic soil under laboratory conditions Ø

Soil (Origin)	Label position	DT ₅₀	Model
		(days)	ϕ ϕ ϕ
95/02, loam, 20°C (Study 1)	1	210	SFO SFO
95/04, sandy loam, 20°C (Study 2)	1	>1000 🔬	SFO ~
95/05, clay loam, 20°C (Study 2)	1	3.5	SFQ Q K
95/06, loamy sand, 20°C (Study 2)	1	>1000	SFQ SFQ
95/02, loam, 10°C (Study 3)	1	¥ 104,Q	0 Stor ~ ~
Label position: $1 = phenyl$	li l	(, ⁰	

Label position: 1 = phenyl

Study 1: KCA 7.1.2.1.2 /02 and KCA 7.1.2.1.2 /03; Study 2: KCA 7.1.2.2 /04; Study 2: KCA 7.1.2 /04; Study 2: KCA 7.1.2.2 /04; Study 2: KCA 7.1.2 /04

Table 7.1.2.1.2-7: Correction factors for soil temperature and mosture content

	0		<u>, 0</u>	
Soil (Origin)	Incubation	Finctopation Q	pF 2	Correction factory
	temperature	moisture*	moistur®**	
	(°C)°∼y		Ó [♥] (%,₩₩)	
95/02, loam, 20°C (Study 1)	Q 20 [×]	26.96	25 .52	0 1.000
95/04, sandy loam, 20°C (Study 2)	209" >	y 29.13 s	~23.19°	L 1,000
95/05, clay loam, 20°C (Study 2)	رض 20 ک	°∂-35.012°	O 62.45	Č °€⁄ð67
95/06, loamy sand, 20°C (Study 2)	ູ 📣 ້ 20 🧭	\$ 17. 40	28028	0.714
95/02, loam, 10°C (Study 3)	× 10-	26.96 O	_ 3 5.52	◎ 0.387
			\sim \sim	

* Values given in study report

** Calculated values according to FOCUS, 2000 1.2 /04; Study KCA 7.1.2.1.2 /05 Study 1: KCA 7.1.2.1.2 /02 and K A 7.1 2.1.2 /03

Table 7.1.2.1.2-8: Kormalised laboratory DT50-values in aerobic soil for metabolite RPA 202248 for jose as modelling input parameters in environmental exposite assessments

Soil (Origin)		DT 50	V DTŵ	ftm *	Chi ² - Øror	t-test	VA
Ô	·0 ·	(days)	(days)	V.	(%)		
95/02, loam 20°C (S	Study	~~ 44 <i>@</i>	149	1.00	№ 1.3	<2e-16	Good
95/04, sandy loam, 2	20°CAStudy	D 10.4	~34.4 O [*]	1.00	14.7	<2e-16	Good
95/05, clay loam, 20	£6 (Study 2) 2 7.8 (⊅"92∢5	\$1.00 Å	5.6	<2e-16	Very good
95/06, loamy sand,	Ф [®] С (Study	2) 21.0	69 .7	O _{1.00}	13.5	8.20e-15	Good
95/02, loam, 10°C (Study ()	S 170°	\$\$7.4	1.60	17.0	<2e-16	Acceptable
) AA.2 A	0 ⁻ 47 <u>9</u>	₩ <u></u>	5.1	1.4e-11	Good
Mean (geometric)**		18.9					
Average 🔊	Ò			1.00			

* Geometric mean of 22.8 days calculated first for soil \$5/02.

Study V. KCA 7.1.2. K2 /02 and KCAO 1.2. 12 /03; Study 2: KCA 7.1.2.1.2 /04; Study 3: KCA 7.1.2.1.2 /05 * ffm = formation fraction. When ftm had been optimized to 1 during evaluations, the data was refitted with ffm = 1; VA = Visual assessment for acceptance of fit

TIM = 1; VA = Visual assessment for ** Includes result for soil , ster KCA 7.1.2.1.2 /06.

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Table	7.1.2.1.2-9:	Normalised	laboratory	DT ₅₀ -values	in	aerobic	soil	for	metabolite	RPA	203328	for
	1	use as modelli	ing input pa	rameters in e	nvi	ronment	al exj	posu	re assessmei	nts	(Ũ

Soil (Origin)	DT50	DT90	ffm *	Chi ² - error	t-test	VAV	F
	(days)	(days)		(%)	ð		1
95/02, loam, 20°C (Study 1)	210	698	1.00	3.8	2e-16	Excellen	1
95/04, sandy loam, 20°C (Study 2)	1000	1000	0.868	4.8	0.180	S Good	r.
95/05, clay loam, 20°C (Study 2)	2.3	7.7	1.00	16.9	🎙 <2e-16	Acceptable 🗶	<i>.</i>
95/06, loamy sand, 20°C (Study 2)	1000	1000	0.966	19.4	0.5 🗶	Acceptable	
95/02, loam, 10°C (Study 3)	40.3	134	1.00	4.8	<2e-16	Scellett	Ś
Mean (geometric)	121		Å		Z	Q A S	, 0 ⋈
Average		la A	0.96	Q″ °	~~ ^		, «

* ffm = formation fraction. When ffm had been optimized to 1 during evaluations, the data was refitted with ffm = 1; VA = Visual assessment for acceptance of fit Study 1: KCA 7.1.2.1.2 /02 and KCA 7.1.2.1.2 /03 Study KCA 7.1.2.1.2 /04; Study 3: KCA 7.1.2.1.2 /05

, III. Conclusion

The evaluation according to FOCKS kinetic gordance resulted in values for half-lives of the metabolites RPA 202248 and RPA 203328 in aerobic soil for use as modeling input parameters in environmental risk assessments

SFO was the appropriate kinetic model to derive non-normalised values for the DT_{50} . The non-normalised half-lives were then referenced for moisture (pF 2) and temperature (20%).

The values derived from taboratory tests in five EU soils are regarded as suitable and reliable for use as modelling endpoints in environmental exposure assessments.

The geometric mean of normalised half-layes was 20.3 days for metabolite RPA 202248 and 121 days for metabolite RPA 203308.

CA 7.1.2.1.3 Anaerobic degradation of the active substance

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Report:	5; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;
Title:	RPA2017 Anaerobic Agratic Metabolism
Report No:	KO16766 0 ~ ~ ~ ~
Document No: 🖉	QM-192388-04-9 0 0 0 0
Guidelines: 爷	USEPA (=EPA): No 162-36 1982) Deviation not specified
GLP/GEP	yes a sh
07	

The rate of degradation was calculated within the respective study on route of degradation in anaerobic soil (KCA 7.12.1.3/0).

The data requirement has been addressed under Point 7.1.1.2.1.4 of the Dossier submitted and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.

Within the evaluation it was concluded that isoxaflutole degraded rapidly in anaerobic soil in the laboratory. The half-life was less than 2 hours (< 0.08 days).

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CA 7.1.2.1.4 Anaerobic degradation of metabolites, breakdown and reaction products.

Report:	n; ; ;	;1995;M-	192288-01
Title:	RPA201772 Anaerobic Aquatic Metabolism	Â,	
Report No:	R016760	O'	
Document No:	M-192288-01-1	A	6 2 Y
Guidelines:	USEPA (=EPA): N, 162-3 (1982);Deviation not	specified	
GLP/GEP:	yes 🔗	Ũ	D. C. C.
	8	Q. Ø	

The data requirement had been addressed under Point 7.1.1.2.1.5 of the Dossier submitted and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments

The evaluation revealed that the pathway for degradation of Boxaflutole under actionic and anacrobic conditions is the same to form RPA 202248 and RPA 205834 as major metabolic ender actions with traces of the benzoic acid RPA 203328 For RPA 202248 and RPA 205834, hat lives of 316 days and 131 days were calculated in the study report.

It should be noted that RPA 205834 was observed at trace level 9.e. significantly below 5% AR, in aerobic soil degradation. RPA 205834 showed maximum values of occurrence at very early time points in the test (i.e. 28% after 0.25 days) thus indicating that this compound was formed from ring-opening of the active substance directly.

Keeping in mind the fast degradation of the active substance under acobic conditions and that there is time between appreciation and a potential flooding of the field, a formation and occurrence of RPA 205834 under conditions of the field is highly unlikely. Moreover, the establishment of anaerobic conditions requires time available then for aerobic degradation of the active substance during this transition period.

In addition, isoxaflutole is intended for use in corn where macrobic conditions in soil do not occur for long periods and usually not on a full field plot scale. Metabolites formed under anaerobic conditions will be degraded when the soil tarns back to aerobic conditions after a period of low oxygen content. This will prevent accumulation of metabolites is the soil. For these reasons aerobic degradation studies on specific anaerobic metabolites degradation and reaction products in soil are not required.

Moreover it is a common observation that only structural elements are converted being susceptible to reduction (i.e. nitro groups). It is thus predictable that a conversion or transformation of other structural elements like those in isoxaflutole and its residues is simply slowed-down, resulting in the observed 'stability' of the components observed under the conditions of anaerobic soil testing.

This more generally applicable conclusion can be derived from information available when screening Bayer-internal tests submitted as well as regulatory data published meanwhile in EFSA Conclusion Reports



CA 7.1.2.2 **Field studies**

CA 7.1.2.2.1	Soil dissipation studies
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CA 7.1.2.2.1 Soi	l dissipation studies				
Report:	b;	;1996	;M-234225-01	ð	
Title:	RPA 201772: terrestrial fiel	d soil dissipation	n study	S 4	
Report No:	C034014		ß	.0 "(?	
Document No(s):	Report includes Trial Nos.:			<u></u>	
	AR53-94	Ĉs	Å		
	M-234225-01-1	- V			
Guidelines:	Deviation not specified	L	Ő¥		
GLP/GEP:	yes	"O"			

Report:	p; 2001;M-2000;6-01, ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~
Title:	Isoxaflutole: Kinetic modeling analysis of a European terrestrial field soil dissipation
	study
Report No:	C015653
Document No:	M-200916-01-1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
Guidelines:	Deviation not specified
GLP/GEP:	

Report:	g 2012; 0 428564-01 ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~
Title:	Determination of the residues of AE 0540092 if on soil after spraying of AE 0540092 WP
	20 in the field in Germany, United Kindom, France (North), Maly and Spain
Report No:	10-22 5 02 O ^V S ^V Q ^V
Document No(s):	Report ingludes Trial Nos
	$\sim \sqrt{0}$
Ő	$\tilde{\mathcal{G}}$ $\tilde{\mathcal{G}}$ 10-2 $\tilde{\mathcal{G}}$ $\tilde{\mathcal{G}}$ $\tilde{\mathcal{G}}$ $\tilde{\mathcal{G}}$ $\tilde{\mathcal{G}}$
	Δ, 10 0 702-95
, ST	0° 10°2702-04 ~ ~ ~ ~ ~ ~
. 0	10-2702-05 m 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
ð í	× 10-2702-06 × × ×
<i>"</i> ۵	M-428564-01-1 A X X
Guidelines:	Commission Directive 95/36/E/C amending Council Directive 91/414/EEC
	(Annexes I and II, Fate and Behavior in the Environment), July 14, 1995; not specified
GLP/GEP:	$\operatorname{Pyes}_{\mathcal{X}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}} \xrightarrow{\mathcal{Y}}} \xrightarrow{\mathcal{Y}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}} \xrightarrow{\mathcal{Y}}_{\mathcal{Y}}} \xrightarrow{\mathcal{Y}} \xrightarrow{\mathcal{Y}}} \xrightarrow{\mathcal{Y}} \xrightarrow{\mathcal{Y}}} \xrightarrow{\mathcal{Y}} \xrightarrow{\mathcal{Y}} \xrightarrow{\mathcal{Y}} $
, S ⁷	

Report:	b; ;1997 M-234239-01	
Title:	Storage stability in soft from the European terrestrial field soil dissipation isoxaflutole and	
	metabolites study PO4/009	
Report No:	C034021 7 4 2 4	
Document No(s):	Report Acludes Trial Nos.:	
	95-857 95-857 97	
Guidelines:	Deviation not specified	
GLP/GEP:	ves a ves	
Report:	g; ;2013;M-464	592-01
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Title:	Kinetic modelling analysis of AE 0540092 and	AE B197555 from a field soil residue stude
	conducted in Europe (Normalisation to 20degre	e and pF2) $\sim \sim \sim \sim \sim$
Report No:	VC/13/007A	
Document No:	M-464592-01-1	
Guidelines:	EU Council Directive 91/414/EEC, as amend	ed by Commission Directive 95/36/ÉC of
	July 1995, Section 5, Point 7.1.1 and Commis	sion Regulation (EC) No 1107/2009 of 21
	October 2009;not applicable	
GLP/GEP:	no	

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Report:	u;	2013	;M-464899-01		
Title:	Isoxaflutole: Kinetic m	nodelling endpoin	t summary 🖉		, <u>4</u>
Report No:	VC/13/007J	- Age			
Document No:	M-464899-01-1	K ذ	5 × 4		, w
Guidelines:	Commission Regulati	ionQEC) No 1107	2009 of 21 Octo	ber 2009;not ap	plicable
GLP/GEP:	no	1. <u>0</u> ,0		~~ O'	

The data requirement had been addressed under Point 7.1 7.2.2 of the Dossier submitted and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Although all of the laboratory DT₅₀-values of the active substance were less than the specified triggers the study was performed due to variable half-liver for metabolites RPA 202248 and RPA 203328 from tests with the active substance depending on the kinetic model and method of calculation used.

A study had been therefore performed at four locations in the @U investigating the dissipation of isoxaflutole, ReA 202248 and RPA 203328 under practical conditions of the field (KCA 7.1.2.2.1/01) followed by kinetic analysis of the data (KCA 7.1.2.2.1/02)

The evaluation during the Annex, I inclusion process revealed half-lives referenced to standard conditions (20°C, pF 2 moisture) for isexaflutole, RDA 202248 and RPA 203328 in soil under field conditions at four sites in the EU Half-lives were found to range from 0.28 to 1.53 days for isoxaflutole (overage: 0.87 days), 5.83 to 10.41 days for RPA 202248 (average: 8.71 days) and from 15.1 to 56.25 days for RPA 203328 (average: 34.04 days).

KCA 7.1.2.2.1 /01 and its kinetic evaluation under KCA 7.1.2.2.1 /02 were submitted earlier within the process of Annex I inclusion? Following preview of these existing data against latest guidance in kinetic evaluation of field data (EFSA, 2610), it was decided to perform a new study in order to generate robust data for environmental risk assessments. The re-evaluation revealed, for example, that there was a lack of sampling intervals to allow for a robust kinetic evaluation.

The new field data generated under KCA 7.1.2.2.1 /03 and evaluated under KCA 7.1.2.2.1 /04 and KCA 7.1.2.2 //05 this superseded the existing field data and their evaluations. Finally, in order to fulfill latest guidance according to EFSA (EFSA, 2013) laboratory and field data were evaluated to derive modeling endpoints as input parameters for environmental risk assessments (KCA 7.1.2.2.1 /06).



Report:	y;	;2012;M-4	28564-01		<u>,</u>
Title:	Determination of the residues	s of AE 0540092	2 in/on soil after	spraying of AE	05400 2 WP
	20 in the field in Germany, U	Inited Kindom, I	France (North),	Italy and Spain	S S
Report No:	10-2702			~	
Document No(s):	Report includes Trial Nos.:			Â,	
	10-2702-01		l.	D'	
	10-2702-02		2	A.	ð Va
	10-2702-03	۵.	×"	<u>`</u> ~	
	10-2702-04	G	a) ×		Y ŵ' .
	10-2702-05	¥.	Q.	.0 .7	
	10-2702-06	Å	L, O'	Q	Ô ^y W
	M-428564-01-1	A	Q' ~~	~ <i>k</i>	Č (V
Guidelines:	Commission Directive 95/3	🖉 amending	Council Direct	five 91/404/EE	à Ű
	(Annexes I and II, Fate and	Behavior in th	e Environment	t), Juny 14, 1995	not specified;
GLP/GEP:	yes				

Executive Summary

The dissipation of RPA 202248 (SE 0540092) in soil was investigated ander European field conditions after application of the formulated test substance at a nominal rate of 100 g/ka onto bare soil at the six sites, i.e. (Germany), (Germany), (Germany), (Kather (Brited European), (Northern France), (Germany), (Cartage (Italy)) and (Spain). The sites were representative for northern and southern corregions of Europe

After spraying of 0.5 kg/ha of the product AF0540092 WP 20 onto bare soil residues were incorporated into the soil directly after treatment. At day zero soil samples were taken directly after spraying and after soil incorporation. Sampling was continued up to 413 days post-application to a maximum depth of 100 cm.

After processing of samples by lomogentisation, milling and extraction these were analyzed for residues of RPA 202248 and RPA 203328 using HPAC/MS/MS detection in the presence of stable-labeled internal reference standard.

Quantifiable residues of RPA 202248 and RPA 203328 were detected in the top 30 cm of soil at all test sites thus indicating a low poblicity in the course of the study.

At the last sampling interval residues of RPA 202248 in soil were less than the LOD of 1.5 μ g/kg soil. Metabolite RPA 203328 was detected a maximum concentrations starting from 9.11 g/ha to 32.5 g/ha within the first 70 days after treatment. The residues of RPA 203328 showed a decline below the LOD of 1.5 μ g/kg soil at all sites by DAT-89 to DAT-88. The results thus indicate the degradability and transient character of this metabolite under the conditions of the field.

In terms of best fits to measured data the dissipation of RPA 202248 in the field can be described by mono- as well as hiphasic kinetics with the bi-phasic model DFOP to be the best fit for three test sites (**1999**), **1999**

On the basis of best statistical and visual fits values of dissipation half-lives for comparison with trigger or dpoints were found to range from 8 to 40 days. The associated values for the DT_{90} ranged from 37 to 177 days.



at site. Based on the results a rapid degradation of RPA 202248 was demonstrated for a field situation at sites

AE 0540092 (RPA202248) WV 20 (Wettable Powder) fl02000024362 ...m in formulation 2009 g/kg ...m in formulation 2000 g/g/kg ...m in formulation 2009 g/kg ... ind other aga agricultural prace of the study. Priorite of the study o according to local agricultural practice. These are not considered to have an influence on the results and the outcome of the study. Prior to application the software prepared to have a fine crumb structure.



Trial No.	10-27	702-01	10-27	<u>02-02 🖉 </u>
Site:				
Designation/site	, Ge	ermany,	, U	Inited Kingdom
			Field	
Latitude				
Longitude		Ò		
Country	Germany	- Tr	Onited Kingdom	
Ecoregion	Northern EU		Northern EL	
Plot Size (m^2)	225	A ^U	444 ° 4.	
Site History	2008 Grass		2009 Qass (
Site History	2000 Grass		No plant protection	n products (ised
	No plant protection	on products used		n groadets asou
Distance from weather station	20 km	K C D	Atomal location.	A A co
used for climatic measurements	A.	r Q	Sunshinghours 1	km & Ó
			Dong-term average	es 43 km
Meteorological conditions	Overell Yes		Overfall Yes	
compared to long-term average				Û .
within normal levels (Yes/No)	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~			× 49
Other details, if any	<u><u> </u></u>		Q O O	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
Soil depth (cm)	<u> </u>	\$ 30,050 Q	9-30	‰ 30−50
Surface soil texture (USDA)	Loada	Loam .	Clay	Clay loam
Sand (%)		° 35 °	× 29 0	37
Silt (%)	Ø48 🔊	L 40 × ×	× 26 ×	24
Clav (%)	a 176		45 6	39
oH (water)	× 6,1 «	× × 5.0 °	7.8	7.4
pH (CaCl ₂)	× ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	Å 4.6 <i>∅</i>	7.2	7.2
CEC (meg/100g) 0	[™] 11.7~>	× 96	\$32.3	26.4
Chalk (% CaC 93)	& 0.3 [°]	√ . <0.1 <i>Q</i>	0.6	0.3
Organic carboor (%) \mathcal{S}	0 50 8	0.8	@ 1.9	0.8
Bulk densitx (g/cm ³)) An.d.	🖓 n.d. 🏅	n.d.	n.d.
WHC max pF 0.05 (g/100 g)	64.6	~ 515 ~	59.6	57.7
WHC per 2 (vol/a)	37,3	Ĵ [™] ≪30.2 . O	29.8	24.8
WHC pF 2.5 (volta)	× 22.0 ×	& 28.1 <u>5</u>	28.8	25.5
nitial soil moisture g/100g)	n.d. 7	O ned.	n.d.	n.d.
d = not determined				
		Dy Or		
		ð		
A, O' ôg		J.O.		
		Ň		
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	~Ç			
XY Z A				

Table 7.1.2.2.1-1: Properties of the test sites and their soils (continued)

Trial No.	10-2702-03	10-2702-04
Site:		
Designation	, France, Plot	Plot
Latitude		
Longitude	(Pa	
Country	France	Cormany & N
Ecoregion	Northern EU	Sworthern EU
Plot Size (qm ²)	369	225 0 4 0
Site History	2007 No crop 2008 No crop 2009 Noccrop Use of Roundup flash in 2007, 2008 and 2009	2008 Grass 2009 Grass No plant protection products used
Distance from weather station used for climatic measurements	5 to 6 km	At trial ocation
Meteorological conditions compared to long-term average within normal levels (Yes/No)	Overall Yes	Overall Ye
Other details, if any		
Soil depth (cm)	$\sim 0 - 30$ 0° 34 50 0°	$\begin{array}{c} 0 - 300^{\circ} \\ 0 - 300^{\circ} \\ 0 \end{array}$
Surface soil texture (USD^A)	Silt for Silt loams	Silt loam Loam
$\frac{\text{Sand}}{\text{Site}} \begin{pmatrix} \% \end{pmatrix} \qquad \qquad$		
$\frac{\operatorname{Sin}(76)}{\operatorname{Clay}(97)} = \frac{1}{2}$		
$\frac{\text{Clay}}{\text{pH}}$ (water)	19° 19° 117°	
pH (value) p^{y}	$\begin{array}{c c} & & & \\ \hline \\ \hline$	
$\frac{\text{pri}}{\text{CEC}} (\frac{\text{meg}}{1000}) = \frac{1000}{1000}$		124 123
$\frac{Chec}{Chalk} \left(\% Ca(0)_2 \right) $		0.2 0.1
Organic carbon (%)		
Bulk densit $Q(g/cm^3)$	And nd	nd nd
WHC max $(g/100@)$	54.9 \$ 30.9	54.2 46.5
WHC 55 2 (vol%)	34.8 32.7 2	31.9 26.3
WHC pF 2.5 (vel)	\$ \$\vee\$7.2 \vee\$ 27.4	23.4 20.2
Initial soil moisture (g/100 g)	n.d. n.d.	n.d. n.d.
n.d. = not determined		·

Table 7.1.2.2.1-1: Properties of the test sites and their soils (continued)								
Trial No.	10-2702-05	10-2702-0	06 🔊 🖉					
Site:								
Designation	, Italy,	, Spa	in S					
_	Az.		PLOP					
		Parcel						
Latitude	45° 21' 12.92''N 🖏	419 53' 15.09"N⊗	N ST					
Longitude	11° 11' 30.32''Е 🛛 🖓 🖉	₫° 44' 23.65''₽	<u>2 v 4</u>					
Country	Italy	PSpain 🖉 🖉						
Ecoregion	Southern EU	Southern EU	' O` _O'					
Plot Size (qm ²)	720	200 Q 0	à û					
Site History	2007 Wheat , , , , , , , , , , , , , , , , , , ,	2006 🔗 No exop	L ^a ^A					
	2008 Wheat	2009 Grass	ý ···					
	2009 Wheat C	Use of experimental f	iung teide °					
	No plant protection products used.	(BYF LAN82 SC 900)	10,2006 and					
Distance from weather station used		At trial loop in 2000						
for climatic measurements		At the Alignment of Alignment o	0					
Meteorological conditions		Querall Ves						
compared to long-term average			*					
within normal levels (Yes/No)	Spring to whiter 2010 were very	ð ₂ 0 «.	,					
~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	wet							
Other details, if any								
Soil depth (cm)	30-30 $30 = 50$	× ~0×− 30 ×	30 - 50					
Surface soil texture (USDA)	Silty clayfoam	Sandy loam	Sandy loam					
Sand (%)		\$ 547	59					
Silt (%)	× ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	33	24					
<u>Clay (%) 5</u>	<u>~~~~29~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~</u>	13	17					
pH (water)	& 7 <u>9</u> & 8.0	≪ 6.1	6.0					
pH (CaClor O	0 Q7 2 0 7.8 S	<b>5.8</b>	5.6					
CEC (meg/100g) ""		8.2	9.9					
Chalk (% carbon)		0.1	0.2					
Organic carbon (%)		0.7	0.3					
Bulk density (g/cm ³ )		n.d.	n.d.					
WHC max $(g \oplus 00 g)$	730.2 $0$ $20.2$	37.2	39.0					
WHC pF 2 (%01%) 7 2	33.0°38.9	18.8	15.5					
whe pr 2.5 (V01%) *	0 2007 0 "025.5 7 7 d 7 a nd	15./	13.5 nd					
n d = not determined		11. <b>u</b> .	11. <b>u</b> .					
	OF ST ST							
B & Study design								

#### Table 7.1.2.2.1-1: Properties of the test sites and their soils (continued)

# B. A Study design

B. Study design
B. Study design
B. Study design
B. Study design
B. Study design
C. Study des product/ha in 500 L tha spraying solution in spring (April to June) 2010. The rate was equivalent to a nominal field application ate of 100 g test item/ha. The product was applied onto bare plots followed by mechanical proporation ato the soil directly after treatment. As part of agricultural maintenance of the tield pors the emerging vegetation was either mulched by mechanical measurements or treated with a total herbicide (glyphosate) to keep the plots free of plants. Details on application are summarized in Table 7.1.2.2.1-2.

					<i>a</i> .°
Trial No.	Device	Nozzle Type	Working Width (m)	No. of Nozzles	Pressere (bar)
, Germany 10-2702-01	Knapsack-sprayer / spraying boom	Agrotop 03 110° VG 020292	1.5	673 67	
, UK 10-2702-02	Knapsack sprayer / spraying boom	Teejet Flat Fan 11 ENG SPR 1			
, France 10-2702-03	Knapsack-sprayer / spraying boom	Tecifst TJXR 5001 VS FIS11-30, FR11-137			
, Germany 10-2702-04	Knapsack-sprayer / spraying boom	≪ Agrot @ 03 ~ ↓ 46° ↓ 96° ↓ 96°	21-5 ×		
, Italy 10-2702-05	Knapsack-sprayer	AlbuzAVI 180 110 015 110			
, Spain 10-2702-06	Knapsack sprayer / « spraying boom «	Teget AIXR 1002 SPA1010			2.6

Table 7.1.2.2.1-2:	Data on the application	equipment
--------------------	-------------------------	-----------

2. Sampling and sample processing: Samples were taken for analysis following zero (pre incorporation), 0.1 (post incorporation), 24, 7, 9-11, 13-15, 20-22, 26-31, 49-66, 89-94, 119-126, 170-188 days after treatment (DAT). An additional sampling interval was taken for all sites except (UK) by 353 563 DAT.

 $\bigcirc$ The sampling spots were distributed fondomly across the plots for generating representative total samples. Sampling was performed up to a depth of 10 cm with a soil piercer (Ø 50 mm) for samples taken by day ero. At the later sampling intervals a "Wacker Hammer" (Ø 48-49.5 mm) was used to collect soil cores up to a maximum depth of 30 cm (3-4 DAT), 50 cm (13-15 DAT), 75 cm (20-22 DAT) or 100 cm (after 26-31 DAT till study end)

By day zero and the tollowing a total of 10 to 20 soil cores were taken per treated plot while 10 to 20 cores were sampled from untreated controls by DAT and 353.

The samples were deep-frozen willin 24 hours after collection and stored frozen until further processing for analysis as described under point 4,

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ő 3. Irrigation and weather data: Weather data of air temperature, precipitation including irrigation, and hours of sunshine were recorded by weather stations located at each of the test sites during the tests. The data were brieffy included in Table 7.1.2.2.1-1. Apart from temporary deviations for all sites (Italy) during spring to winter 2010 the records and some significant variables for site demonstrate sypical conditions of weather on site when being compared to their long-term averages.

4. Work up of samples and analytical procedure: After sampling the soil cores were stored frozen for processing and analysis. The frozen cores were cut into 10 cm segments with each segment being combined with those from other cores for homogenization by milling.

Sub-samples of 20 g of homogenised soil each were extracted by addition of 40 mL of an acidified (formic acid, 0.22 mL/L solvent) mixture of water/acetonitrile (1/4, v/v) and the agitation of the

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resulting slurry in a microwave extractor. Fine particles of soil were removed by centrifugation of an aliquot of 1.5 mL of the extract.

Identification and quantitation of analytical targets, i.e. the test item RPA 202248 (AE 0540092) and the benzoic acid RPA 203328 (AE B197555) was performed by use of HPLC-MS/MS coupling techniques and by their detection in MS via the Multiple Reaction Monitoring (MRM) mode. Potential matrix effects were eliminated by addition of an internal standard solution containing a stable resotope labeled reference material of each analytical target after extraction.

The method reported as BCS Method No. 01198⁷ had been validated by investigations in two contrary, soil matrices, a silt loam (**1998**) and a sandy loam (**1998**). The mass spectrometric detection used two MRM transitions for each analyte investigated. We a ratio to 2 358  $\rightarrow$  79 for quantification and m/z 358  $\rightarrow$  64 for confirmation of analyte RPA 202248. A transition m/z 267  $\rightarrow$  159 (quantification) and m/z 267  $\rightarrow$  223 (confirmation) was used for the benzoic acid analyte RPA 203328.

The method was shown to be linear of a range from 0.8 to 106  $\mu$ g/L with associated correlation coefficients ranging from 0.9991 to 99998. Overall mean recoveries were within the range of 70 to 110%. The LOQ of the method was 5  $\mu$ g/kg soil and the LOD was 1  $\mu$ g/kg for each single analytical target investigated.

Studies investigating the storage stability of RPA 202248 and RPA 203328 in frozen soil had been performed earlier with details summarized under KCA 7.1.2.2.1 (04 in the following. The results indicated stability of RPA 202248 and RPA 203328 residues in frozen soil for at least 18 months. RPA 202248 and RPA 203328 were not detected in untreased controls ( $\leq$  LOD).

The results obtained in terms of a concentration (*i.e.* µg analyte/kg of wet soil) were converted into the corresponding rates expressed as g/ha by considering the actual weight of the wet soil core and its volume for determination of the actual built density of the wet soil sample. The actual bulk density was multiplied with the concentration of the given compound in we soil and referred to one hectare area for the calculation of the corresponding rates per kectare as given in Table 7.1.2.2.1-3 and Table 7.1.2.2.1-4 for JPA 202248 and in Table 7.0.2.2.1*5 to Table 7.1.2.2.1-6 for RPA 203328.

5. Calculation of dissipation rates. Following FOCUs Guidance the dissipation of residues of RPA 202248 and RPA 203328 under field conditions was kinetically evaluated by use of the software KinGUI (Vers.1.1) for a fit of experimental data with a standard set of kinetic models available. The quality offsits was assessed by visual inspection and criteria based on the error of chi² ( $\chi^2$ ) and a t-test for testing of significance. The initial concentration of RPA 202248 was included in the parameter optimisation procedure.

## II. Results and Discussion

A. Application verification and recovery: The use of non-radiolabelled test substance implies no possibility for the determination of a complete material balance of all components including those forming NER and volatile products from mineralisation. However, it is not within the objectives of the study to establish a full route of degradation under field conditions. The focus must be therefore on

⁷ The method is detailed in report M-430249-01-1 (KCA 4.2.1 /01, Freitag & Oel, 2012).

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determination of the kinetics and thus rate of dissipation of residues as defined from route studies with radiolabelled active substance in the laboratory.

As some general point, it should be considered that, within the range of deviations observed, the overall rate of dissipation is hardly influenced by the actually measured concentrations. As a rule of thumb and from observations made in laboratory testing such influence may be given for deviations starting with a factor of five or more for the initial concentration under aerobic conditions.

Taking into account these aspects the measured initial mean concentrations of RPA202248 were well to excellently in line with the nominal application rate of 100 p/ha. The values were detailed in Table 7.1.2.2.1-3 and Table 7.1.2.2.1-4. Recoveries ranged from 70.0 g/est item/ha absite

to 87.8 g/ha at site **Example** representing 70.0 to 97.8% of the intended application rate. The range of recoveries is regarded as acceptable within the objectives of the study when considering that concentrations of residues in soil are variable in a field situation by their nature due to a number of parameters of influence and their associated errors:

- Applications are not completely homogeneous
- Potential influence from spray drift
- Following incorporation into soft homogenization is eather challenging under conditions of the field.

The dissipation of more than 50% of the actually applied amount of RPA 202248 occurred at about 15 days (2000), 66 days (2000), 21 days (2000), 26 days (2000), 28 days (2000), 20 days (2000), 26 days (2000), 28 days (2000), 20 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000), 26 days (2000), 28 days (2000), 26 days (2000)

Residues of RPA 203328 showed maximum values of 220 g metabolite/ha for site **1** at DAT-62, 9.11 c/ha for site **1** at DAT-66, 238 g/ha for site **1** at DAT-21, 9.81 g/ha for site **1** at DAT-29 (Table 7.152.2.1-2 and Table 7.152.2.1-6). For all sites a decline of residues below the LOD of 1.5 µg/kg soil was observed for RPA 203328 starting from DAT-89 at site **1** as for RPA 202248 quantifiable residues of RPA 203328 were detected in the top 10 to 20 cm of soil thus indicating a fast degradation associated with low mobility in the course of the study.

Table 7.1.2.2.1-3: Residue data (g/ha) for the dissipation of RPA 202248 (A)	E 0540092) in soil under	field	ð
conditions for sites , , , , , and , , , , , , , , , , , , , , , , , , ,	ý Na l	5.	O,

						(())	(C.	// ×	
10-	(Germany)			(UK)			(France)		
DAT (days)	(g/ha)	(%)*	DAT (days)	(g/ha)	(%)*	DAT (days)	(g/ha)	Q%)**	
0**	73.1	73.1	0**	87.8	<u>چ</u> 87.8	×0**	×81 ~	81	a
0.1	62.3	62.3	0.1	80.7 🕅	80.7	0.1	71.3	74,3	Ç
3	45.9	45.9	4	52.4	52.1		K 49 B	A9.0 0	) Ĩ
7	52.4	52.4	7	74.9	71.7	.7	72.5	5 72.5 Ø	
10	40.2	40.2	10	<b>8</b> .4	48.4		. 68.9	68.9	
15	27.1	27.1	14	‴75.9 。	75.9	~~ 14~	45.6	45.6	
21	25.1	25.1	21 💃	v 57 🖗	م [∞] ₹7.4	<u></u>	41.8	<b>¥</b> 1.8	
29	30.1	30.1	31	\$5.9	^ک 55.90	≥ ²⁸ 0	22.8 🔎	22.8 °	
62	12.6	12.6	66	S \$5.9	35.9	49 <b>C</b>	91.9	1109	
93	6.94	6.9	94	× 25.2	<b>23</b> .2 ,	⇒ <u>9</u> 20	≪J 3.87 [™]	\$.9	
120	4.85	4.9	Ø6	6,99	°~~~6.9_0	Al 9	1,1,3	<u>1.1</u>	
170	1.13	1.1	6¥80 %, [™]	<b>,</b> ¶ 3 _^	\$ 1.h	~ ^O 188 &	<i>⊈</i> ŽÓD _⊘	-	
353	<lod< td=""><td>-</td><td>S - 0</td><td>× - , ×</td><td></td><td><u>کې</u> 358</td><td>S LOD</td><td>-</td><td></td></lod<>	-	S - 0	× - , ×		<u>کې</u> 358	S LOD	-	
DAT: days after	treatment		N Q V		l ő		0 Y		

* Values refer to a nominal rate of 190 g test item /ha

*S* ** Initial concentration for samples before soil incorporation

Ô Table 7.1.2.2.1-4: Residue data (4/ha) for the Arssipation of RPA 202248 (AE 054002) in soil under field conditions for sites and

	(// 1)							
	(Germ	anty) 🔨		( <b>IQ</b> Iy)	9 2		(Sp	ain)
10	)-2702-04	$\sqrt{4}$	<u>ال</u> ہ م	)-2702205	S Q	<b>1</b> 0	-2702-06	
DAT (days)	gg/ha)	` (%)*	<b>BAT</b> (days)	(g/ha) 🗞	· (%)	DAT(days)	(g/ha)	(%)*
0**	70. <b>®</b>	\$9.8	0** >	76.5	76.5	0**	70.0	70.0
0.1 🔊	53.6	لا پ 53.6	Ű K.D.	* 84.3	_@84.3_ @	° 0.1	54.4	54.4
3	51.7	51	۵ × ۵	<b>64</b> .7	\$ 64.7	3	45.4	45.4
4×	55.6	\$5×6	× 7 ×	×83.9	83.9	7	40.8	40.8
11	50.9	\$0.4		60.6	60.6	10	27.7	27.7
14	63.2	A 63.2€		64.8	064.8	14	30.6	30.6
20	45.8	45	. 8 22	5.3	\$ 55.3	21	15.6	15.6
26 🔌	ð 37. <i>3</i> 0	<u></u>	28	`∕≫ 39.3	39.3	29	9.45	9.5
60 🐧	1.13	01.1		7.28	7.3	59	1.13	1.1
89	<lod &<="" td=""><td></td><td>@90 ×</td><td><u>↓</u>1.¥3</td><td>1.1</td><td>93</td><td><lod< td=""><td>-</td></lod<></td></lod>		@90 ×	<u>↓</u> 1.¥3	1.1	93	<lod< td=""><td>-</td></lod<>	-
126	<lod>&gt;</lod>		~ 120	_ ~©ĽOD	-	128	<lod< td=""><td>-</td></lod<>	-
<b>∡</b> ¥∕⊅5	<lqd< td=""><td></td><td>0° 189 ~</td><td>LOD</td><td>-</td><td>178</td><td><lod< td=""><td>-</td></lod<></td></lqd<>		0° 189 ~	LOD	-	178	<lod< td=""><td>-</td></lod<>	-
363	<lod< td=""><td>Ø - Ö</td><td>360 0</td><td><pre>LOD</pre></td><td>-</td><td>351</td><td><lod< td=""><td>-</td></lod<></td></lod<>	Ø - Ö	360 0	<pre>LOD</pre>	-	351	<lod< td=""><td>-</td></lod<>	-

DAT: days after treatment * Values refer to a normal rate of 100 sylest item/ha ** Initial concentration for samples before soft incorporation

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Table 7.1.2.2.1-5: Residue data (g/ha) for the dissipation of RPA 203328 (AE B197555) in soil under field

-		ŕ		
	(Germany)		(UK)	(France)
10-27	702-01	10-2	2702-02	10-2702 <b>-0</b> 3
DAT (days)	(g/ha)	DAT (days)	(etha)	DAT(days) (g/ha)
0	<lod< td=""><td>0</td><td><b>X</b>LOD</td><td></td></lod<>	0	<b>X</b> LOD	
0.1	<lod< td=""><td>0.1</td><td>LOD (</td><td></td></lod<>	0.1	LOD (	
3	1.13	4	LOD S	
7	5.74	7	Correction <	
10	6.63	10	1.13	2 2 × 1.70 ~ ×
15	6.69	14 🐇	\$ 4.54	× 14 × 14.5
21	11.6	21 O	× 502 0	
29	19.5	314	© _ &.34 ~ ~	
62	22.7	60 ~	9.11	13.7 S
93	13.1	Ø4 ú [*] ~		92 \$ 6.8 ⁴
120	9.08	%126%_ [™]	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
170	1.13	× 1800 °	∀ ¥LOD∧y	5 188 5 SLOD
353	<lod< td=""><td>× &amp; Ø</td><td>à- à</td><td>D ST SKLOD</td></lod<>	× & Ø	à- à	D ST SKLOD
DAT: days after to	reatment			

Table 7.1.2.2.1-6: Residue data (g/ha) for the dissipation of RPA 203328 (AEB197555) in soil under field conditions for sites

				<u> </u>	
10-27	(Germany) 702-04	2 2 2 10-2	(Italy) 2702-05	○ [×] <i>Q₁</i> 10-2	(Spain) 702-06
DAT (days)	E (gyha) i	DAT (days)	(gAbă) 🖉	<b>D</b> AT(days)	(g/ha)
0	LOD O		U HOD S	0	<lod< td=""></lod<>
0.1	°∕~LOB∕v	0.1° .~~	KLOD ^O	0.1	<lod< td=""></lod<>
3 🦿	<lqd td="" 🔬<=""><td></td><td>3.78</td><td>3</td><td>8.89</td></lqd>		3.78	3	8.89
20	<u>j</u> Q13	× ، لاکھ		7	15.5
11	×4.20 ×	× 165° ×	7.14	10	15.0
14	\$\$ 7.66 ₹		8.68	14	24.1
20	6 941 8	22 v	1004	21	21.8
26	AN14 2		Å.13	29	32.5
60 🔊	©1.13		2.3	59	24.3
89 🧃	<lod td="" ~<=""><td></td><td>0 1.13</td><td>93</td><td>4.69</td></lod>		0 1.13	93	4.69
1200	<body> <bod< td=""><td>Ø"120 ×</td><td>,≪ <lod< td=""><td>128</td><td>1.13</td></lod<></td></bod<></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body></body>	Ø"120 ×	,≪ <lod< td=""><td>128</td><td>1.13</td></lod<>	128	1.13
175	, ŠĽOD, Š	~ 18 <b>2</b> ~	D [≫] <lod< td=""><td>178</td><td><lod< td=""></lod<></td></lod<>	178	<lod< td=""></lod<>
<i>"</i> ≪3⁄63	LOB .	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	<lod< td=""><td>351</td><td><lod< td=""></lod<></td></lod<>	351	<lod< td=""></lod<>

DAT: days after treatment @

C. Kinetics of dissipation of RPA 202248 (AE 0540092) in soil: The kinetic evaluation of the dissipation of RPA 202248 in soil under field conditions resulted in best fits represented by the biphasic model DPOP for three sites. For the other three sites SFO was determined to be the best fit to the measured data a sit is summarized in Table 7.1.2.2.1-7.

Apart from visual assessment the chi²-scaled error 'quantifies' the quality of fit. For best fits its values ranged from 7 to 15% to result dissipation half-lives ranging from 8.2 to 40 days. The associated values for the  $DT_{90}$  were 37 to 177 days. These values may be compared with trigger endpoints in environmental exposure assessments.

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Table 7.1.2.2.1-7: Non-normalised half-lives from kinetic analysis of dissipation data of RPA 202248 (AE 0540092) in the field for comparison with trigger endpoints								
Location and Trial No.	Kinetic Model	DT50 (days)	DT90 (days)	Visual Assessment *	Exror for Chi ²			
,	SFO	17	58	d,	. O ^v 150 √			
Germany	DFOP	14	<b>Č§</b> 1	× +				
10-2702-01	FOMC	14	₹115	+				
, UK	SFO	50	C 165	0 ⁸ 0 %	15 5			
10-2702-02	DFOP	40 🛒	177 🦼		, [™] 15, 0			
	FOMC	48	175					
, France	SFO	21	。 69. 💞		√v~15 ₂ √v			
10-2702-03	DFOP	21	Q 69 ⁰		≫ ₁₇ ≫			
	FOMC	21 %	ĴĬ (		× 14			
, Germany	SFO	29	<b>∞</b> 97 ~		<b>315</b>			
10-2702-04	DFOP 🔏	29/	> 97 [°] O [°]		17			
	FOMC	×29 ×	- <b>9</b> 9		<u></u> 160			
, Italy	SFO O	28, 5	<b>\$92</b> 0		A2			
10-2702-05	DFOR	28	92	à 30 °C	× _ × 13			
	FOMČ	27	940	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	12			
, Spain	SFO 'N		. <u>3</u> 4 <i>6</i>		× 11			
10-2702-06	[∞] ØFO₽	<b>8.2</b>	37 🖧	× + 6	7			
	6 FOMO	F <u>1</u> Ø	40	S Q X	11			

Best fits from overall evaluation are marked bold

* Evaluation of fits as good (+), spedium(0) or bad (-)

#### LI Conclusion

There was no indication for a significant movement of residues of RPA 202248 and its metabolite RPA 203328 into layers below the top 30 cm of soil. The incorporation into soil directly after treatment assured the exclusion of RPA 202248 and RPA 203328 in soil was indicated by the results of laboratory investigations. Plant optake was minimal due to application to bare soil and by keeping plots free of vegetation of the source of the study.

Residues of KPA 202248 in soil were below the LOD of 1.5 μg/kg soil at all sites by DAT-353 ( DAT-180 (DAT-180 (DAT-90 (DAT-90)(DAT-90 (DAT-9

As tor RPA 202248, quantifiable residues of RPA 203328 were detected in the top 10 to 20 cm of soil. Maximum values of RPA 203328 were detected within the first 70 days after treatment starting from 9.11 g/ha to 32.5 g/ha. The results showed a decline below the LOD of 1.5  $\mu$ g/kg soil at all sites by DAT-89 to DAT-188. The results indicated the degradability and thus transient character of this metabolite under conditions of the field.

In terms of best fits to measured data the dissipation of RPA 202248 in the field can be described by mono- (SbO) as well as bi-phasic kinetics with the bi-phasic model DFOP to be the best fit for three sites (**1999**, **D**, **1999**, **UK**, and **1999**, **SP**) while it was the SFO model for the other three sites (**1999**, F, **1999**, D, and **1999**, I).

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On the basis of best statistical and visual fit values of dissipation half-lives for comparison with trigger endpoints were found to range from 8 to 40 days. The associated values for the  $DT_{90}$  ranged from 370to 177 days.

Based on the results a rapid degradation of RPA 202248 was demonstrated for a field situation representative for the northern and southern EU.

The kinetic evaluation of field dissipation data was performed separately for the determination of endpoints for use as modelling input parameters (KC& 7.1.2.2.1/05).

Report:	x; (1997; M) 234230 01 0 4 4 4
Title:	Storage stability in soils, from the European terrestrial field soil dissipation is exaflutore
	and metabolites study P94/009 🦴 🖉 🖉
Report No:	C034021 $Q'$ $Z'$ $Q'$ $Z'$ $Q'$ $Z'$ $Q'$ $Q'$ $Q'$ $Q'$ $Q'$ $Q'$ $Q'$ $Q$
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	M-234239-01 1 6 6 6 6 7 6 6 7 6 7
Guidelines:	Deviation and the specified a second specified a second specified a second se
GLP/GEP:	yes y y y y y o

of the active substance coxaflatole and its metabolites RPA 202248, The storage stability RPA 203328 and RPA 205834 in soil under freezer storage conditions as used for samples from field trials was assessed

Untreated soil from four European test sites Fy, Germany (D) and (UK) of a field dissipation study was used for this storage stability study. Soil samples were fortified each with 50 µg/kg of active substance isoxaflut@te, RPA 202248, RPA 203328 and RPA 205834

Soil samples were malysed by day 0 and after 3 12, and 18 months of storage in the freezer at 6, 8 <-18 °C.

Soil samples were analyzed for the four analytical targets according to method AR 106-948. Identification and quantitation of the analytical digets was performed by reversed phase high performance liquid chromatography (HPLG) using UV detection at 270 nm or 300 nm with a limit of quantitation (LOQ) of 5 µg ag soil

The mean recovery during analyses of the samples was 101 to 104% for the active substance isoxaflutole (n = 7, for each soil), 92 to 97% for RPA 202248 (n = 6), 99 to 83 to 103% for RPA 203328 (n  $\approx 6$ ) and 93 to 100% for RPA 205834 (n = 6). Consequently, an average of clearly more than 70% of the initial forthied amount was recovered from the stored samples for all sampling intervals and for all potential analytical targets.

The recovered amounts in four soils after 18 months (approx. 540 days) of storage in a freezer ranged from 98 to 105% for the active substance isoxaflutole, 89 to 100% for RPA 202248, 71 to 106% for

⁸ The analytical method has been submitted in the existing dossier and evaluated during the process of Annex I inclusion (see KCA 4.1.2, doc. M-201587-01-1).

RPA 203328 and 85 to 103% for RPA 205834. The results are summarized in Table 7.1.2.2.1-8 to Table 7.1.2.2.1-11.

Table 7.1.2.2.1-8: F	Recoveries of act	ive substance	isoxaflutole in soil after	r frozen storage
Soil	Month	Recovery	Overall mean	
		[%]	recovery [%]	
(I)	0	101*		
	3	108	Č	4 4 X X
	6	98	104	
	9	104		
	12	105	st i	
	18	105		
(F)	0	110*		N & N N
	3	101		
	6	99 🦼		
	9	95		AGING
	12	105		
	18	85 . 4		
Germany (D)	0	Q104* 💥		
	3	Q* 107	6 6 2	jo o o v
	6			Y D LO K,
	9	1904	, w k o	
	12	<u>% 105 6</u> ^y	n or	
/* *`	180	98		
(UK)				
		× 112		
	6		~~ 1 <u>0</u> 3~ _@	
				A Sa
		0 03 60		
* Mean values of tu	vo			
		ĭ À	````_@^	Ø
Table 7.102.2.1-9: F	Recoveries of RP	A 202248 in <u>s</u>	oil after frozen storage	/ 
ioil "	Monthy '	Recovery	verall mean recovery [%]	
(I)		 		
	. 646 2	<u></u>		
~9	0 90	107	§ 97 97 €	
4	120 2	) <u> </u>	çı ç	
Ø.		<b>(89</b> )		
(F)		>> 99*√	\$ <u></u>	
~	× 65° ?	90		
	ý ý	_98 x	99	
L.	4 V12 0	~~M01 Q		
	√ 18 8 [*]	^{~~} 100		
Germany (D)		) <del>39*</del>		
		85		
<u> </u>	<i>₽ 9 ₹</i>	87	92	
	~ 1 <u>2</u> ~	96		
	18	91		1
(UK)	0	102*	93	
	6	106		
	9	91*		<u> </u>

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12	74
18	94

#### Table 7.1.2.2.1-10: Recoveries of RPA 203328 in soil after frozen storage

	12	74		0
	18	94		
* Mean values of tw	vo replicates		•	'
	-			
T 11 71 2 2 1 10	D · (D	DA 202220 ·		
1 able /.1.2.2.1-10:	Recoveries of R	PA 203328 In	soll after frozen storage	
5011	Nionth	recovery	overall mean	
	0	105*		
(1)	6	88	. ¥*	
	9	70	R I	
	12	82		
	18	71		
(F)	0	105*	<u>k</u> , o 2	
(1)	6	93		
	9	105	. @103. @ Q	
	12	106		
	18	1.05 0		
Germany (D)	0	₫¥%		
	6	85 0		8 8 5 V
	9	× 11,50	\$ 100° 5°	
	12	°~97		
	18 🔊	<u>(106*</u> )		
(UK)	0 Ø	O 108		
	6× 1	86 4		
	<u>⊸</u> 9 §°'	<i>≈</i> ² 95* [©]	Ø97 ≿∽ Ö [×]	& 27
	12 ¹⁰	<u>م</u> 89 گ	ĴĴ Ĵ .	O' Ý
	S 187 ×	) 1 <b>06</b> /		ý "Q
* Mean values of to	vo replicates 😽			
Table 7 1 2 2 1 1.	Receiveries of R	PA 205834/in	soil after frazen starag	
Soil 0	Menth	Recevery	Overall mean	
				D.
		105*		
(1)	S 6 €	Q 99		
Č		104 ×		
	A2 ~	Č 94		
~~	0 18 0	98		
		1.95*	ĝ, ŭ	
Ô,ª	<u>,</u> %6 , Q	Ø8 🔊		
	~~ 9 1	> 100	<u>,</u> \$\$`100	
L.	1200 2	<u>96</u>		
v o		103° a	P ⁻	
Germany (D)		_^¶03* Q́		
Q^′	<u>~ 6 8 </u>	× 85		
(		/ 80*	93	
		100		
		95		
(UK)	O ^y Qy	103*	ļ	
~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Ő	105		
Č	9	90*	96	
	12	96	4	
1	18	85		

* Mean values of two replicates



Conclusion

The storage stability of the analytical targets isoxaflutole active substance and its potential residues in soil, RPA 202248, RPA 203328 and RPA 205834, under frozen conditions has been demonstrated for a maximum storage time of 18 months.

Kinetic evaluation of field soil dissipation study KCA 7.1.2.2.1 /03

Report:	q; ;20187;M-464592-04
Title:	Kinetic modelling analysis of AE 0540092 and AFOP197555 from a field will residue study
	conducted in Europe (Normalisation to 20degree and pF2)
Report No:	VC/13/007A
Document No:	M-464592-01-1
Guidelines:	EU Council Directive 91/414/EEO, as a mended by Commission Directive 95/36/EC of
	July 1995, Section 5, Point 7.1, Pand Commission Regulation (EC) No 1102(2009 of 21
	October 2009;not applicable V V V V V
GLP/GEP:	

Executive Summary

A kinetic evaluation was conducted for residue data of metabolites RPA 202248 (AE 0740092) and RPA 203328 originating from field dissipation study conducted a six European sites (KCA 7.1.2.2.1 /03) to derive modeling endpoints for use in environmental exposure assessments. The evaluations were performed according to FOCUS kinetic guidance with the software tool KinGai, version 2. Residue data of RPA 202248 and RPA 203328 were the step reference for temperature and moisture (20°C, pF2 moisture) to result in normalised values for the DT_{50} . The calculations included daily soil temperatures and moisture contents as supported by the model FOCUS PEARL, version 4.4.4. The normalisation of daily weather data followed the use of the Arrhenius approach for temperature correction (Q₁₄ of 2.58) and the Walker equation for moisture correction (B-factor of 0.7).

The evaluation resulted in simple first order (SFO) model fits each applied to the time-step normalised data sets of RPA 202248 and RPA 203528 for five test sites with the only exception for site

Following the use of an triverse Hockey Stick (HS) approach to RPA 202248 data, this model was found to describe the situation best for site **constant**. The fits were evaluated on the basis of detailed statistical analysis including visual assessment, chi²-error statistics, significance t-test and correlation analysis. Optimisation of fits to measured data was performed by use of bi-phasic kinetic models. The results are summarized in Table 7.1.22.1-12.

The overall geometric mean normalised (20°C, pF2 moisture) half-life in the field was 15.5 days for RPA 202248 and 11.4 days for RPA 203328.

 Table 7.1.2.1.2.12
 Normalised G0°C, pF2 moisture) field DT50-values for RPA 202248 and RPA 203328 in

 Soil for use as modelling input parameters in environmental exposure assessments

N.S.	Parageter 0 3	RPA 202248	RPA 203328
R.	DTG, range (days)	10.2 - 29.8	2.0 - 43.3
	Tean (geometric) DT ₅₀ (days)	15.5	11.4
	Formation fraction from RPA 202248	-	0.92
	(average)		

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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

I. Material and Methods

Residue data were pre-processed by setting values between LOD (1.5 μ g/kg per single analyte) and LOQ (LOQ = 5.0 μ g/kg) to the measured values. Values < LOD were set to 0.5 LOD, in cases where one sampling date earlier or later the residue at the same depth had been >LOD, or 9 at the same sampling date the residue in the soil layer above had been >LOD. The surve was cut off after the first non-detect (< LOD), if no later value > LOQ followed. At day zero, salues being < LOD in the deeper of soil layers were set to zero.

Actual values for residues of RPA 203328 were converted to parent composed equivalents prior to further data processing.

Daily values of soil temperature and moisture were derived from daily weather data on precipitation, irrigation, maximum and minimum air temperature, humidity or vapour pressure, wind speed and global solar radiation by simulation with the software FOCUS PEARL. The use of the model assured that the mass balance is correct and that meaningful soil fridraulic values are obtained. With ainfall and potential evapotranspiration being the main driving forces for soil pore water flow these can be measured or calculated with adequate precision respectively.

The time step normalisation approach considered actual conditions in the field from study weather data in combination with the model used for evaluation to result in a correction for the degradation kinetics in ever changing outdoor conditions. The approach resulted in degradation rates and half-lives at reference conditions to be readily transferred to other climatic conditions (FOCUS, 2000). Normalisation was performed by use of a Q10 of 2.58 (average activation energy of 65,400 J/Mol) for temperature and a B-factor of 0.7 in the Walker equation for provide correction.

Following time-frep normalisation of residue data in principle, bur kinetic models could be applied for fitting, *i.e.* Completing of the compartment (FOMC, Gustafson-Holden), hockey-stick (HS) and double-first order in parallel (DFOP)

The identification of the appropriate finetic model to describe the field residue data was based on the three major criteria visual inspection, the scaled chi²-error ϵ and the significance t-test.

Following the approach by FOCUS. The use of SEO is preferred as the first step of evaluation. In case of its visual acceptability, $r_{c}chi^{2}$ refror ϵ not significantly exceeding 15% and low values for the t-test, no bi-phasic models were to be pested $\frac{1}{2}$ $\frac{1}{2}$

I Results and Discussion

The application of the simple first-order (SFO) model resulted in acceptable fits to measured data for RPA 202248 for all sites with the exception for site **second second**. The use of the SFO model resulted in acceptable fits for metabolite RFA 203328 in all cases.

With the combined 'all-SFQ approach for RPA 202248 and RPA 203328 failing acceptance criteria for site **and the fitting** to measured data was optimized by applying a simulated lag-phase within an inverse Hockey-Stick (HS) kinetic model for RPA 202248 and the SFO model for RPA 203328 data.

Optimized fits were thus obtained for the combined modeling from use of an 'all-SFO' approach for five sites and the combination of a bi-phasic approach (HS model) for RPA 202248 with the SFO



model for site **Example**. The results are summarized in Table 7.1.2.2.1-13 for RPA 20248 and Table 7.1.2.2.1-14 for RPA 203328.

 Table 7.1.2.2.1-13: Kinetic modeling parameters derived for RPA 202248 from field data after fine-step normalisation to reference conditions (20°C, pF2 moisture)
 Second seco

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Site	Kinetic	DT ₅₀ field	DT90 field	Chi ² - error	t-test	S VAS	Ô
	model	(days)	(days)	(%) 🔬	^		ſ
, D	SFO	15.6	51.9	21.6	<0.001	Sood O	,©
, UK	SFO	29.8	99.1C	17.0	<0.000	Goge	
, F	SFO	15.2	50 5	1∳.0 ⊘°	لي 001€	Good 🔍	
, D	HS	10.2	\$26.1	∕≫18.5 Ø	~~~0.001 ⁰	Acceptable	
, I	SFO	17.5	ζ, 58.1 _€ ,°	⇒ 10,3 ×	0° <0.001	Excellent	
, E	SFO	11.2	O* 37@*		< 001	Excellent	
Mean (geometric)		15.5			O [×]		

VA = Visual assessment for acceptance of fit^{*}

Table 7.1.2.2.1-14: Kinetic modeling parameters derived for RPA 203328	from field	data	after time-step
a compation to a first on a disting (200C a F2 state	Ĉ	Ň	
normalisation to recerence conditions (2020, pr amoistane)		ຶ້	°∼√

				- All All All All All All All All All Al			
Site	Kinetic 🚕	PT 50 field	DT90 field	∂ffm *	Chi error	f f est	VA VA
	model 🗶	(days)	Adays) _			Ŭ (A)	
, D	SFO	29.6	S~98.4€	1.000	17.0	<0.00₽	Very good
, UK	SFO∕	j 13.7	456	A.645 S	21.5	× <0,001	Acceptable
, F	SFO A	¥14, 3 Q	Q.6	0.853	\$23.9 ¢	₀ 0907	Good
, D	SFO Ø	20	@ 6.6 🔊	1.000	1.80	0.001 کرک	Excellent
, I	SFO SFO	<u></u> , 0∳.4 ~	¥ 14.7	1 30 00	© 9 <u>.</u> 4	< 0.001	Very good
, E	₽ SĘΘ	<u>√</u> ¥3.3 ~	1447.0	°¢\$1.000 €	28 .0 ~	< 0.001	Acceptable
Mean (geometric	<u></u>	11	w a	ÿ 🂫			
Average	<u> </u>	<u> </u>		002	0 0		

* ffm = formation fraction. When fim had been optimized to 1 during evaluations, the data was refitted with ffm = 1 VA = Visual assessment for acceptance of fit $\sqrt{2}$

= Visual asse	ssment for	accepta	nce of f	it 🔊 .			
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Following FOCUS kinetic guidance the geometric mean referenced (20°C, pF2 moisture) half-lives (DT_{50, free}) for the degradation of RPA 202248 and RPA 203328 under field conditions were calculated to 15.5 days and 11.4 days respectively. For the formation of RPA 203328 from its predecessor metabolite FPA 202248, an average value for the formation fraction of 0.92 was calculated.

The values are regarded as suitable endpoints for use in modelling of residues of RPA 202248 and RPA 203328 in environmental exposure assessments.



Overall summary of rate of degradation in aerobic soil for use as modelling input parameter in environemental risk assessments

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Report:	u;	;201ँ3;M	-464899 😡	, Ø	
Title:	Isoxaflutole: Kinetic mode	lling endpoint su	immary		
Report No:	VC/13/007J	A	Q. %.	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
Document No:	M-464899-01-1	Q0	\sim 0°	~~ <u>`</u> 0`	Ø Ø
Guidelines:	Commission Regulation	EČ) No 1407/20	99 of 21 Octob	er 2000 not apr	nlicable
GLP/GEP:	no	Y O V			4

The report summarized the kinetic evaluations of aerobic soil degradation data from studies performed under laboratory and field conditions for terms of normalised (20°C, pF2) half-lives and their DegT₉₀ under KCA 7.1.2.1.1 /07, KCA 7.1.2.1.2 /08 and KCA 7.1.2.2.9 /05. The evaluation was performed according to EFSA Guidance for the active substance isovarilutors and metabolites RPA 202248 (AE 0540092) and RPA 203328 (AE B197535) to result in appropriate Deg 5_0 values for use as modelling endpoint in environmental risk assessments.

For the active substance isoxathutole, a geometric mean normalised DegTS of OP days was calculated on the basis of laboratory data and in the absence of field data. The value was regarded as acceptable for use in modelling evaluations.

For metabolite RPA 202248, a geometric mean normalised DegT of 155 days was calculated on the basis of field data available. For comparison with laboratory data an alpha factor of 0.25 was used. The evaluation showed that the Test confirmed that field studies showed a shorter DegT50 than laboratory studies'.

For metabolite RPA 203328, a geometric mean normalised DegT₅₀ of 11.4 days was calculated on the basis of field data available. For comparison with laboratory data an alpha factor of 0.25 was used. Again, the evaluation showed that the Test confirmed that field studies showed a shorter DegT50 than laboratory studies'.

The overally results are summarized in Table 7.5.2.2.1.5

Table 7.1.2.2.1-15: Summary of mean normalized (20°C, pF2 moisture) DegT₅₀-values for isoxaflutole, RPA 202248 and RPA 203328 for use as modeling endpoints

Compound	Mean BegT ₅₀	♥ Formation fraction (ffm) (-)
		()
Isoxafluto	0.8	-
RPA 202248	A 1\$.5	1.00 from isoxaflutole
RPA 203328	<u>مَ</u> الم	1.00 from RPA 202248
	L.	

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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

CA 7.1.2.2.2 Soil accumulation studies

The data requirement was addressed under Point 7.1.1 2,3 of the Dossier submitted and en within the process of Annex I inclusion as published in the corresponding Monograph of RMS Netherlands (February 1997) and its amendments.

The evaluation revealed that the values for the DT_{90} of isoxal utobe from laboratory and field the performed were all significantly less than one year thus with no indication for accumulation of the active substance in the soil environment. ubstance active substance in the soil environment

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- Adsorption and desorption CA 7.1.3.1

CA 7.1.3.1.1 Adsorption and desorption of the active

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Report No:	M-06/5963-	Ø \$-1 "S	× ~ ~	õ de	, a a	71		
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GLP/GEP: O	🛛 yes 🟑	&	× Q	Å C				
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to soil was investigated under conditions of the The adsorption of laboratory in:

four soils and a sedment under standard conditions of batch equilibrium tests following application of phenyl@L-14@labeled active substance (KCA 7.1.3.1.1 /01).

The data requirement had been addressed under Point 7.1.2.1 of the Dossier submitted and evaluated within the process of Appex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.

The evaluation within the Amer I inclusion process revealed that the active substance isoxaflutole was moderately adsorbed to soil. Values for the adsorption K_{F.OC} ranged ranged from 93 to 136 mL/g while values for Freundlich coefficients 1/n were from 0.90 to 1.00 when excluding the sediment. Following the DAR the corresponding $K_{F,OM}$ ranged ranged from 54 to 79 mL/g.

Re-calculation of adsorption to soil data for the active substance:

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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

However, it was found later that values of K_F, K_{F,OC} and 1/n had not been calculated correctly due to non-consideration of interstitial water remaining in the soil after centrifugation at each adsorption and desorption step performed in the study. The data were thus re-evaluated with details of re-evaluation? reported in the amended original report under KCA 7.1.3.1.1 /01:

The adsorption/desorption characteristics of [phenyl-UL-14C]isoxaflutole (RPA 20 determined for four soils and a sediment under standar conditions of batch equilibrium tests and concentration range of two orders of magnitude (1.5, 0,75, 0.15 and 0,015 mg/L) at 20°C in the dark.

In the definitive test the overall mean recoveries anged from 97.2 to 92.8% AR% for the sand spil 93/24, 100.4 to 105.6% for sandy loam soil 93/7, 97.3 to 100.8% for clap loam soil 93/30, 963 to 100.2% for silty clay soil 93/32 and 95.8 to 100.9% for the boam sediment 93/31 for all test concentrations and thus found to be in the acceptable range.

mL with The revised Freundlich adsorption $coefficients (K_F)$ ranged ′from≥0.24@to to, \$49 mL/g corresponding values referenced to organic without K_F, tosrange (Table 7.1.3.1.1-1). Values for the Freundlich coefficient of actorption 1/n ranged from 0.87 to 1.03.

Soil Soil				CEC		Ads Koc (mL/g)	Ads 1/n
Sand, US (93/24)	\$ 0.5g		₹ ⁷ 7.2	5.ª	\$0.24	48	1.03
Sandy loam, OK (93,5)	0.9	6.3		6 4.9 . ∅	0.84	93	0.87
Clay loans UK (93/30) 🔊	\$2.4 €	22.4	6.9	12.9	1.68	70	0.91
Silty (189, UK (93/32)	7.5	A3.3 ~	9 4.8 Y	\$.7	8.15	109	0.94
Sediment, UK (93,37)	8.7	× 15.4	5 .8	13.0	13.0	149	0.91

Table 7.1.3.1.1-1: Sorption behaviour of isoxaflutole (RPA 201772) in four soils and a sediment

Values for sediment no considered

desorption of metabolites, breakdown and reaction Adsorption and

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$\sqrt{2}$ $\sqrt{2}$ $\sqrt{3}$; Agrended $\sqrt{2}$ $\sqrt{2004}$ $\sqrt{2}$ $\sqrt{2}$	
Title: [14]-RPA 202248 Adsorption/desorption to and from four soils - Addendum report	
Report North CX/03/020A	
Document No: M-06655-03-1	
Guidelines: y 91/414/EEC; 95/36/EC;Deviation not specified	
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Report:	\$; ; ;	;	;1996;M-
	066483-01		
Title:	[14C]-RPA 202248: Adsorption/desorption to and from	n four soils a	nd an aquatic sediment
Report No:	11486	~	
Document No:	M-066483-01-1	Â,	
Guidelines:	EPA, Subdivision N, 163-1; Deviation not specified	O,	
GLP/GEP:	yes	A	
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Report:	ü; ; ; ; ; ; 1996; W 209743-02
Title:	RPA 203328: Adsorption/desorption to and from four soils
Report No:	C025651
Document No:	M-209743-02-1
Guidelines:	EU (=EEC): 95/36/EC Annex 1, Section 7, 12; OECD: 106; USEPA (=EPA):
	Subdivision N (1982); Deviation por specified and the specified in the spe
GLP/GEP:	yes O V X X A

Report:	1 ,5, 1 ,2,9,6,1,2,5,8,6,5,1,9,4, ○ ○ √ √ √ √ √
Title:	[14C]-RPA20332& Adsorption / Cosorption to and from four soils and a sediment
Report No:	R000447 Q 4 Q 4 Q 4 Q 4 Q 4 Q 4 Q 4 Q 4 Q 4
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	M-15865 p01-1 x & X & V & V & V
Guidelines:	USEPA EPA) N, 163-1, (1982); Devration ovt specified
GLP/GEP:	yes y of the or the state of the

Report:	2; ;2@2;M-429399°01
Title:	[cyclopropy]-14CFAE 0540092: Adsorption/desorption in five different soils
Report No:	$A_{S2}00$ γ γ γ γ γ γ
Document No:	A-429399-01-4 ~ ~ ~ ~ ~ ~
Guidelines: 📎 🏑	EC, Commission Directive 95/36/EC Amending Council
	Directive 94/414/EEC (Ampexes JA HII, Fate and Behavior in the Environment);
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	"Adsorption/Desorption", Jon. 21, 2000 🖉
	US EPA, Fate, Transport and Transformation Test Guidelines OPPTS 835.1220
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	and Soil Adsorption/Desorption Isotherm; not specified
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Title; 🔊 🔬	(14 G) RPA Ø	3328 Adsorptio	on / desorption in	n five soils	
Report No:	C000534				
Document No: 🖉	M-229091-01				
Guidelines: 🚿	ÊU (≂€EC) :≾	3/36/EC; OEC	D: 106; USEPA	A (=EPA): 163-1	
	US & PA OPP	TS 855.1230;se	e page 3		
GLP/GEP	yes a	8			



The <u>adsorption of metabolite RPA 202248 to soil</u> had been investigated in the laboratory in two studies using:

- four soils under standard conditions of batch equilibrium tests following application of prenyl distribution of the standard conditions of batch equilibrium tests following application of the standard conditions of batch equilibrium tests following application of the standard conditions of batch equilibrium tests following application of the standard conditions of batch equilibrium tests following application of the standard conditions of batch equilibrium tests following application of the standard conditions of batch equilibrium tests following application of the standard conditions of the sta
- four soils and a sediment from the US under standard conditions of batch equilibrium tests following application of phenyl-UL-¹⁴C- labeled test substance (KCA-7.1.3.1.2 /02)

The point was addressed under Point 7.1.2.2 of the Dossier submitted and evaluated within the process of Annex I inclusion as published in the corresponding Monograph. of RMS The Netberlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.

The evaluation during the Annex I inclusion process revealed that metabolite RPA 202248 was moderately adsorbed to soil. For the first study, $K_{F, adv}$ cranged from 54 to 134 mL/g while values for Freundlich coefficients 1/n were from 0.922 to 0.944. Following the DOR K from ranged from 32 to 79 mL/g.

For the second study and following the amendment to DAR the corresponding values for the adsorption $K_{F, ads, OC}$ ranged from 94 to 159 mL/g when excluding the sediment. Values for Freundlich coefficients 1/n were from 0.922 to 0.955 \sim

Re-calculation of metabolite RPA 202248 acorption to sop data

Because the residual water remaining in soil after centrifugation9 had not been considered adequately within the calculation of values of K_F , K_F , M_F , $M_$

New adsorption to soil data for metabolite RP@ 202248:

The existing adsorption data to soil were amended by new data in order to generate a robust data package describing the adsorption behavior of RPA 202248 in soil (KCA 7.1.3.1.2 /05).

The <u>adsorption of meriodite RPA 203328 to sold have been investigated under conditions of the</u> laboratory in two studies using: $\sqrt{203328}$ to sold $\sqrt{2003328}$ to sold $\sqrt{2$

- four soils under standard conditions of batch equilibrium tests following application of phenyl-UL-C-labeled test substance (KCA 7, 3.1,2,03).
- four soils and a segment from the US under standard conditions of batch equilibrium tests following application of phenyl-UL-¹⁴C tabeled test substance (KCA 7.1.3.1.2 /04).

The point was addressed under Point 7.52.2 of the Dossier submitted and evaluated within the process of Anne 21 inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

The evaluation finally resulted in invalidation of the existing adsorption data of metabolite RPA 203328.

Consequently there is no detailed description of this existing data in this update.

⁹ Term also used is 'interstitial water'.



A new study investigating the adsorption behaviour of RPA 203328 in five soils was therefore performed (KCA 7.1.3.1.2 /06).

Re-calculation of adsorption to soil data for metabolite RPA 202248:

It was found later that values of K_F , $K_{F,OC}$ and 1/n had not been calculated correctly due to nonconsideration of interstitial water remaining in the soil after centrifugation at each adsorption and $\int_{-\infty}^{+\infty}$ desorption step performed in the study. The data water thus re-evaluated with details of re-evaluation reported in the amended original reports under KCA 7.1.3.1.2 /01 and KCA 7.2.3.1.2 /02:

Report:	_; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;
	03; Amended: 2004-04-08
Title:	[14C]-RPA 202248: Addsorption/desorption to and from four soils - Addendum report
Report No:	CX/03/070A
Document No:	M-066355-03-1 Q 4 2 2 2 2 2 2 2 2 0
Guidelines:	91/414/EEC; 99/36/EC; Deviation not specified 2 2 2
GLP/GEP:	yes Q a a a a a a a a a a a a a a a a a a

The adsorption/desorption characteristics of [phenyl-UL-¹⁴C]RP@ 202248 to soil were determined for four soils under standard conditions of barch explibitium tests at a concentration range of two orders of magnitude (5, 1, 0.2 and 0.04 mg/L) at 20° S in the dark.

In the definitive test the overall mean recoveries ranged from 100.0 to 101.0% AR% for the sandy loam soil 93/22, 98.6 to 99.6% for loam soil 95/02, 97.9 to 101.1% for sandy loam soil 95/04 and 98.9 to 101.7% for the loamy sand soil 95/06 for all test concentrations and thus found to be in the acceptable range.

The revised Freundlich adsorption coefficients (K_F) ranged from 0.12 to 1.87 mL/g with corresponding values referenced to organic carbon (K_F, oc) to range from 11 to 99 mL/g (Table 7.1.3.1.2-4). Values for the Freundlich coefficient of adsorption 1/n ranged from 0.863 to 0.908.

Soil Soil		% Clay	pH (CaCl2)	CEC	Ads K _F (mL/g)	Ads Koc (mL/g)	Ads 1/n
Sandy loam, US(93/22)	1,5	\$5.2	5.5	5.7	0.35	27	0.863
Loam, UK (95/02)	≈ ≈1.9 ~C	20.0	4.3	6.3	1.87	99	0.907
Sandy Learn, UK (95/04)	2.2	8.8	4.7	5.7	1.65	75	0.908
Loamy sand K (95/06)	1.1	7.4	6.7	6.9	0.12	11	0.863

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CEC = Cation Exchange Capacity



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Title:	[14C]-RPA 202248: Adsorption/desorption to and from	n four soils an	d an aquatic sediment
Report No:	11486	~	
Document No:	M-066483-01-1	D D	
Guidelines:	EPA, Subdivision N, 163-1; Deviation not specified	O,	
GLP/GEP:	yes	A	

The adsorption/desorption characteristics of [phenyl-UL-¹⁴C]RPA 292248 to soil were determined for four soils and a sediment under standard conditions of batch equilibrium tests in a concentration range of two orders of magnitude (5, 1, 0.2 and 0.04 mg/L) at 20°C in the dark.

In the definitive test the overall mean recoveries ranged from 9305 to 101.1% AR% for the day soil 96/12, 99.1 to 100.2% for the sand soil 96/13, 982 to 99.8% (exclusion of one outlier) for barry and soil 96/14, 94.6 to 99.4% for silt loam soil 96/45 and 26.5 to 98.9% for the barry cand test concentrations and thus found to be in the acceptable range.

The revised Freundlich adsorption, coefficients (K_F) anged from 0.13 to 0.27 mL/g with corresponding values referenced to organic carbon $(K_{F,OC})$ to range from 22 to 62 mL/g when excluding values for the sediment (Table 7.12.1.2-29). Values for the Freundbich coefficient of adsorption 1/n ranged from 0.749 to 0.908, again after exclusion of values for the sediment.

1 abie 7.1.3.1.2-2. Supption	n penaviour ut t	XIGA 2022 40		polis allu	scument		
Soil Soil		%Clay	y y (wa€er)		(mL/g)	Ads Koc (mL/g)	Ads 1/n
Clay, US (2012)	U 1.2 Å	56	6.2 @	32.8	0.27	22	0.749
Sand, (% (96/13)	0.4	2.8	6.8	~6	0.13	35	0.873
Loamy sand, US (9644)	A 30.3	€.8	Ø.4	یر 1.8	0.21	62	0.908
Silt loam, US (96/15)	0.5	× *12.8	\$ 6.5 \$	10.1	0.19	41	0.862
Loam sediment, US (96/16)	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	12.8	<u>~</u> 56	12.6	5.92	119	0.874
CEC = Cation Exchange Ca	apacity 🛇 🖉	N Di	Ŵ				

Table 7.1.3.1.2-2: Socrition behaviour of RPA 202248 in four US soils and sediment

CEC = Cation Exchange Capacity C



Report:	b; ;2012;M-429399-01
Title:	[cyclopropyl-1-14C] AE 0540092: Adsorption/desorption in five different soils
Report No:	AS200
Document No:	M-429399-01-1
Guidelines:	EC, Commission Directive 95/36/EC Amending Council
	Directive 91/414/EEC (Annexes II + III, Fate and Behavior in the Engironment);
	July 14, 1995
	OECD Guideline for Testing of Chemicals, No 106
	"Adsorption/Desorption", Jan. 21, 2000 $(1 + 2)^{\gamma}$
	US EPA, Fate, Transport and Transformation Test Guidelines OPP18 835, 0220
	Sediment; not specified
GLP/GEP:	yes Q A A A A A A A A A A A A A A A A A A

Executive Summary

The adsorption/desorption characteristics of eyclopropyl-1 AE 0540092 were determined for five soils (three EU, two US) in a concentration range of two orders of magnitude.

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In the definitive test the overall mean recoveries were 94.5% AR ± 94.8% 5%#or 94.6% AR $\pm 0.8\%$ for soil 99.0% ARO \pm 0.7% for soil $AR \pm 0.6\%$ for soil and for all test concentrations and thus found to and 96.3% AR + 0.6% for soil be in the acceptable range?

The Freundlich adserption coefficients (K) ranged from 0.1016 to 0.6162 mL/g with corresponding values referenced to organic carbon $(K_{F,OC})$ to range from 1456 to 20.8 mJ g. Values for the Freundlich coefficient of a Borption 1/n ranged from 0.09995 to 0.9999

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Aaterial and Methods

A. Materials

- CIRPA 202248 (AE 0,200092) cyclopropy 1. Test Materia
 - Specific radioactivity: 408 MBd/mg (210.26 µCi/mg)
 - Radiochemical purity > 98% (HPIC)

2. Soff:

arbon content and texture. The characteristics and the US are sommarised in Table 7.1.3.1.2-3. Sorption tests were performed with five soils covering a range of pH, organic ∧ carbon content and texture. The characteristics of soils originating from Germany

Table 7.1.3.1.2-3: Ch	naracteristics of to	est soils			
Soil			•	*	•0
	(WuW)	(HaH)			
Geographic Location (City/State/Country)	Germany	Germany	Çêrmany	California a	Nebraska
GPS coordinates					
Pesticide use history	None within th	e last four years pr	iger to sampling	Note within the last two years prior to sampling d	None within the Last four years • prior of samping
Collection	Field	satepled with sho	vel/spade into a da	ited plasic bags (b	uçkets)
procedures		S V S			Ú r
Sampling depth		🖉 0-220 cm 🃎		≫ 0–18°cm √	20-20 cm
Storage conditions		Ô Ô	$O = B C^{C}$		°∕γ
Storage duration	102 days	102 days ^C	102 days Q	806 days	& 806 days ^C
Soil preparation	Ŵ.	Air-c	lried and sieved to	2 m	0
Soil Taxonomic Classification	N/A O	N/A U			N/A
Soil Series ^D	N/A ×	N/A	N/A O	N/A	N/A
Textural Class (USDA)	sandy loam	Suit loam	no de loagen	sandy loam	silt loam
Sand (%) A	≈ 57 *		37 🔊	Sec.0	12.7
Silt (%) ^A	S 30 (<u></u> 40	32.6	60.8
$\operatorname{Clay}(\%)^{A}$	<u>° 13 6</u>	<u> </u>	<u>~ 23 ×</u>	v 11.4	26.5
pH (0.01 M CaCl ₂)	5.1	6.3 [°]	, <u>"</u> .3 ~	6.7	6.6
pH (Water)	5.405	<u>6.5</u>	≪j∛7.5, O°	6.8	7.2
Org. Matter ^B (%)	<u>} 3,</u> 44 <u>v</u>	Å .99	& 7.5 <u>Å</u> [®]	1.1	2.9
Org. Carbon (%)	ž <u>2</u> .0 ^(*)	<u>`</u> ? 2.90°	0 4.4	0.7	1.7
CEC (meq/100 g) 🧐	10.3	0° 129 0	£ 9.2	16.1	16.1

^A) According to USDA classification;[®] testing facility. testing facility.* CEC: Cation exchange capacity; No

B. Study design Study were weighed each into centrifuge tubes to which ca. 18 mL of 0.01 M aqueous calcium chloride was added. The resulting slure was pre-equilibrated for at least 16 hours followed by the addition of 2 mL of the corresponding application solution to result in a final volume of 20 mL and a soil/solution ratio of 1:1(1:2). Within pre-tests controls containing no soil were prepared for determination of stability of the test item in calcium coloride solution and adsorption tests to the walls of the test vessels.

Initial nominal concentrations of the ¹⁴C-test substance in the aqueous phase were 1, 0.3, 0.1, 0.03 and 0.01 mg/L thus covering two orders of magnitude.

¹⁰ 20 g of soil ; 10 g of soil II. **Bayer CropScience**

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No organic co-solvent was present in samples during the tests. An initial pH of 6.7 was determined for the aqueous 0.01 M calcium chloride solution in the absence of soil. Each determination was performed in duplicate by shaking with an overhead shaker in the dark at 20±1°C. Preliminary tests included the determination of the adequate soil-to-solution ratio (24 hours and the corresponding adequate equilibration times needed for adsorption (total shaking time of 120 hours) for all soils.

For the definitive test an adsorption step of 24 hours was followed by a single desorption step of 24 hours at all test concentrations. The desorption step was carried out by replacing the aqueous supernatant after adsorption by fresh aqueous 0.01 Maalcium chloride solution for each sample? For work-up the aqueous supernatant was separated from soil by secantation and centrifugation (10 min, 5000 rpm). Radioactivity in water and soil extract@was_determined by liquid scintillation counting (LSC). Non-extractable radioactivition soilowas determined by combustion followed by LSC to establish a full material balance. C AND O

Finally the adsorption parameters were calculated using the Foundlich adsorption sotherm.

2. Analytical procedures: Radioactivity was determined by kight scintillation counting (CSC). The purity and stability of the test item was investigated boreversed phase radio HPL Canalysis using 14Cflow-through detection techniques.

II Results and Discussion

A. Mass balance and results of preliminary tests

Solubility and stability tests in aqueous 0.01 M calcium chloride solution confirmed stability under the conditions of the test Pre-tests on adsorption to the walls of test versels by shaking an aqueous solution of the test substance in the absence of sold showed no adsorption.

0

Pre-tests or adsorption equilibrium to soil resulted in mase balances ranging from 96.7 to 97.9% of AR for a soils after 120 hours (Table 7.1.3, 1.2-4). In parallel, the stability of the test substance was confirmed by parental mass balances of nore than 90% of ARS

For the definitive tests the overall mass balance for individual samples for all concentrations and all JA.5% Foil soils investigated ranged from 91.9 to 100.3% of AR. The resulting mean recoveries (Table 7.1.3.1.2-) to 99.0% (soil) with standard) to $\pm 1.5\%$ AR (soil and

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Isoxaflutole						

Table 7.1.3.1.2-4: Prelimi determinat	nary test: Reco ion of adsorption	overy of [cyclo] equilibrium afte	propyl-1- ¹⁴ C]RPA er 120 hours.	202248 in	samples from
Soil			II		

Supernatant*	48.9	35.4	48.1	52.9	
Soil extract*	41.6	55.6 (3 42.5 √	43.4 🔊	~46.4 ~~
Non-extractable residues in soil, if measured	5.5	6.3	5.7 5.7	1.50	3 6.8 J
Total recovery of test item	90.5	91.0	900	96.3	\$ 97 .6
Total recovered radioactivity	97.9	\$\$ ⁵	~96.7 °	⁹ 97.8	\$ 97.4 °
* Test item was shown to be sta	able by HPLC an	alvere 9			\sim

Table 7.1.3.1.2-5: Definitive test	: Total	recovery of	[cyclopropy0-1-	¹⁴ C(RPA	↓ 292248 ≸ ø	samples	following
adsorption and	lesorpt	ion phase of 4	8 hours in total			Å.	Ő

Soil					Ĵ.
Test concentration (mg/L)					¥ _
1.01	.^\$ 96.5♥	r 95.8°	Ø4:8	98.2	95.7
0.30	× 950 C	j \$5.1 m	∜95.2 °√°	98,9	95.6
0.10	94.1 ×	94.5	<u>93</u>	98,8	96.3
0.03	A 91.9	<u>~</u> 94.0	🖓 <u>9</u> 5.6 🔧	298.9	96.6
0.01	\$ 95 ₀ 0	9405 ~	j ∂ 3 .9 ‰	مَحْ 100.3	97.3
Mean recovery	94 ?5	/ JA.8 J	94.6 ^O	≫ 99.0	96.3
SD SD	°,₽1.5 «♪	± 0.69		t = 0.7	± 0.6

Values derived from mean values of duplicate samples in terms of percentages of AR. SD = standard deviation

B. Transformation of parent compound: The stability of the sest substance was confirmed by HPLC analysis to result n patental mass balances of more than 90% of AR.

C. Findings: The Offinitive tests were performed at a ratio of 1:2 for soils and . Pre-tests showed that the II and a ratio of 1:1-for soils and

adsorption equilibrium was ceached after 24 hours. The definitive tests were therefore performed for an adsorption time of 24 hours to lowed by a desorption step of 24 hours.

The adsomption behaviour of [Syclopforpyl-12+C]RPA 202248 could be accurately described within a nominal concentration range of 0.01 mg/L to 9.0 mg/L by the Freundlich equation for all soils (Table 7.1.3.1.2-6). The appropriate constants $K_{r(ads)}$ of the Freundlich isotherms ranged from 0.1016 to 0.6162 mL/g with associated Freundlich exponents 1/n to be below 1 for all soils (0.8586 to 0.9185). To some extent the desorption behaviour to soil was thus affected by the concentration of the test item. The corresponding correlation coefficients of adsorption isotherms ranged from 0.9995 to 0.9999 therefore indicating a good linear fit to the measured data. When being normalized for organic carbon content of sour values of adsorption K_{oc} varied from 11.6 mL/g (soil) to 30.8 mL/g in maximum (soil

Desorption constants K _{F, des} according to Freundlich ranged from 0.1149 mL/g (soil) to 0.9194 mL/g (soil). The corresponding values for $K_{OC, des}$ ranged from 14.5 mL/g (soil II) to 46.0 mL/g (soil



Once adsorbed to soil particles values of RPA 202248 for desorption K_{OC} were slightly higher than the corresponding values of adsorption K_{OC} thus indicating some strengthening of binding.

								6			
	Seil			Adsorpti	on		🔬 Desorption 🖉			S a	D
Soil	type	pH*	K _F (mL/g)	1/n	R ²	Koc**	K _F (mL/g)	, ^v 1/n	\mathcal{R}^2	Koc*** (mlg)	, Q
	sandy loam	5.1	0.6162	0.8999	له 0.9998 ©	y ∦ 30.8	0,0194	0.8826	0.9986	6.0 ×	Ď
	silt loam	6.3	0.3709	0.8643	0.9997	12.8	0.426	0.8918	J.9993	14	
II	loam	7.3	0.5331	0.8586	\$0.9995 0	° 12,40°	0,6375*	0.8239	0.9989*	¥4.5*	
СА	sandy loam	6.7	0.1016	0.9185,	0,9999	J14.5	0.1149	009100	0.998	164	
NE	silt loam	6.6	0.1971	9.9157¢	0.9998 0.9998	176	0.2535	0.8900	00997	Õ 14.9	

Table 7.1.3.1.2-6: Adsorption and desorption constants of RPA 202248 (AE 0540092) in soil

Values in terms of Freundlich $K_{C, ads}$ were moderate for metabolite RPA 202248 to range from 11.6 to 30.8 mL/g for the five only tested

The desorption constants according to Freundlich (KSC, des) were sughtly higher than the respective

energiest concentration energiest concentrati

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Overall summary on adsorption of RPA 202248 in soil:

Investigations into the adsorption of RPA 202248 in soil were performed in three studies in total covering a whole range of soils as detailed in studies KCA 7.1.3.1.2 /01, KCA 7.1.3.1.2 /02, and KCA 7.1.3.1.2 /05. The adsorption data from 13 soils as compiled in Table 7.1.3.1.2-7 esulted in an arithmetric mean $K_{F,OC}$ value of 34.9 mL/g and a mean value of 0.876 for 1/n for use we environmental risk assessments.

,Ŵ

1 abic 7.1.5.1.2-7. 1 abit ption b	chaviour o		Jo m son				
Soil	%OC	% Ctay	BH	CEC	AdoKr	ads Koç	Ads
501		0	(CaCl2)	1 D	OmL/g)	(m/ g)	A1/n
Sandy loam, US (93/22)	1.3	5.2	5.3 ,5.3	5.7 Å	00	ی پر 27 ^{کر}	0.263
Loam, UK (95/02)	1.9	20.0	√ 4.3 ×	× 43	×1.87 0	2 9 9	Ø.907
Sandy loam, UK (95/04)	2.2	₩ 10 ⁻ 8.8 ¹	4.7	5.7 8	1.65	£75 4	0.908
Loamy sand, UK (95/06)		7.4	. 7	6.9	29.12		0.863
Clay, US (96/12)	[™] 1,22 [™]	×\$6.8	6.2	\$2.8	© 0.27	۵	0.749
Sand, US (96/13)	Ø38	2.8	6.8	1.58	A)3	J 35	0.873
Loamy sand, US (96/14)	0.346	68 4	6.4	% 84	0.21 0	62	0.908
Silt loam, US (96/15)	2 097	Q 12.8			0.19	41	0.862
Sandy loam	×2.0	137	\$5.1	103	~ 0 %6162	30.8	0.8999
Silt loam	2.0	& 13 ×	6.0	\$2.9	0.3709	12.8	0.8643
Loam II 🖉	4.4 A	22	Ø.3 0	19.4	0.5331	12.1	0.8586
Sandy	0.7	<u>1</u> 1.4		06.1			
CA S A		Ô [¥] v ^Y	6.7	S ^Y	0.1016	14.5	0.9185
Silt loam	\$ 1.7 °	~26.5	6.6	16.1	0.1971	11.6	0.9157
Mean (average 2 2		\sim \sim)' '			34.9	0.876

r	Table.	712177.	A	hab and area	af DDA	202240:		
	гаріе	/	Adsorption	Denaviour	OT KPA	202/2/401	n :	SOI
			rasorpron					

CEC = Cation Exchange Capacity (mval/100 g soil)

pH: Value according to 0.01 M cateium coorde method

K_F: Freuhallich coefficients of adsorption (**) 1/n: Stope of the Freuhallich adsorption isotherms K_{oc}: Adsorption coefficient per organic carbon (K \$100/% organic carbon)



Report:	ਤ; ;; ;2004;M-229091-0	1
Title:	(14C)-RPA 203328: Adsorption / desorption in five soils	Q d
Report No:	C040534	
Document No:	M-229091-01-1	
Guidelines:	EU (=EEC): 95/36/EC; OECD: 106; USEPA (=EPA): 163	
	US EPA OPPTS 835.1230;see page 3	
GLP/GEP:	ves	N N N

Executive Summary

-O V The adsorption/desorption characteristics of [phenyl-UL-¹⁴C]ROA 203328 (AE determined for five soils in a concentration range of two orders of magnitude. In the definitive test the overall mean values of recoveries for all concentrations were 1023% of $R \pm 0.9\%$ for soik $105.4\% \pm 2.3\%$ for soil , 10 .3% # 0.5% for soil $00.6\% \pm 0.8\%$ and $97.0\% \pm 0.7\%$ for soil thus indicating quantitative recoveries for soil Values for the coefficients of adsorption according to Freundlich (K, Cads) ranged from 0.00881 to 0.0283 mL/g for the three soils evaluable. The convesponding and Values for the Freundlich values related to organic carbon, (Kocxads) ranged from 0.88 to 1.49 mL/g coefficient of adsorption 1/n for the three soils ranged from 0.97 Sto Adsorption of RPA 203328 to soil was therefore A. Materials I. Material and Methods 1. Test Material: [pherod-UL **C]RP 203328 (AE B197855) Ŕ Specific regioactivity: 2.86 MBg/mg (269.3 MBg/mmol) -289 -289 -289 -289 -289 -289 -289 -289 -289 -289 -289 -289 -280 -2 -280 --Radiockemical purity 99.8% (HPLC) Sorption tests were performed with five soils covering a range of pH, organic carbon content and texture. The characteristics of soils originating from the UK

Table 7.1.3.1.2-8: Cl	naracteristics of to	est soils			Le D
Soil	(03/18)	(03/20)	(03/23)	(03/29)	(63/35). (63/35).
Geographic					×, ×,
Location	Field Station	Farm,	Grounds Farm,		
(City/State/Country)	, , UK	, UK	, UK		Germany C
Grid reference		4.			
Soil preparation		(Dartly)	Rir-dried and sieve	ed to Ormm 🔊	f. A
Textural Class	sandy loam	clay loam 🍙	sandy cla	Sandy Joam	silo loam
(USDA)	-			A Ö ^v «	
Sand (%) A	72.4	⁴³ .2√ ⁹	Q ⁷ 49,7 0	68.3	لا 28 الم
Silt (%) ^A	11.0	Q 23.5	J4.1 X	25.3	61.1
Clay (%) ^A	16.6	Q* 23.3 ×~	≪36.0	<u>, ~ 6,4</u> , <u>,</u>	Ø0.3
pH (0.01 M CaCl ₂)	7.6 Q	õ 7.3 õ	~ 5,90° (\$ 5.6
pH (Water)	7.7	× 7.6°	S SG S	°7.1 °0	6.5
pH (1 M KCl)	7.2~	× 7.1 "	<u>~~6.3</u>	6.3	O [♥] 5.6
Org. Matter ^B (%)	2.6% &		<u></u>	γ 10	2.8
Org. Carbon (%)	\$Q.5 O	~ <u>1.9</u>	3.3	<u> </u>	1.6
CEC (meq/100 g)	39.4	<u>172</u>	× 47.8	× 8.3	9.6
Moisture content (%)	14.7 0	8.54 O	\$19.49°		19.91
Max. WHC @	A3.9 5	£ 44,7 s		@27.5	60.8
$\frac{1}{1000} \frac{1}{1000} \frac{1}{1000} \frac{1}{1000} \frac{1}{10000} \frac{1}{10000} \frac{1}{10000000000000000000000000000000000$	20.7 /		× 44 0 \$	17.6	28.5
0.33 bar $(\%)^{C}_{C}$			~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	©	20.5
CaCO ₃ -equivalents	ي 72.5 جي ^ع	3.80		1.0	1.0
Phosphorous, total (mg/kg)	749 P 5		608.35	855.5	664.8
Nitrogen, total $\sqrt[6]{(g/kg)}$	A 567.8	× 1722,0 × ×	2692.2	826.0	1148.1

^{A)} According to SDA classification: ^{By}% Organic matter =% organic carbon x 1.72; ^{C)} Max. WHC = maximum water holding capacity; CEC Cation exchange capacity

B. Study design

1. Experimental conditions: For the definitive test samples of 40 g dry weight of soil (2 mm) were weighed each into centrifuge tubes to which ca. 40 mL of 0.01 M aqueous calcium chloride was added. The resulting slurry was pre-equilibrated for at least 12 hours followed by the addition of 1 mL of the corresponding application solution to result in a final aqueous volume of 40 mL and a soil/solution ratio of 4.1. Being part of pre-tests control samples containing no soil were prepared the same way for determination of stability of the test item in calcium chloride solution and for testing of adsorption to the walls of the test vessels. The initial nominal concentrations of the ¹⁴C-test substance in the aqueous phase were 1, 0.3, 0.1, 0.03 and 0.01 mg/L thus covering two orders of magnitude.



No organic co-solvent was present in samples during the tests. An initial pH of 6.3 to 7.2 was determined for the aqueous 0.01 M calcium chloride solution in the absence of soil. Each determination was performed in duplicate by shaking with an overhead shaker in the dark at 20 \pm 1°C. Preliminary tests for all soils included the determination of the adequate soil-to-solution ratio (24 hours) and the corresponding adequate equilibration times needed for adsorption (total shaking time of 168 hours).

For the definitive test the adsorption time of 24 hours for **solution** foil and 168 hours for the other soils was followed by a three desorption steps of 24 hours each at all test concentrations. The desorption step was carried out by replacing the agreeous supernation (10 mL for **solution** soil) after adsorption by fresh aqueous 0.01 M calcium filoride solution for each sample. The For work-up the aqueous supernatant was separated from soil by decartation and centrifugation (10 min, 5000 rpm). Radioactivity in water and soft extracts was determined by liquid scintillation counting (LSC). Non-extractable radioactivity in soil was determined by combustion followed by **b**SC to establish a full material balance.

2. Analytical procedures: Radioactivity was determined by fiquid scintillation counting (LSC). The purity and stability of the test item was investigated by reversed phase radio-HPOC analysis using ¹⁴C-flow-through detection techniques

2 ST 1. Results and Discussion

B. Mass balance and results of preliminary tests

Ŵ

Solubility and subility tests in aqueous 0.01 M calcium chloride colution confirmed stability under the conditions of the test. Pre-fests on adsorption to the walls of test wessels by shaking an aqueous solution of the test substance in the absence of soil showed no adsorption. Pre-tests showed also the stability of the test substance of der the conditions of the test.

For the definitive tests the overall mass balance for individual samples at all concentrations and for all soils investigated ranged from 95.9 to 108.0% of AR. These values resulted from exclusion of two values of single replicates of the 0.00 and 003 mg/L test soncentrations each in the soil. The resulting mean recoveries (Table 7.1.3.1.2.9) were found to range from 97.0% for soil 105.4% for soil associated with standard deviations ranging from \pm 0.5% AR for soil to $\pm 2.3\%$ AR for soil

Table 7.1.3.1.2-97 Definitive test: Total recovery of [phenyl-UL-¹⁴C]RPA 203328 in samples following an Adsorption place of 168 hours* and three desorption steps of 24 hours each in total

Soil Test corcentration (mg/L)					
1.00	100.8	101.6	101.3	100.7	98.2
040 2	101.9	104.4	101.5	100.9	97.2
0.10	102.8	105.8	100.3	99.1	96.7
0.03	102.6	107.3**	101.5	100.5	97.1
0.01	103.4	108.0**	101.7	101.6	95.9
Mean recovery	102.3	105.4	101.3	100.6	97.0



Soil					s s s s s s s s s s s s s s s s s s s
Test concentration (mg/L)					
SD	± 0.9	± 2.3	± 0.5	≈ 0.8	£0.7
Mean values of duplicate sample	es in terms of per	centages of AR		S.	4 2
* Adsorption phase was 24 hour	rs for	soil		«O"	
** Values for single replicates of	lue to mass balan	ces out of accepta	able range for other	Feplicate	
				×,	S & O
B. Transformation of pa	rent compour	d: The stabilit	ty of the test su	bstance was o	apfirmed by
HPLC analysis in pre-tests a	and in the definition	itive test.	Å	, o ~ ~	
C Findings. The definitive	a tasts wara nar	formed at a rat	tio of 1:1 formal	soil Dro Bot	showed that
the adaption aquilibrium		form 24 hours			
the adsorption equilibrium	was reached a	iner 24 nours		sen where it's	ad not been
reached for the other soils.	The definitive	tests were the	retore performed	i for an adsor	ption time of
24 hours for soil	and for 168 h	hours for the of	her. Each adsorp	tion phase was	s torlowed by
three successive desorption	steps each for 2	24 hours.		y Q L	
Desorption values were infl	uenced by the l	ow overall ads	erption and thus	high values of	radioactivity
remaining in the interstitial	water after the	adsorption ste	p. The values pr	esented in the	study report
were therefore not summa	rized in more	detail in this	document since	The values c	aculated are
regarded to have no scientif	ic calidity.				
As a consequence of the po	or adserption	the adsorption	behaviour off 14C	-RPA 203328	could not be
accurately described by	the Freundach	n equation for	r solls	and	
(Table 7.1.3.1.2-10). For th	e remaining so		Â,	and th	ne adsorption
constants K _{F(ads)} from Fre	undligh isothe	rms vanged f	rom ₂ , 0.00881 to	0.0283 mL	g (mean of
0.0185 mL/g) with assocrat	ed Freundlich	exponents $2/n$ s	ignificantly beto	w 1 (0.569 to	0.853) and a
mean of 0.6490The desorp	tion behaviour	To soil was the	us affected by th	e concentratio	on of the test
item.					
The corresponding correlati	on coefficients	of adsorption it	sotherms ranged	from 0.9978 to	o 0.9998 thus
indicating a good linear fit	the measure	ed data. Norm	alisation for orga	anic carbon co	ontent of soil
resulted in values of adsorpt	tion $\mathcal{K}_{\mathrm{OC}}$ rangin	g from 0.88 (sc	pil 1) to 1.49 mL/g	in maximum
(soil) with an arithmet	tic mean of 1.1	im L/g	ý.		

No desorption constants $\mathcal{C}_{F, d}$ are presented in this document since the overall low adsorption prevented the determination of deliable data.

		a 🔪 (N N	Adsocotio	on		1 st Desorption			
Soil	type	рН	″ K ⊮ (₩0]L/g) ^	o¶∕n	R ²	Koc (mL/g)	K _F (mL/g)	1/n	R ²	Koc*** (mL/g)
	sa a ay toam	7.6	* * *	*	0.130	*	0.0159	0.692	0.995	1.06
	Clay oliver		0.0283	0.569	0.9998	1.49	0.0840	0.613	1.000	4.42
	sandy clay	5.9	*	*	0.585	*	0.0759	0.747	0.976	2.30
	sandy loam	6.3	0.00881	0.853	0.978	0.88	0.0523	0.944	0.998	5.23

Table 7.1.3.1.2-10: Adsorption and desorption constants of RPA 203328 (AE B197555) in soil

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	-:14									
	silt loam	5.6	0.0183	0.649	0.994	1.14	0.0460	0.594	0.997	2.87
pH: Value giv K_F : Freundlich 1/n : Slope of K_{oc} : Adsorptic R^2 : Regression * No use of re ** Mean of va	en as det n coeffici the Freu on coeffici n coeffici sults due ilid soils	termined ients of ac ndlich ad cient per ient of Fr to too lo	with 0.01 M dsorption (' sorption/de organic car eundlich eq w correlation	A calcium **) and d sorption : bon (K x juation on	n chloride s esorption (isotherms 100/% org	solution (***), respe ganic carbon	ectively n)	R. A.		
				III.	Conclusi	ion 🔍	S B.	Å	Å,	o _d u'
Values of adsorption in terms of $K_{F, OC}$ were therefore low for metabolite RPA 203328 to range from 0.88 to 1.44 mL/g. With virtually no to very low adsorption observed no scientifically meaningful value, could be determined for the corresponding desorption constants $K_{F, OC}$										
CA 7132	Δge	d sornt	ion 🗳	б ^у	`~``` &```?``					
Ageing of isoxaflutole residues in sol and the use of desorption parameters were not considered in current risk assessments.										
CA 7.1.4.1.	1 [°] Coh	imn lea) Iching of	the act	ive subs	tance	L.			
Column leac) hing stu	dieswit	whe activ	e substa	nce isoxat	flatole we	e not perf	formed.		
This data requirement had been addressed ander Point 7.1.3.1 4 of the existing Dossier and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.										
The evaluation	on reve	aled that	instead of	of perfor	aning are	olumn lea	ching stu	dy, the 1	mobility i	in soil is
assessed by	lata on t	heir pers	sistence (¢	g. half-	lives, und	ler aerobic	condition	is in the	laboratory	y and, by
the adsorption	on to so	il./These	e data allo	ow for a	n adequa	te descrip	tion of th	e behavi	iour of th	e parent
compound in	soil in	environ	nentærisk	assessn	ents.					
A column lea	aching s	tudy wit	h sarent c	ompound ¢ metabo	d is theref olites, br	fore regard eakdowr	led as not a and rea	necessar	y. roducts	
Column leag	ano sta	dies with	n soil met	abolites	of isoxafl	utole were	not nerfo	rmed		

tabolites of isoxaflutole wer

Column leading studies with soil metabolites of isoxaflutole were not performed. This data requirement had been addressed under Point 7.1.3.1.2 of the existing Dossier and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.


The evaluation revealed that instead of performing a column leaching study, the mobility of metabolites RPA 202248 and RPA 203328 in soil can be adequately assessed by data on their persistence (e.g. half-lives) under aerobic conditions and the adsorption to soil. These data allow for a vale vale description of the mobility of soil-born residues in environmental risk assessments. Column leaching studies with metabolites are therefore regarded as not necessary.

CA 7.1.4.2 Lysimeter studies

Lysimeter studies with the active substance isoxaflttple were not performed. This data requirement had been addressed under Point 7.1.3 2 of the existing Dossier and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

The evaluation revealed that the potential for mobility of iso aflutors residues to ground water can be adequately assessed by the simulation of vulnerable scenarios representative for the EU. The simulations are able to cover a range of worse vases rather than to be limited to soil and climatic conditions reflected by lysimeter studies

Separate lysimeter studies with isoscaflutole are therefore regarded as not neces

Field leaching studies CA 7.1.4.3

Field leaching studies with the setive substance isoxaflutole were bot performed. This data requirement had been addressed under Point 0.1.3.3 of the existing Dossier and evaluated within the process of Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

The evaluation revealed that the potentia for mobility of isoxaflutole residues to ground water can be adequately assessed by the simulation of vulnerable scenarios representative for the EU. The simulations arg able & cover a range of vorse eases rather than to be limited to soil and climatic Separate field leaching studies with isoxaftitole and therefore regarded as not necessary. conditions reflected by lysumeter studie

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

- CA 7.2 Fate and behaviour in water and sediment
- Route and rate of degradation in aquatic CA 7.2.1 photochemical degradation)
- Hydrolytic degradation CA 7.2.1.1

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CA 7.2	Fate and behaviour in water and sediment
CA 7 2 1	Doute and note of domination and is a straight on the straight of the
CA /.2.1	Koule and rate of degradation in aquatic systems (chemical and
	pnotocnemical degradation is a set of the se
CA 7.2.1.1	Hydrolytic degradation of a star star star star star star star st
Penort:	v:
Title:	Hydrolysis 14C-R $@A20k772$
Report No:	R002384 0 4 5 7 6 6 6
Document No(s)): Report includes Trial Nos.:
	M-16255 01-1 V V V V V
Guidelines:	USEPA (=EPA): N, 161-1; Deviation not specified
GLP/GEP:	$\underbrace{\operatorname{yes}}_{\mathcal{O}} \xrightarrow{\mathcal{O}^{\vee}}_{\mathcal{O}^{\vee}} \underbrace{\mathcal{O}^{\vee}}_{\mathcal{O}^{\vee}} \mathcal$

The hydrolysis of isosaflutol was investigated in

sterile aqueous buffer solution at pH 5, Fand 2 after application of phenyl-UL-14C-labeled active C AND substance and incebationat 25°C (KCAP7.2. 54 /01) ų. K

The data requirement was addressed under Points 2.9.1 and 7.21.1 of the Dossier submitted and evaluated within the process for Annex Linclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.

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The evaluation revealed that the hydrolytic behavior & isoxaflutole is well understood. Therefore, no additional hydrolysis studies are necessary. The had-lives of isoxaflutole under conditions of sterile aqueous buffer hydrolysis have been summarized in Table 7.2.1.1-1.

Hydrolysis was shown to be dependent on pHoesulting in half-lives of 11.1 days at pH 5 to decrease to values of 20.1 hours (pH 7) and 3.2 hours (pH 9) at 25°C.

Table 7.2.1.1 Half-Mives of soxaflutole in sterile aqueous buffer at 25°C

	pH	Half-life
	Ö 5	11.1 days
	7	0.84 days (20.1 hours)
Ô	9	0.13 days (3.2 hours)



As a result of a pH-dependent rearrangement, the diketonitrile compound RPA 202248 was formed as the sole product of hydrolytic transformation and observed at maximum values of 58.3% of Ar at 2 pH 5, 53.9% at pH 7 and 95.5% at pH 9. Since no further transformation products were observed, RPA 202248 was shown to be stable under conditions of hydrolysis in sterile achieves buffer solution

The proposed pathway of hydrolytic degradation of isoxaflutole in sterile aqueous Č) V summarized in Figure 7.2.1.1-1.

Figure 7.2.1.1-1: Proposed hydrolysis pathway of isoxa@utole in sterile aqueous buffe





Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Report	······································	
Title:	Isoxaflutole - Quantum vield and environmental half-life in water	n n n n n n n n n n n n n n n n n n n
Report No:	R014778	
Document No(s):	Report includes Trial Nos.:	
	M-189753-01-1	×
Guidelines:	BBA: Part IV. 6-1. First level. (1990) : ECETOC: Technical Report	8.12. (1994):
	UBA: (1990); Deviation not specified	
GLP/GEP:	yes Ar O C	
Report:	9; ; ; ; ; ; 2002; M-214288-01 ~ ~ ~	\swarrow
Title:	Photolysis in distilled water RPA202248	ÿ
Report No:		
Document No(s):	Report for the first field of the field of	
Guidelines:	Deviation nor specified a diversity of the diversity of t	
GLP/GEP:	Y no L B A V V D V	

The direct photolysis of isoxaflutole was investigated in

• sterile aqueous buffer at \$H 5 following application of phonyl-UL-14C-labeled active substance and irradiation at 55°C at 50°North labrude light intensity in July (KCA 7.2.1.2 /01).

The point was addressed under Point 2.9.2 and 7.2.1.2 of the Dossier submitted and evaluated within the process for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.

The evaluation revealed that photolytic degradation of isoxaflutole was moderate to result in a photolytic half the of 6.7 days when being referenced to natural sunlight at summer solstice and latitudes of 20 to 50 N.

The active substance was degraded from 98.5% AR at start to 39.3% at the end of irradiation while degradation in eark controls was insignificant (88.5% AR) following the same incubation time. Photodegradation products were extensively formed to result in at least 18 minor components each observed at levels below 5% AR in the course of the study.

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Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Two components, reported as Met 14¹¹ and Met 20, were observed at maximum values of 9.3% (Met 14) and 16.8% AR (Met 20) thus approaching or exceeding the trigger of 10%. Structure elucidation resulted in the two compounds included in the proposed photolysis pathway of isoxal utole in sterile aqueous buffer as summarized in Figure 7.2.1.2-1.

It should be noted that the formation of the two structures Met 14 and Met 20 resulted from opening of the two ring systems, the isoxazole ring and the cyclopropane ring. In offect, these structures are closely related to the major soil and aquatic metabolite RPA 202248 rather than to the parent molecule.

Efforts for synthesis of Met 14 and Met 20 failed, however the high stocctural similarity RPA 202248 justifies that any potential effects to aquatic organisms are related to RFA 202248 rather than to the parent compound.

Photolytic processes would contribute to the elimination of isoxetlutole from the aquatic experiment to some extent when just concluding on the results of the sterile aqueous buffer photolysis study In total, direct photolysis processes to not contribute significantly to the overall climination of isoxaflutole from the aquatic environment. This conclusion comes when considering additional fast and significant pathways of degradation in the equatic environment;

The contribution of abiotic hydrolysis (KCA 7.2 1/01) is already significantly higher (DT₅₀ of 0.84 days in water at pH 7) than the potential portion coming from photological processes (DT50 of 6.7 days) at the same pH. This rough comparison did not consider the contribution of biotical-induced hydrolysis as documented by a DT₅₀ of 0.6 days if non-oerile natural water (CA 7.2.2.2 /01) or a DT₅₀ of 0.34 to 0.53 days in Water of total sediment/water systems (KGA 7.2.2.3 /01).

These general considerations may be taken not account when assessing the probability of occurrence and potential effects of metabolites Met 14 and Met 20 since these compounds do result from direct transformation of the active substance isoxafpuole offy. Other metabolites as those originating from biotic processes (i.e. RPA 202248, RPA 203328 and RPA 205834) were observed each below 3% AR and thus to a minor extent in the course of the study.

The overall conclusion is that it is extremely whitely to observe any direct photo-degradation product in the outdoor aquatic environment of ginating from the active substance isoxaflutole directly. Figure 7, 41.2-1: Proposed photolysis pathway of iso's aflutole in sterile aqueous buffer





Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole



The quantum yield of isoxaflutole was determined by:

irradiation of non-labelled active substance in aqueous solution at 25°C for 600 seconds (ECEPOC method), paralleled by the determination of the photon concentration by actinometry (KCA 7.25, 2 /02).

The point was addressed under Points 2.9.2 and 7.2.1.2 of the Dossier submitted and evaluated within the process for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update.



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The evaluation revealed a quantum yield of 7.83 x 10^{-4} . This value was used for the calculation of or environmental half-lives of isoxaflutole in water according to the method of Frank and Kloepfter. The resulting values are summarized in Table 7.2.1.2-01.

Kloepffer		(Cas	s, '	
Month	(b)	The Hali Gars)		(days)
January		9082		Å 378 Å
February	49	3922		163 ~~~
March	S.	19268 🔊 🎽		°78.7 [≪]
April	O K	983 6	NO ON	£ 41.0 £ °
May	A. O	769 4		\bigcirc 32
June		673 0		28.0
July		919 S		30.0
August		702 8		@ ^v 29.3
September		1190		× 4%6
October		2925 5 1		98.9
November		\$532 Q		\$272
December		8466		[©] 769
			5 ~ ~ ~ <i>2</i>	

Table 7.2.1.2-1: Environmental half-lives of isoxaflutol	e in water	according the	methodof	Frank	and
Kloepffer	>.	, ^{so} '	~~	° N	Ň

In addition, the behavior of metabolite RPA 202248 was prossigned under conditions of photolysis in distilled water. The study summerized briefly below contains new supplemental information on the behavior of RPA 202248 in a non-buffered aqueous system 2000 contains and contain

°N'	
Report:	⊒; ;, ;, ;, ;, ;, ;, ;, ;, ;, ;, ;, ;, ;,
Title:	Photolysis in distilled water RPA202248
Report No:	$\mathcal{S}^{T} = C \mathfrak{Q} \mathfrak{Z} \mathfrak{Z} \mathfrak{Z} \mathfrak{Z} \mathfrak{Z} \mathfrak{Z} \mathfrak{Z} Z$
Document No(s):	Report Deludes Frial Nos.:
	36668 ~ ~ ~ ~
×	$\mathcal{L} = \mathcal{L} = $
A A	M-214288-01-1
Guidelines:	Veviation not specification
GLP/GEP:	The A of A
<u> </u>	

The photolysis of RPA 202248 was devestigated in distilled water at $25 \pm 2^{\circ}$ C under artificial sunlight (xenon lamp out-off wave length below 290 nm) for 7 days, equivalent to a maximum period of 10 days of natural sublight. The investigations were performed with non-labelled RPA 202248 at the two test concentrations 0.05 mg/L and 0.5 mg/L.

Analysis be HPLC/MS/MS showed that no decline had occurred in irradiated or dark control samples.



It was therefore concluded that RPA 202248 was stable under the conditions of aquatic photolysis. Photolytic processes thus do not contribute to the overall elimination of RPA 202248 from the application of environment.

CA 7.2.1.3 Indirect photochemical degradation

Being regarded as higher tier option no studies on the behavior of soxaflutole and its residues under conditions of indirect photochemical degradation have been performed.

Conditions of indirect photochemical degradation have been performed.

CA	7.2.2	Routeand	kate o	f biologi	cal dee	adation	in aquat	ic systems
~ .			0	×		~	1	
CA	7.2.2.1 🔻	"Ready bi	odeģiva	dabólíty		^o		

Report: 🖉 🏾		;1996;M-211235-01
Title:	Determination of ready biodegrability:	Carbon dioxide (CO2) evolution test (modified
A CONTRACTOR	Sturin test) with iso aflutole	
Report No:	C026345	
Document No:	@ M-2112\$5-01-1	
Guidelines:	Deviation not specified	
GLP/GEP: 🔊	v yes a v	

The ready biodegradability of isoxaflutole was investigated experimentally by:

• Ancubation of non-labelled active substance at a concentration of 30 mg a.s./L in an activated sludge of pH 7.4 to 7.7 under the conditions of a modified Sturm test at 20 to 22°C (KCA 7.2.2.1/01).



The data requirement was addressed under Point 7.2.1.3.1 of the Dossier submitted and evaluated within the process for Annex I inclusion as published in the corresponding Monograph of RMS the Netherlands (February 1997) and its amendments.

Consequently there is no detailed description of this existing data in this update

The evaluation revealed that isoxaflutole can be regarded as not readily biodegradable. This and ing was confirmed by the results of biological degradation tests performed in surface water (K&X 7.2.2.2) (01) and the existing sediment/water study (KCA 7.2.2.3)(01).

¢, Being a new data requirement this point had not been addressed on the Existing Dossier submitted and evaluated within the process for Annex Kincluston as published in the corresponding Monograph of RMS The Netherlands (February 1997) and it's amendments

Report:	4; ;2013;M-459354-02; ;mended: 2013;07-02
Title:	[14C] Isoxaflutole: Aerobic mineralization in surface water 0
Report No:	D62847 ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~
Document No:	M-459354-02+1 2 4 m 4 m
Guidelines:	OECD Guidelines for the Festing of Chemicals, Guideline 309, April 13, 2004; not
	applicable 2 2 2 2
GLP/GEP:	thes of a contract of the cont

Executive Summary

The mineralisation of phenyl-UL-14C-labeled active substance is xaflutole was investigated in nonsterile natural water of pH, 125 at test concentrations of 20.9 µg a.s./b (low dose) and 108.5 µg a.s./L (high dose) (Samples were incubated at 3.3 ± 0.75 °C in the dark for a maximum of 58 days.

Microbia activity of the test water was demonstrated by meubation of phenyl-UL-14C-labeled benzoic acid serving as reference. The mean reference radioactivity ranged from 100.1 to 104.8% AR for treated samples of both doses.

Values of the test substance in the test water decreased pipidly from 87.5% of AR for the low dose (93.3% for high dose) at time zero to 1.2% (5.4% for high dose) after five days of incubation with no detects in the following.

Isoxafluçõe was transformed the major and predominant product RPA 202248. Formation of other transformation products including volative components and carbon dioxide was minimal accounting for less than 2.5% in maximum for a single component in the course of the study.

Values of the DT 50 of isoxad utole under conditions of mineralization testing were calculated to be 0.68 days (lev dose) and 0.99 days (high dose). No DT₅₀-values could be determined for metabolite RPA 202248 since insufficient degradation was observed under the conditions of the test.



Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

I. Material and Methods

A. Materials

1. Test Material: [Phenyl-UL-¹⁴C]Isoxaflutole Specific radioactivity: 4.34 MBq/mg Radiochemical purity: >99% Chemical purity: not reported Sample ID: KML 9369 / 253323/A

2. Test water

at Moeldor 2-1 m depth) from a Jake at Moeht The natural water used for the test was freshly conjected (0 to Aargau (AG), Switzerland. Water samples were characterized as summarized in Table 7

	Table 7.2.2.2-1:	Physico-chemical	characteristics	of test water	•
--	------------------	------------------	-----------------	---------------	---

			<u> </u>		,
Water			/ <u>_</u> O'	Pond Water	Ž
pH	5° %'			7.4	Ű
Colour	10%		N N	not reported	
Concentration of dissolved	xygen at coa	lection (mg/I		~ ⁰ 7.44 ~ ⁰	×
Total hardness (°dH)		R L	Q.	12.00	Z
Biological oxygen demand (mg/L	4. o		? <4_0	0
Total organic carbon (TOC,	mg/kg) (13.05)
Dissolved Organic Carbon (DOC, mg (b)	4 0	,	¥4.05	
Total phosphorus (mg/L)	ĝ Õ		S (1	0.20	
Total nitrogen (mg/L)	, 	\$ \$		3076	
Nitrate M_3^- (mg/L)	Å.		Ø ,	6.54	
Nitrite DO2 ⁻ (mg/L)) ₍₁)	<0.8	
Ammonium H_4^+ (mg/L)	× , 67	XX A	Ka v	2.94	
	×	à _c u			

vas possed through @ 0.2 mm sieve. Before start of incubation the test

B. Study design

1. Experimental conditions: Samples of 300 mL test water each were filled into all-glass incubation flasks and pre equilibrated pror to treatment at approximate study conditions (darkness, 20 °C). The test was performed with ibenyl UL-14 Cesoxat atole a initial concentrations of 11.08 µg/L (low dose) and 109 40µg/L (high dose). For the potential generation of metabolites, samples were also incubated that had received an exaggerated dose of 1000 up L. Following application the samples were attached to flow-through systems allowing moisturized air to pass through and with traps to collect ¹⁴C-carbon dioxide and other volatiles (2 M agreeous potassium hydroxide and ethylene glycol). Samples were incubated at $2\frac{3}{2}3 \pm 0\frac{3}{2}5$ °C in the dark for 57 days in maximum.

In addition, samples containing untreated water, solvent controls and biological controls were incubated under the same conditions and removed for analysis at selected time points. Solvent controls and biological controls contained the reference substance phenyl-UL-14C-benzoic acid.

2. Sampling: Duplicate samples each of both test concentrations were removed for analysis after 0, 1, 2, 5, 14, 28 and 57 days of incubation.

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

Samples for determination of microbial activity (biological controls) were investigated after 0, 5 and 14 days of incubation. Solvent controls were taken for analysis after 14 days of incubation. Finally, of sterile controls were removed for analysis after 55 days.

The complete samples were immediately processed and HPLC analysis was usually performed the same day. Therefore no additional investigations of storage stability were necessary. The pH, oxygen concentration and the redox potential was determined at each sampling interval.

3. Analytical procedures: The water was analysed directly without a concentration step prior to analysis. The ¹⁴C-material balance was established for each sample tollowing analysis of the water and determination of volatile radioactivity in the traps for high dose samples and following quantitation of radioactivity in water by LSC, analysis was performed by reversed phase HPLC and ¹⁴C-trow-through detection techniques. Samples of the low cose were analysed by TLC followed by ¹⁴C-detection (phosphor imaging).

Based on the lowest integrable peak, the LOD was estimated to be about 1.0% of AR for a compound in low dose and about 1.2% AR for these in the high dose samples of the second s

4. Kinetic evaluation: The kinetic evaluation was performed for the active substance isoxaflutole and its readily formed metabolite RPA 202248 with the software KinGUL collowing FOCUS kinetic guidance (2006) to obtain bestoris to the measured data.

JI. Results and Discossion

The temperature was maintained a 23.3 \pm 0.75 \pm during the test. Biological activity of the test water was confirmed by the degradation of reference substance phenyl-L/L-¹⁴C benzoic acid within 14 days of incubation. The pH oxygen concentration and redox potential of the test water was shown to be within the same range for treated samples and for untreated controls.

The material balances and distribution of radioactivity are summarized for irradiated samples and dark controls in Table 7.2.2.2-2 (low dose) and Table 7.2.2.2.3 (high dose). The mean material balances were 102.4% \pm 1.9% AR for low dose samples and 102.0% \pm 2.2% for the high dose. The complete material balances indicate no significant losses of radioactivity from samples in the course of the test including processing to random size.

Formation of 4 C-carbon doxide or other volatile components was negligible to account for less than 0.2% of AR for both concentrations tested.

Biotransformation of phenyl-UL-¹⁴C-labeled isoxoflutole was fast to result in a decline of values from 87.5% AR at time zero to 2.2% after 5 days for the low dose and from 93.3% AR at time zero to 5.4% after 5 days for the high dose of begradation of active substance was quantitative also in sterile controls as it is documented by a value below the LOD after 55 days of incubation.

Formation of miner fractions added up to maximum values of 3.8% after 57 days distributed into two components with none present at more than 2.1% AR for the two test concentrations in the course of the study.

RPA 20224 was observed as the major and predominant transformation product at maximum values of 101.3 AR by day 5 at the low dose and 102.9% by day 14 at the high dose. RPA 202248 was also in sterile samples. This serves as an indication that the transformation of isoxaflutole to RPA 202248 is driven chemically by pH (sterile controls) or by both, biotical and chemical, as demonstrated from incubation of the microbial viable samples.

The kinetic evaluation of the data underlined the fast degradation of isoxaflutole to form RPA 20248 to result in DT_{50} -values of 0.68 days for the low dose and 0.99 days for the high dose. The values were derived from the SFO kinetic model as the best fits to measured data. No experimental value of the DT_{50} could be calculated for RPA 202248 since no decline was observed under the conditions of the test. The results of kinetic evaluations are summarized in Table 7.2.2.2-4

		Q	ġ,	Sampl	ingmeter	val (days)	Ô, ĝ	
Component		Ś	Â,		5 4	^م ر 14	28×	57
Isoxaflutole	Mean*	87.5	45.6	2.0	1.2	0.0	0.0 J	م ک 0.0 (
	SD 🔬	± 0.3	±0.9	æ9.5	<u>_</u> 2	$O^{\pm 0.0}$	±0.0 [©]	±Q:0
RPA 202248	Mean	¢14.7	√ 59.0 °	\$96.5 ₄	P101.3	98 6 7	28.5	099.9
(Diketonitrile, M1)	SDÔ♥	£40.9	\$±1.2	±1.10°	±102	£2.2	©±0.8 Ø	≥ ±0.2
Unknown M2	Mean*	0.0	0.0	£19.6		^O 2.1 ^O	0.7	2.1
	ØSD , ∜Ű	±0.0	€0.0 گ	©±1.6.C	$\mathbb{V}_{\pm 0.0}$	ź ±	% ≢0.1	±0.1
Unknown M3	Mean*	چ 0.0 چ	0.0	0.4	~Q.O	©1.2	© _{0.8}	1.7
- Second Second		±0,0°	±0.0	±0.4	¢\$0.0	10.0	±0.1	± 0.0
Total radioactivity	Mean*	£02.3 /	@04.6	°101⊉	102.5	102,0	100.0	103.7
in water	SP C	, ±0.5	≠0 <u>3</u> 3	±1.9	0.0	₹¥2.6	±0.8	±0.2
14CO	Mean*	n.a.	<u>`</u>	S ^{0.1}	ر <0.1_@	<0.1	< 0.1	< 0.1
	∕ SD _{&}	Spi.a.	£40.1 *	$\mathbb{D}^{2}\pm0.0$	± 0.0	± 0.0	±0.0	± 0.0
Other volatiles	Mean* 🖇	n.a.Q	<0,0	~@?¶`	Ø.1	< 0.1	< 0.1	< 0.1
	SD A	file.	f9a.	©n.a.	n.a.	n.a.	n.a.	n.a.
Total relienstivity (9/2	Mean*	402.3	§104.8	1010	102.5	102.1	100.1	103.7
		±0,5	+0?3	±1.5	±0.0	±2.6	±0.8	±0.1

Table 7.2.2.2-2: Degradation of [phenyl-UL-¹⁴C]isoxaflutole in aerobic natural water greated at low texpressed as percentage of total applied radioactivity

Values given as percentages of intrally applied radioactivity SD = standard deviation Mean values of two peplicales

n.a. = not analysed or not app@cable

Table 7.2.2.3: Degradation of phenyl-UL-¹⁴ Pisoxaflutole in aerobic natural water treated at high dose, expressed as percentage of total applied radioactivity

				Sampl	ing inter	val (days)		
Component 2 A	у ^г	0	1	2	5	14	28	57
Isoxatiutole	Mean*	93.3	58.3	19.5	5.4	0.0	0.0	0.0
	SD	±1.7	±1.2	±3.5	±0.4	± 0.0	± 0.0	± 0.0
RPA 202248	Mean*	11.0	43.6	83.6	93.3	102.9	102.8	101.0
(Diketonitrile, M1)	SD	±2.5	±0.6	±2.9	±0.8	±0.1	±0.2	±1.3
Unknown M2	Mean*	n.d.	n.d.	n.d.	2.4	n.d.	n.d.	n.d.

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

	SD	n.a.	n.a.	n.a.	±1.2	n.a.	n.a.	n.a. 。
Total radioactivity	Mean*	104.3	101.9	103.0	99.9	102.9	102.8	1000
in water	SD	±0.8	± 1.8	±0.6	±2.4	±0.1	±0.2	ja 1.3
1400	Mean*	n.a.	< 0.1	< 0.1	<0.1	S9 .1	<0.1	<0.10
CO_2	SD	n.a.	± 0.0	± 0.0	±0.0	€_±0.0	±0,0,	
Other valatilas	Mean*	n.a.	<0.1	<0.1	<0.1	∛ <0.1	≈ <u>0</u> .1	\$0.1 K
Other volatiles	SD	n.a.	± 0.0	± 0.0	±Ø	±0.0	$2.0^{-40.0}$	
Total radioactivity (%)	Mean*	104.3	100.9	103.1	99 9.9	102.%	1029	101.0 O
	SD	±0.8	∰±1.8	±0.6	±204)°	±0%1	~≠0.2	Cu_±1.3

Values given as percentages of initially applied radioactivity

SD = standard deviation; * Mean values of two replicates

Table 7.2.2.4: Kinetic evaluation	of the	degradation	of lph	envl-BL		aflatole	infaerobic	natural
water under condition	s of ÓE	COP 309 testin	g 🔊	\sim	A Contraction of the second se	S.	§ .?	

	()				
Compound / Dose	Kinetic Model	days)	DT90 (days)	For Chi ²	r^2
Isoxaflutole / Low	ŚFO	¢۶ 0.6	2.27 K	× 16.14	0.974
	j FOMO	, 0, 6 3	2.10	~~ 19 0 6	0.972
Isoxaflutole / High	[≫] §£F0 [∞]	.	⁄ <u>3</u> ,20°,	≈ \$ 776 <u>\$</u> ₹	0.987
*	J KOMC ?	O _{0.89}	×2.94	الالمي 11.60 (K)	0.985
RPA 202248 / Low	SFQ	U n.🖓	n.c.	O″n.c.∳	n.c.
	FOME	n.c. a	n.e	L 1009.	n.c.
RPA 202248 / High	SFO (∽n.c. √	12C.	n.c.	n.c.
	FOMCO	n.c.	Sn.c.	n.c.	n.c.



sterile natural water was nsignificant under the 'pelagic' conditions of the test.

20224 Was abserved as the single and predominant major transformation product thus requiring RPA consideration in environmental risk assessments.



The DT₅₀ of isoxaflutole in water under conditions of aerobic mineralisation testing was calculated to be 0.68 days for the low dose and 0.99 days for the high dose tested thus indicating some dependency of transformation rate upon the test concentration.

No experimental value of the DT₅₀ could be calculated for RPA 202248 since no decline was observed under the conditions of the test.

under the conditions	of the test.	4	
Overall the results an	re well in line with observation	s madem existing water/sedimer	nt fests.
CA 7.2.2.3 Wa	ter/sediment study		
Report:	i;	: @95;M& 58508-@1	A A L'
Title:	RPA 201772: Degradation and	retention in two water / sediment sy	stems &
Report No:	R000377		
Document No:	M-158508-01-1		A O
Guidelines:	BBA: Part IV, Section 5.1, (14	90);nørspecified	Å Å
GLP/GEP:	yes yes		Š, L

Report:	g; 2013;2€-4649¢€01 ° ° ~
Title:	Isoxafferole: Kinetic prodelling evaluation of water sediment degradation study data to
	derive total Stem DF 50 values
Report No:	VCM3/008A & & & & & & & & & & & & & & & & & & &
Document No:	M-464901-10 0 0 0 0 0 0 0
Guidelines:	EU Council Directive 91/414/EEC as amended by Commission Directive 95/36/EC of
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N.	Qcrober 2009; not applicable 2 2 2
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Ŏ,	
Report: 🔊 🖗	n; ;2043;M-464903,01
Title:	Is xafly or Exactly the second s
	Werive Water phase DTS0 values 20 20
Report No:	VC/43/008B

Document No: M₄4649032Ø1-1 🔬 Compcil Directive 91/414/CEC, as amended by Commission Directive 95/36/EC of **Guidelines:** July 1995, Section 5, Point Land Commission Regulation (EC) No 1107/2009 of 21 October 2009; not applicable **GLP/GEF**

flotole under conditions of water/sediment testing was investigated in: The degradate

two contrasting sediments and their associated water following application of phenyl-UL-14Clabelled getive substance and incubation at 20°C (KCA 7.2.2.3 /01).

The dat Drequirement was addressed under Point 7.2.1.3.2 of the Dossier submitted and evaluated within the process for Annex I inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments.

Bayer CropScience Document MCA: Section 7 Fate and behaviour in the environment

Isoxaflutole

Consequently there is no detailed description of this existing data in this update.

The evaluation revealed that isoxaflutole was degraded rapidly to form metabolites RPA 2022 and RPA 205834 as major metabolites (maximum of 70.3% AR and 26.4%, respectively). The formation of RPA 202248 and RPA 205834 was observed in parallel from the active Substance. As a result of biotic processes, both initially and fast formed metabolites could thus serve as predecessors for the benzoic acid RPA 203328. In addition, RPA 207048 was observed as a minor component below \$ AR in the water or the sediment. Being transient this compound was likely to deamination of the predecessor compound RPA 205824.

Half-lives for degradation ranged from 0.34 to 0.53 days for isoxaffutole while values were estimated to 255 to 700 days for RPA 202248 and 52 to 97 days for RP estimated for RPA 203328.

Due to their occurrence as major metabolites in water/sedument, 205834 and RPA 203328 had been defined as residues to be considered environmental risk assessments for surface water.

The results of degradation tests in water/sediment systems under conditions of the laboratory resulted in the metabolic pathway sommar ded in Figure 9.2

Figure 7, 22, 3-1, Proposed pathway of metabolism of isoxaflutole in water/sediment systems



Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole



Executive Summary

The kinetics of degradation of iso callutore in total systems was evaluated from data of tests performed in two water/sediment systems with phenyl- 10^{-14} C-radiolabeled active substance (KCA 7.2.2.3 /01).

The kinetic evaluation followed FOCO'S guidance to derive values for the degradation in total sediment water systems from best fits to measured data for use as modeling endpoints in aquatic exposure assessments and for comparison against trigger endpoints. Separate analysis was performed for isoxaflutole and its metabolites RPA 202248 (AE 0540092) and RPA 205834 (AE 0692291) at Level I for the total systems with results summarised in Table 7.2.2.3-1 and Table 7.2.2.3-2.



For modelling endpoints the kinetic evaluation resulted in an 'all-SFO' fit of residue data for the system. Best fits were obtained for the system from combination of the Hockey Stick model for parent substance with the SFO model for the metabolites. A geometric mean value for the DegT₅₀ of 0.36 days was calculated for the parent compound isoxaflutole. For metabolites the values are 237 days for RPA 202248 and 76.2 days for RPA 205834. The evaluation, resulted in average formation fractions of 0.71 for formation of RPA 202248 and 0.23 for formation of RPA 205834 each from the parent compound. system the kinetic evaluation resulted in an For persistence endpoints and for the SFO' fit for residue data. Best fits were obtained for the system from combination of Hockey Stick model for parent substance with SEO models for the metabolites. A worst case Degt 50 in total system of 0.57 days was derived for the parent compound is a flutore (vstem). For metabolites the values are 334 days for RPA 202248 and 81 4 day of r RPA 205834. Worst case values for the DegT₉₀ in total systems are 1.9 days for the parent coppound more than 1000 days for RPA 202248 and 270 days for RPA 205834

 Table 7.2.2.3-1:
 Modeling endpoints, half-lives and formation fractions for the degradation of isoxaflutole residues in total @ater/sediment@ystems/accor@ng tofOCUSLevel

	M.S.		· · · · · · · · · · · · · · · · · · ·		<u> </u>
Compound	\sim		Label	Dieg T 50 m	ffm
compound			~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		<i>(()</i>
	, Ô	o _s y		(days)	\$ N
Isoxaflutole	× A			0.57	- S
	*			023	- 1
	Û,		Mean (geometric)	<u> </u>	-
RPA 202248				334	0.75
<u> </u>				168	0.68
ð,	S IO	0' &	Mean (geometric)	3 237	-
line and the second sec		à s	N verage	<u>×</u> -	0.71
RPA 205834				81.4	0.25
	, Ŭ , Ŏ			O 71.5	0.22
~) { ,	×, 0'	Mean (geometric)	76.2	-
		v dv s	Average	-	0.23
Label 1 = phenyl; ffm	= formation fra	action 🔊			

 Table 7.2.2 3-2:
 Persistence endpoints: falf-lives for the degradation of isoxaflutole residues in total water/sediment/systems according to FOCUS Level I

Compound V	System Q	Label	ffm	DegT50 (days)	DegT90 (days)
Isoxaflutole		2 1	-	0.57	1.9
	7 É	1	-	0.21	1.2
RPA 202248		1	0.75	334	>1000
		1	0.68	168	559
RPA 20583405	× ×	1	0.25	81.4	270
		1	0.22	71.5	237

Label 1 = henyl; ffm = formation fraction

Document MCA: Section 7 Fate and behaviour in the environment Isoxaflutole

I. Material and Methods

The kinetic evaluation was based on data of a water/sediment study (KCA 7.2.2.3/01) conducted with a phenyl-UL-¹⁴C-labeled isoxaflutole in a sandy (**1999**) and a loamy sediment (**1999**) and their associated water at 20°C in the dark for a maximum of 100 days.

Data pre-processing

Generally, replicates were taken into account separately. The data were checked for consistency and clear outliers. Data for non-extractable residues (NER) and CO_2 were not fitted within the evaluation (open system).

For the residues in the total sediment/water systems the following procedure was applied:

- For data processing of day zero samples, radioactivity assigned to metabolities, non-extractable residues (NER) and CO₂ was added to the parent compound and thus metabolitic concentrations were set to 0 %. Parent compound was attributed to the water phase only thus esulting in a value of zero tor the sediment phase, since the test substance was applied to the water phase.

- Residues values below the limit of detection (LOD) = 0.01% of ΔR) were set to 0.5 times the LOD for the first non-detect at the end of the curve. The curve could be cut at this time paint in case of no later detects. For metabolites, the dast non-detect at the beginning of a curve was set to 0.5 times the LOD for occurrences later than day 0.

Kinetic models

The kinetic evaluation of water/sediment data was performed according to FOCOS Level I to result in degradation kinetics in total systems

No evaluations according to Level II were performed since not regarded as mandatory. For lower-tier calculations or the comparison with persistence triggers a Level I evaluation of the dissipation may be often appropriate.

Contrary to the parent, for metabolites it may be often neither feasible nor meaningful to differentiate between SFO and the biphasic model using Level I and a simultaneous fit of the complete metabolic pathway (i.e. considering formation and decline of metabolites). A bi-phasic approach would result in too many free parameters needed to bescribe such systems. Even for SFO the number of free parameters is often at the limit and the use of bi-phasic konetics could easily multiply the number of free parameters.

The inferring of kinetic degradation parameters followed the proposed metabolic pathway as given in Figure 7.2.2.3-1.

Each compound was represented by one compartment as the total of measured occurrences in water and sediment with no values associated with a sink compartment. Between compartments transformation reactions were assumed to proceed only one-way. The initial amount of the parent compound was free fittee and the initial amount for metabolites was fixed to a value of zero. All data were weighted equally thus corresponding to an absolute error model.

At reast four kinetic models consisting of single first-order (SFO), first-order multiple-compartment (FOMC Gustafson-Holden), double first-order in parallel (DFOP), and the hockey-stick (HS) model were available, in principle, according to the set of models proposed by FOCUS.

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While best-fits should be taken to derive trigger or persistence endpoints SFO should be used to derive modeling input parameters if an acceptable fit can be obtained.

Before use of bi-phasic kinetic models FOMC, DFOP and HS the following major cases were taken into account:

1. A check whether a degradation or dissipation to 10% of the initial amount M₀ was reached within experimental period, then the estimation of the DT_{50} could be simplified according. O the relation $DT_{50} = DT_{90}/(\ln(10)/\ln(2))$. By this method the equivalence SFO-curve presets the bi-physic curve at the time DT_{90 bi-phasic} and consequently the residue values at earlier times are over-prediced. 2. In case a value of 10% for M_0 was not reached within the runtime of the study **POMC** should not be used to derive modelling endpoints.

3. In case a value of 10% for M_0 was not reached within the contine of the stude, the DP_{50} could be derived for DFOP and HS models from the slow part of the bi-phasic curve using the relation $DT_{50} = \ln(2)/k_2$.

Statistical evaluation

The identification of the most appropriate kinetic model for the description of speriphental data according to FOCUS is mainly based on the three coveria of visual assessment of fits of calculated transformation curves to experimental data, the value of error of CR -square (χ^2) dest and a single-sided significance t-test.

The choice of the appropriate kinetic model was primarily based an visual assessment of the fit and the Chi²- (χ^2 -) error.

Within the current saluation, single first-order (SFO) kinetics had been tosted first, since SFO is being used as the simplest kinetic model almost exQusively in environmental exposure models. In case the SFO furthoutd not be visually acceptable or in case of significant exceedance of value for χ^2 -error of 15%, bi phasic models were rested Finally the model was chosen which was visually acceptable and provided a significantly better $\hat{O}t$ in terms of the χ^2 -terror.

The approach avoided the use of over-parameterised models simply and only being chosen on the basis of a marginally better fit finally it should be noted that a value of χ^2 -error below 15% should only be considered as guidance and not as an absolute out-off criterion. This is true, in particular, for the modelling of metabolite data with grors for χ^2 being higher, but with fits still representing a reasonable description of their formation and degradation behaviour.

II Results and Discussion

The kinetic evaluation of water-sectment data was performed according to FOCUS Level I to result in degradation sinetics in total systems (Table 7.2.2.3-3 and Table 7.2.2.3-4). No evaluations according to Level Wwere performed.

For the two data sets under investigation, the application of an all-SFO kinetic model to the parent substance and metabolite data resulted in acceptable fits for the system.

For the data, the fits could be optimised by a bi-phasic approach applying the Hockey Stick model to the parent substance data and the SFO model for the metabolites RPA 202248 and



RPA 205834. Apart from low Chi²-errors there was also no sign of systematic variations of the residuals. Consequently, no further testing of other kinetic models was considered necessary.

Table 7.2.2.3-3: Modelling endpoints, half-lives for the degradation of isoxaflutole and metabolites in the systems Level I n

C 1	G (р т	рт	66	CI 2			Ĉn
Compound	System	Deg 1 50	Deg 1 90	ffm	Chr->	t-test 🚬 🤤	\ \$% A _≪ []	Ĩ
		(days)	(days)		(%)	×,×		, ,
Isoxaflutole		0.57	1.9 💎	-	5.9	2.67e 95	Qery good	L
		0.23	0.78	- "(D [™] 10.8	1.04e-05	Good	,0″
	Mean (geometric)	0.36	A C	Q.	_ °		0,0	,¥
RPA 202248		334	©⊳1000	Q.75	<i>J</i> .4	~Q.0005830	Sery good	
		168 🔬	55%	@ .68 🔬	8.6	r 0.0 00 254 🦕	Good	
	Mean (geometric)	237					4	
	Average	4		0, 5 1	Õ			
RPA 205834		81	~ 270~	<u>≈</u> 0.25	A16.4	0.000343	Good Good	
		6 ^{4,5} , ~	287 <u> </u>	€0.22 ₀	23,6%	0.0001603	Good	
	Mean (geometric)	<i>₹</i> 76.2 <i>€</i>			Û,			
	Average	K O	\sim	0,23	×		4 J	

ffm = formation fraction; $Chi^2 = Chi$ square error; t-est = t-probability test; $\Im A$ Vosual Assessment : SFO model for parent substance, SFO for pretabolites

: Hockey Stick model for parent substance, SFQ for metabolites

Table 7.2.2.3-4: Persistence endpoints, half-lives for the degradation of isoxaflutole and metabolites in total systems Level

	$\mathcal{O}_{\mathcal{V}}^{\mathcal{V}}$					0 '	¥	
Compound	System	S D	egT 50	Degi 90	[©] ffm _ℳ	Chi ₂	t-test	VA
			days)	(dáys) 🔊	Γ_Q ^γ	(%)		
Isoxaflutole	O		0.570	× ^{1.9}		5.9	2.67e-05	Very good
			0,21 🔊	2 1.2	0.	\bigcirc	k1 5.06e-08	Excellent
			4 8	"0" (i	v , \overline{v}	e 2.3	k2 0.00729	Excellent
RPA 202248		N.C.	°334	\$1000°S	0.55	7.4	0.005830	Very good
~~¥		Ś	168	°∼y 559	0%68	8.6	0.000254	Good
RPA 205834		Ô,	\$1×.4 @	2000 _{>}	0.25	16.4	0.000343	Good
			71.5	\$ ²³⁷ \$	0.22	23.6	0.001603	Good

ffm = formation f action, $Chi^2 \in Chi$ square error; t-test = t-probability test; VA = Visual Assessment

Ø

M. Conclusion

-14@-labered isocraflutole under conditions of a water/sediment test was The degradation of pheny TI shown to proceed via the formation of metabolites RPA 202248 and RPA 205834.

ŝ For modelling endpoints the kinetic evaluation resulted in an 'all-SFO' fit of residue data for the stem. Best fits were obtained for the system from combination of the Hockey Stick model for parent substance with the SFO model for the metabolites. For persistence endpoints and for the system the kinetic evaluation resulted in an 'allfugor residue data. Best fits were obtained for the SFO' system from combination of Hockey Stick model for parent substance with SFO models for the metabolites.



For modelling endpoints the kinetic evaluation resulted in a geometric mean value for the DegT₅₀ of 0.36 days for the parent compound isoxaflutole. For metabolites the values are 237 days for RPA 202248 and 76.2 days for RPA 205834.

For evaluation against persistence triggers the worst case DegT₅₀ in total system is 0.57 days (

202248 and 81.4 days for RPA 205834. Worst case values for the DegT, in total systems are 1.9 days for the parent compound, more than 1000 days for RPA 202248 and 200 days for RPA 202834.

The results can therefore be used as input parameters for modelling in environmental risk assessments and for evaluation against persistence triggers $\sqrt{2}$ $\sqrt{2}$ $\sqrt{2}$ $\sqrt{2}$

	$\underline{A} \overline{\partial}^{*} \overline{\partial}^{*} \overline{Q} \underline{A} \overline{\partial}^{*} \overline{\partial}^{*} \overline{\partial}^{*} \underline{A} \underline$
Report:	*; (20)3;M-#64903;04, 0° (
Title:	Isoxaflutole: Kinetic modelling evoluation of water sediment degradation study data to
	derive water phase DT to walue water water phase DT to walue water
Report No:	VC/13/008B
Document No:	M-464903-0% 1 & & & & & & & & & & & & & & & & & &
Guidelines:	EU Council Directive 91/2414/EEC, as genended by Commission Directive 95/36/EC of
	July 1995, Section 5, Point 7 and Commission Regulation (EC) No 107/2009 of 21
	October 2009; not applicable or 🔨 🚿 🖉
GLP/GEP:	

Executive Summary

The kinetics of dissipation of isoxaflutole from water was evaluated from data of tests performed in two water/sediment systems with phenyl-UL 4 C-radiolabeled active substance (KCA 7.2.2.3 /01).

The kinetic evaluation followed POCUS guidance to derive values for the dissipation from the water phase in ediment/water systems from best fits to measured data for use as modeling endpoints in aquatic exposure assessments and for comparison against trigger endpoints. Separate analysis was performed for isocallutole and its metabolites RFA 202248 (AE 0540092) and RPA 205834 (AE 0692291) at pevel 1 for water with results summarised in Table 7.2.2.3-5 and Table 7.2.2.3-6.

For modelling endpoints the kinetic evaluation resulted in an 'all-SFO' fit of residue data for the

SFO model for parent substance and RPA 202248 with the DFOP model for RPA 205834. A geometric mean value for the Dis B_0 of 0.36 days was calculated for the parent compound

isoxallutole. For metabolites the values are 62.2 days for RPA 202248 and 28.3 days for RPA 205834.

For persistence endpoints and for the **sector system** the kinetic evaluation resulted in a SFO fit for residue data of parent substance, a Hockey Stick model for RPA 202248 and a DFOP fit for RPA 202834 Best fits were obtained for the **sector system** from combination of Hockey Stick model for parent substance and RPA 202248 with the FOMC model for RPA 205834.

A worst case $DisT_{50}$ of 0.57 days in water (**Example 10** system) was calculated for the parent compound isoxaflutole. For metabolites the values are 57.1 days for RPA 202248 and 16.7 days for



RPA 205834. Worst case values for the $DisT_{90}$ in total systems are 1.9 days for the parent compound, 328 days for RPA 202248 and 163 days for RPA 205834.

Table 7.2.2.3-5: Modelling endpoints, half-lives for the dissipation of isoxaflutole and metabolites from water systems Level I

Compound	System	Label	DisT ₅₀	
		Ĉ.	(days)	
Isoxaflutole			0.81	
		L 1	Ø?23	
	Mea	an (geometric)	0.36 °	
RPA 202248			× 539 Q	
			× ×1.8	
	Nea	an (geometric)	£ 62.2 ×	Star and a star and a star a sta star a star a sta
RPA 205834	<i></i>		330	
	*	× × ×	× <u>2</u> 3.6	
	Mei Mei	an (geometric)	<u>^28.3</u>	
Label 1 = phenyl	j ² 4 .			

Table 7.2.2.3-6: Persistence endpoints, half-lives for the dissipation of isoxaflutole and metabolites from water, Level L

	.6	×.	Ci 🗸	(index)	Y <u>'~' . 0</u>	A
Compou	nd 🦓	©″ Sys	tem	Label	DisT ₅₀	DisT ₉₀
	V	1	S AS		(days)	(days)
Isoxaflutole		Ş, v		Ì ,¥⊳	° [∞] 0457 °√	1.9
	<u> </u>		Š	J. a.	©	1.2
RPA 202248	5°, 0°	۲	¥	× 1 ×	\$ 26,1 ¢	328
	' >> [~~		۶´ ا	5745	289
RPA 205834 O				Ĉ	¢6.7	95.5
	* .	K)) }		15.7	163
Label 1 z p henyl	Ô.			\$ \$	^o	
4 Y	N N	\$	A' N	(. A	Ý	

A Material and Methods

The kinetic evaluation was based on data of a water/sediment study (IIA 7.2.2.3/01) conducted with phenyl-UL₂¹⁴C-labeled isoxafluter in a sandy (**14**C) and a loamy sediment (**14**C) and their associated water at 20°C in the dark for a maximum of 100 days.

Data pre-processing

Generally, replicates were taken into accounce parately. The data were checked for consistency and clear outliers. Pata for non-extractable residues (NER) and CO_2 were not fitted within the evaluation (open system).

For the residues in the water the following procedure was applied:

- For data processing of day zero samples, radioactivity assigned to metabolites, non-extractable residues (NER) and CO_2 was added to the parent compound and thus metabolite concentrations were set to 0 % Parent compound was attributed to the water phase only thus resulting in a value of zero for the sediment phase, since the test substance was applied to the water phase.

- Residues values below the limit of detection (LOD = 0.01% of AR) were set to 0.5 times the LOD for the first non-detect at the end of the curve. The curve could be cut at this time point in case of no

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later detects. For metabolites, the last non-detect at the beginning of a curve was set to 0.5 times the LOD for occurrences later than day 0.

Kinetic models

The current kinetic evaluation of water-sediment data was performed according to FOCUS Level to result in dissipation kinetics in water.

No evaluations according to Level II were performed since not regarded as mandatory. For lower ther calculations or the comparison with persistence triggers a Level I evaluation of the dissipation may be often appropriate.

Contrary to the parent, for metabolites it may be often neither feasible for meaningful to differentiate between SFO and the bi-phasic models, using Level I and a simultaneous fit of the complete metabolic pathway (i.e. considering formation and decline of metabolities). A bi-phasic approach would result in too many free parameters needed to describe such sostems. Even for SEO the number of free parameters is often at the limit and the use of bi-phasic kinetics could easily multiply the number of free parameters.

The inferring of kinetic degradation parameters followed the proposed metabolic pathway as given in Figure 7.2.2.3-1.

Each compound was represented by one compartment as the total of measured occurrences in water and sediment with no values associated with a sink compartment. Between compartments transformation reactions were assurated to proceed only one-way. The initial amount of the parent compound was free fitted and the initial amount for metabolites was fixed to a value of zero. All data were weighted equally that corresponding to an absolute error model.

At least four crnetic model consisting of single first-order (SFO), first-order multiple-compartment (FOMC, Gustafson-Holden), double first-order in parallel (DFOP), and the hockey-stick (HS) model were available, in principle, according to the set of prodels proposed by FOCUS.

While best-fits should be taken to derive frigger or persistence endpoints SFO should be used to derive modeling input parameters if an acceptable fit can be obtained.

Before a use of bi-phase kinetic models FOMC, BPOP and HS the following major cases were taken into accounts

1. A check whether a degradation or destination to 10% of the initial amount M_0 was reached within experimental period, then the estimation of the $DT_{50} \approx DT_{90}/(\ln(10)/n(2))$. By this method the equivalent SFO-curve meets the bi-phasic curve at the time $DT_{90 \text{ bi-phasic}}$ and consequently the residue alues at earlier times are over-predicted.

2. In case a value of 10% for M_0 was not reached within the runtime of the study, FOMC should not be used to derive modelling endpoints.

3. In case a value of 10% for M_0 was not reached within the runtime of the study, the DT₅₀ could be derived for DFOP and HS models from the slower part of the bi-phasic curve using the relation DT₅₀ $\leq \ln(2)$ ≤ 2 .

Statistical evaluation

The identification of the most appropriate kinetic model for the description of experimental data according to FOCUS is mainly based on the three criteria of visual assessment of fits of calculated

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transformation curves to experimental data, the value of error of Chi-square (χ^2) test and a single-sided significance t-test.

The choice of the appropriate kinetic model was primarily based on visual assessment of the stat and the Chi²- (χ^2 -) error.

Within the current evaluation, single first-order (SFO) kinetics had been tested first, since SFO is being used as the simplest kinetic model almost exclusively in environmental exposure models. In case the SFO fit should not be visually acceptable or in sase of a significant exceedance of value for χ^2 -error of 15%, bi-phasic models were tested. Finally the model was chosen which was visual acceptable and provided a significantly better fit in terms of the $\chi^2_{\text{error.}}$

The approach avoided the use of over-parameterised models simply and only wing chosen on the basis of a marginally better fit. Finally it should be noted that a value χ^2 -error below 15% should only be considered as guidance and not as an absolute cut-off criterion. This is true, in particular for the modelling of metabolite data with $\frac{1}{2}$ or χ^2 being migher but with fits still copresenting a reasonable description of their formation and degradation behaviour.

M Results and Discussion The current kinetic evaluation of water-sediment data was performed according to FOCUS Level I to result in dissipation kinetics in water. No evaluations according to Kevel IN vere performed.

For the two data sets under myestigation the application of an all-SDO' kibetic model to the parent substance and metabolitodata resulted in acceptable fits for the system with results summarised in Table 12.2.3-9 to Table 7.2.2.3-12

data, the SFO model was acceptable for the parent compound isoxaflutole and For the metabolite RPA 202248 For the data of metabolite RPA 205834 the fits could be optimised by a biphasic approach applying the DFOP model. Apart from low Chi2-errors there was also no sign of

phasic approach applying the DFOP model. Apart from low Chi²-errors there was also no sign of systematic variations of the residuals. Consequently to further testing of other kinetic models was considered necessary.

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Table 7.2.2.3-10: Modelling endpoints, half-lives for the dissipation of isoxaflutole and metabolites from water systems Level I

Compound	System	DisT ₅₀	DisT ₉₀	Chi ²	t-test	VA	Ì
		(days)	(days)	(%)	ð		
Isoxaflutole		0.57	1.9	5.9	2.72e-05	Very good	
		0.23	0.78	10.8	1.20e-05	S Good	Ô
	Mean (geometric)	0.36	4	Ś	× °*		J.
RPA 202248		53.9	179	11.7	0.0056	Sood O	_
		71.8	239	9.2	6.47g-05	Acceptable	l ô ^y
	Mean (geometric)	62.2	, O	A A	_0		Ķ
RPA 205834		33.9*	₩13*	15.6	K_0.00525	Very good	1
		23.6	78.4	11.4	0.000774	K Good	
	Mean (geometric)	28.3 💃				\sim \sim	
fm = formation	n fraction; Chi ² = Chi s	square error; Ψ -t	est ₹1-probt	ility test; VA 🔊	√isua®Åsses≰r	ment 🕰 💡	,

: SFO model for parent substance and metabolitic RPA 202248 DFOP for RPA 205835 with 101 50

back-calculated from slow phase

: SFO for parent substance SFO for each metaboli

Table 7.2.2.3-12: Persistence endpoints, half-lives for the dissipation of isoxaflutole and metabolites from water, Level I

		· ~(>	V		. 0	
Compound	System	🖉 Dis 🏹	🖓 Dis Toy	[~] Chi ²	🖉 t-test	VA
		(days)	(days)	\ (%)		
Isoxaflutole	L.	0.57		× 5.9 🕺	25,2e-05	Very good
		× 0.21			ktv 5.13e-08 k2 0.00571	Excellent
RPA 202248		26.1	3 2 8 5		k1 0.00398 k2 0.07538	Excellent
				5.4 5.4	k1 0.00491 k2 4.61e-05	Very good
RPA 205834		0 ⁶ .7	95.5 S	~19.6 0	k1 0.48412 k2 0.00525	Very good
**				گ≫ 1.0	β contains zero	Excellent

ffm = formation fraction; Chi² = Chi square error; t-test = t-probability test; VA = Visual Assessment SCO model for parent substance, Plockey Stick for RPA 202248, DFOP for RPA 205834 Hockey Stick model for parent substance and RPA 202248, FOMC for RPA 205834

IIDConclusion

The degradation of phenyl-I⁴C abeled soxaflutole under conditions of a water/sediment test was shown to preceded via the formation of metabolites RPA 202248 and RPA 205834.

For modelling endpoints the kinetic evaluation of the dissipation from water resulted in an 'all-SFO' fit of residue data for the **system** system. Best fits were obtained for the **system** from combination of the SFO model for parent substance and RPA 202248 with the DFOP model for RPA 205834.

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For persistence endpoints and for the system the kinetic evaluation resulted in a SFO fit for residue data of parent substance, a Hockey Stick model for RPA 202248 and a DFOP for for RPA 205834. Best fits were obtained for the Stick® system from combination of Hockey model for parent substance and RPA 202248 with the FOMC model for RPA 205834.

For modelling endpoints the kinetic evaluation resulted in a geometric mean value for the 0.36 days for the parent compound isoxaflutole. For metabolites the values are 62,2 day RPA 202248 and 28.3 days for RPA 205834.

For evaluation against persistence triggers the worst case DisTs, from water is 0.57 days system) for parent compound isoxaflutole. For metabolites the values are \$1.1 days for PA 202248 and 16.7 days for RPA 205834. Worst case values for the DisT with total systems are 9.9 days for the parent compound, 328 days for RPA 202248 and 163 days for RPA 205834

The results can therefore be used as input parameters for modelling in environmental osk assessments and for evaluation against persistence triggers.

Irradiated water sediment study CA 7.2.2.4

This point is regarded as a new optional data requirement in the EU. The degracation of isoxaflutole is well understood under standard conditions of water/sediment testing in view of the overall limited potential of photoletic degradation as demonstrated and er KCA 7.2.1.2/ 2 and KCA 7.2.1.2/ 03 the conduct of an utadiated water/sedement study is not regarded to result in a significantly better understanding of the behavior of isoxaflutore and its residues in the aquatic environment.

was therefore not performed An irradiated water/sedunent sud

Degradation in the saturated sone CA 7.2.3

This data requirement had Been addressed under Poin 07.2.1.4 of the Dossier submitted and evaluated within the process for Annex Enclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments

L. The evaluation revealed that the results of risk assessment in ground water demonstrated no significant risk for a contamination of sub-soils or the saturated zone by the parent compound and its metabolites, when applied according to good agricultural practice.

Therefor the parate investigations on the degradation in the saturated zone are not regarded as necessary.



CA 7.3 Fate and behaviour in air

CA 7.3.1 Route and rate of degradation in air

CA 7.3	Fate and behaviour in air
CA 7.3.1	Route and rate of degradation in air
Report:	4; ;1996M-213115-01 × × ×
Title:	Isoxaflutole - Estimation of the rate of photochemic transformation in the atmosphere
	under tropospheric conditions $\sqrt{2}$ $\sqrt{2}$ $\sqrt{2}$ $\sqrt{2}$ $\sqrt{2}$
Report No:	
Document No(s)	: Report includes Trial Nos.: 2
	95-116
	M-213115-01-1
Guidelines:	OECD: Environment Monograph No. 61 (1992); Demation not specified
GLP/GEP:	yes a grand gr

Report:	°;
Title:	Isoxaflutole: Estimation of degradation by photo-oxidation in air Model calculation
	according to Atkinson
Report No:	C038406 , 4 , 2 , 2 , 2 , 2 , 2 , 2 , 2 , 2 , 2
Document No(s):	Report instudes Frial Nos.: Of L and C
	CX/03/081 ~ ~ ~ ~ ~ ~ ~
	M-224922-00-1 ~ Q
Guidelines:	Deviation not specified & A A
GLP/GEP:	
Report:	v; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;
Title:	RPA202248: Estimation of degradation by photo-oxidation in air Model calculation

This data requirement had been addressed under Point 7.2.2 of the Dossier submitted and evaluated within the process for annexo inclusion as published in the corresponding Monograph of RMS The Netherlands (February 1997) and its amendments

Q, Ŋ The evaluation revealed that based on rapid degradation in the atmosphere, isoxaflutole would not remain stable and thus available for long range transport due to its susceptibility for reactions with photochemically produced bydroxyl radicals.

The existing calculation of the rate of transformation of isoxaflutole in air (KCA 7.3.1 /01) had been performed proof to the availability of the software AOPWIN introduced later by US EPA being regarded as the actual standard for this kind of calculation.

The rate of photochemical transformation of isoxaflutole in air has therefore been re-calculated as briefly described below (KCA 7.3.1 /02).



In addition, the rate of transformation of the rapidly formed metabolite RPA 202248 in air was calculated following the same approach as presented for the active substance with results summarized below (KCA 7.3.1 /03).

The value for the vapour pressure of isoxaflutole had been determined to 32×10^{-7} Pa at 20°G as submitted under Point 2.3.1 of the existing Dossier.

Report:	k; ;2003; 🕅 - 224922-01 🖉 👘 🖓	
Title:	Isoxaflutole: Estimation of degradation by photo-padation in air Model calculation	ition (
	according to Atkinson	
Report No:	C038406	
Document No(s):	Report includes Trial Nos.: 🖤 💦 🥎 🖓	Ż
	CX/03/081 & & & & &	₩J [°]
	M-224922-01-1 O & A A	4
Guidelines:	Deviation not specified v v v v v v v v v v v v v v v v v v v	
GLP/GEP:		4 G
		AS .

Executive Summary

The half-life time of isoxaflutole in air was estimated on the basic of structure activity relationships for the degradation of chemicals in the atmosphere developed by Atkinson¹² and transferred into the computer program AOPWINTM (v 91). The half-life of isoxaflutole in air was estimated to 1.83 days based on a typical daylight concentration of 1.5 x 10⁶ hydroxyl radicals cm³ air and a 2 hour day.

The calculations showed that is exactly a susceptible for reactions of hydroxyl radicals with various parts of the molecule thas contributing to the degradation of the substance in the atmosphere.

Nor isoxaflutole is available for longe-range transport neither the substance has a potential for accumulation in the atmosphere. Furthermore, the overall potential for occurrence of isoxaflutole in the atmosphere is additionally limited due to the own vapor pressure of the substance and its use in seed treatment.

I. Materialand Methods

Based on an approach based on structure-activity relationship (SAR) by Atkinson, the half-life and chemical diffetime of isoxaflutole in an was estimated by use of the computer program AOPWINTM (version 1.91, Syrague Research Corp.).

The program allows for an estimatation of rate constants for the reaction of photolytically produced hydroxyl radicals with organic chemicals in the atmosphere. In case of presence of olefinic or acetylenic subgroups in the chemical under assessment the calculation includes estimations of rate constants for the reaction of ozone with these subgroups. From estimated rate constants half-lives of the chemical if the atmospheric are calculated based upon average concentrations of hydroxyl radicals. AOPWINT requires the input of chemical structure via a Smiles code and an estimate for the concentration of hydroxyl radicals in the atmosphere.

¹² Atkinson, R.: Kinetics and mechanism of the gas-phase: Reactions of the hydroxyl radical with organic compounds under atmospheric conditions. Chem. Rev. 85, 69-201 (1985)



Considering the chemical structure of isoxaflutole, its degradation in air will mainly be determined by reactions with hydroxyl radicals rather than from reaction with ozone due to the absence of susceptible moieties in the molecule.

The calculations were based on a typical concentration of 1.5 x 10⁶ hydrox radicals/cpf air for daylight and a 12 hour day (average daylight time for a whole year).

II. Results and Discussion

The half-life of isoxaflutole in air was calculated to 1.83 days based on a mean atmospheric by drox \mathcal{A} radical concentration of 1.5×10^6 hydroxyl (\mathcal{O}) radicals/cm³ during 12 hours of dayligh as summarized in Table 7.3.1-1.

		~ <i>C</i> _
	(AODUUN)	
Table 7.3 1-1. Half-life of isovafilitole in air	(ACPW LN°	versa:91
rable 7.0.1 1. Han me of isoxunutore in an		

Parameter	Valor in the second
Time	(hours/da)
OH radical concentration	(radicals/cm) $(radicals/cm)$ $(radicals/cm)$
OH rate constant	$(c_{0} x m) e_{cule_{1}}^{2} x s^{-1}) = \sqrt{5.85} 10^{-12} $
Half-life	(hours) 5 6 7 7 21.9 7 5
\$ }	(da\$s)
- A	

Isoxaflutole can therefore be considered susceptible for degradation processes in air by reactions with photolytically generated bydroxyl radicals. These processes are expected to contribute significantly to the overall degradation of the substance in the atmosphere.

III. Conclusion

Being susceptible to photolytic reactions if the atmosphere it can be concluded that nor isoxaflutole is available for long range transport neither has a potential for accumulation in the atmosphere. Furthermore, an overall potential for accumulation occurrence of isoxaflutole in the atmosphere is additionally limited due to the low vapor pressure of the substance (3.2×10^{-7} Pa at 20°C).

1	
Report: 🖉 🏾	u;
Title:	RPA202248: Estimation of degradation by photo-oxidation in air Model calculation
	according to Atkinson
Report No:	C0088405
Document No(s):	Report pcludes frial Nos.:
	CX/03/080
	M-224920-01-1
Guideline	Deviation not specified
GLP/GROP:	

Executive Summary

The half-life time of RPA 202248 in air was estimated on the basis of structure-activity relationships for the degradation of chemicals in the atmosphere developed by Atkinson and transferred into the computer program AOPWINTM (v 1.91).

The half-life of RPA 202248 in air was estimated to 0.62 to 1.42 days based on a typical day fight \bigcirc concentration of 1.5 x 10⁺⁶ hydroxyl radicals/cm³ air and a 12 hour day.

The calculations showed that RPA 202248 is susceptible for reactions of hydroxyl radicals with various parts of the molecule thus contributing to the degradation of the substance in the amosphere.

The compound is thus not available for long-range transport and has neither a potential for accumulation in the atmosphere. Furthermore, the overall potential for occurrence in the atmosphere is additionally limited due to the low vapor pressure of the substance and its use is seed freatment.

I. Material and Methods

Based on an structure-activity relationship (SAR) approach by Atkinson, the half-life and chemical lifetime of isoxaflutole in air was estimated by use of the computer program $AOPWIN^{TM}$ (version 1.91, Syracuse Research Corp.).

The program allows for an estimation of rate constants for the reaction of light-induced hydroxyl radicals with organic chemicals in the atmosphere. In case of presence of olefinic or acetylenic subgroups in the chemical under assessment the calculation includes estimations of rate constants for the reaction of ozone with these subgroups. From estimated rate constants half-lives of the chemical in the atmospheric are calculated based upon average concentrations of hydroxyl radicals. AOPWINTM requires the input of chemical structure *via* a siniles code and an estimate for the concentration of hydroxyl radicals in the atmosphere.

The actual structure of the diketonitrile compound RPA 202248 can be described by at least two different forms that can exist in parallel und rapidly interconvert between these two forms. The forms have an 'encl-type' structural element as indicated by the strong addic character of the molecule with a measured pKa of \mathbb{C}^7 (see KCA 2.8). It should be noted that the commonly used 'all-ketone' structure cannot explain this finding.



Considering these characteristics of RPA 202248, its degradation in air will be determined by reactions with hydroxyl radicals and with ozone with susceptible moieties being present in the molecule for both types of reaction. However, the ozone-induced reactions are slower contributing significantly less to the overall elimination from the atmosphere than reactions with hydroxyl radicals.



The calculations given below were thus based on a typical concentration of 1.5×10^6 hydroxyl radicals/cm³ air for daylight and a 12 hour day (average daylight time for a whole year).

II. Results and Discussion

Half-lives of two enol-forms of RPA 202248 in air were calculated to 0.62 to 1.42 days based on a mean hydroxyl radical concentration in the atmosphere of 1.5×10^6 hydroxyl (Oft) radicals/cm² during 12 hours of daylight as summarized in Table 7.3.54.

Table 7.3.1-1: Hall-lives of e	noi iorins of RFA 202248 ar (AOP WIN, versi.91)
Parameter	Ly go SValiter L C
Time	(hours/day)
OH radical concentration	$(radicals/cm3)$ \sim
Enol form I	
OH rate constant	$(\text{cm}^3 \text{ x} \text{ yrolecule}^1 \text{ x} \text{ s}^1)^{1/2}$ $733 \times 10^{1/2}$ $733 \times 10^{1/2}$
Half-life	(hogos) i i i i i i i i i i i i i i i i i i i
	(days) & S & A 42 & C
Enol form II	
OH rate constant	$(cpn^3 x modecule x s^{-1})$
Half-life	(hours) (hours
	(days) (~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~

RPA 202248 can also be considered susceptible for degradation processes in air by reactions with photolytically generated hydroxyl radicals. These processes are expected to contribute significantly to the overall degradation of th e substance in the atmosphere.

Being susceptible to photolytic seactions in the atmosphere it can be concluded that RPA 202248 is

not available for long-range gransport neither has a potential for accumulation in the atmosphere. Furthermore, an overall potential for an occurrence of RPA 202248 in the atmosphere is additionally limited due to the low vapor pressure of the substance (2.6 x 10⁻⁷ Pa at 20°C, see KCA 2.2).

Transport via air

Being a new potential regenerement this had not been addressed in the Dossier submitted or evaluated within the process for Amex I inclusion.

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In view of values for vapour pressure measured being below the triggers of 10⁻⁴ Pa for soil and 10⁻⁵ Pa for plant, no study on transport of the active substance isoxaflutole or RPA 202248 via art is regarded as necessary.

The combination of low half-life in the atmosphere (1.83 days) with a low vapour pressure $(3.22 \times 10^{-7} \text{ Pa at } 20^{\circ}\text{C})$ results in a low value for the Henry constant $(1.84 \times 10^{-5} \text{ Pa x } 10^{-5} \text{ x myle}^{-1} \text{ at } 20^{\circ}\text{C})$ 20°C), indicating very low portioning from water to all Consequently, isoxafluted is clearly not subject to transport via air. Under conditions of the environment the active substance is subject to rapid transformation to form RPA 202248. This metabolite has the potential to form salts soowing even lower potential for volatilization from soil or water surfaces than the parent active substance

CA 7.3.3 Local and global effects Being a new potential requirement this had not been addressed in the Dossier submitted of environment the process for Annex I inclusion

Isoxaflutole is applied to a limited number of crops (i.e. majze only at a two field application rate. Following application residues are rapidly degraded in combination with low potential for volatility these aspects indicate that residues are not present under outdoor conditions to form a risk for the environment short-term or long-term to set effects at local or global level.

The potential for local effects from use of soxal tole is considered in risk assessments performed following its use under field conditions in particulation considering factors like spray drift. The combination of exposure assessments with potential effects pleasured in soil and surface water do thus cover the environmental compartments of interest. In contrast and since there is no aerial application envisaged, air is not a compartment regarded to be major compartment of potential for isoxaflutole occurrence following its intended use in the field.

The setting of global effects like contributions to global warming potential (GWP), ozone depleting potential (OPD), photochemical ozone creation potential (POCP) would require a high probability for the molecule assessed of evaporate and this occup in the gas phase. This probability can be expressed by the volatility in terms of the vapour pressure and the Henry constant. The very low potential of isoxaflutore residues to accur in the atmosphere has been addressed before under KCA 7.3.2.

Any accumulation in the proposphere would require high volumes of active substance applied and a significant volatility combined with persistence in the gas phase. The latter characteristic has been addressed under KCA 7.3 to result in no long-term persistence of isoxaflutole residues in the atmosphere.

An activitication potential (AP) would require the generation of acidifying gases like sulfur dioxide or nitrogen og des in a free form. An eutrophication potential (EP) would require the generation of ammonia or phosphorous compounds acting as nutrients.

There are no indications that the degradation of isoxaflutole residues in the environment via biological of physico-chemical processes would result in products that have a potential for acidification or



eutrophication of the environment. Even when this would be the case and to set a potential effect this would require amounts of active substance applied in the field being several orders of magnitude higher in comparison to the low seasonal application rates and from use in maize crops only.

CA 7.4 Definition of the residue

CA 7.4.1 Definition of the residue for risk assessment

The route and rate of degradation of isoxaflutole had been investigated after application of radiolabeled active substance to various soil, ground water and surface water test systems in the laboratory. Following the observation of metabolites and transformation products above the trigger values set in the relevant tests, these are potential residues to occur in the environment thus to be considered in the corresponding environment risk assessments.

Residue definition for soil:

Besides the parent compound isoxaflutole metabolites RPA 202248 (AE 0540092) and RPA 203328 (AE B197555) had been considered in the existing environmental risk assessments due to their occurrence at >10% of AR in aerobic soil degradation tests.

to those addressed already. 🔊 🐇

Residue definition for ground water:

The risk assessment for ground water includes by default all components defined for the risk assessment in soil which is the active substance soxaflutole and metabolites RPA 202248 and RPA 203328.

Residue definition for surface water:

The risk assessment for surface water includes by default the active substance isoxaflutole and those components defined for risk assessment of soil and ground water, i.e. metabolites RPA 202248 and RPA 203228.

In the existing surface water, box as δ such a maximum in water/sediment tests.

No additional transformation products of isoxaflutole were observed in sterile buffer hydrolysis and phototysis, tests on mineralization in surface water or sediment/water tests that would require consideration in surface water risk assessment

CA 7.4.2 Definition of the residue for monitoring

Following rise assessments in soil, ground water and surface water according to the GAP defined, the environmental safety of all components under assessment could be demonstrated according to the requirements set.

It is therefore justified to define the parent compound isoxaflutole and its metabolite RPA 202248 as the relevant residue for monitoring in soil, ground water and surface water.

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CA 7.5 **Monitoring data**

Isoxaflutole

CA 7.5 Monitoring data No formal monitoring program was requested or required to address this point for isoxaflutole or residues in soil and water in the EU. Moreover and specifically for isoxaflutole, there are populished data from formal monitoring residues in soil and water in the EU. Moreover and specifically for isoscallutole, there are no published alla from formal apointoring programs outside BCS available that would indicate a specific concern or findings of isoscallutole and its residues in remote environmental areas not being splicet to the infended use.



List of metabolites observed in environmental fate testing

O U In the original study reports on biotic or abiotic transformation of isoxaflutole the metabolites are denominated by different synonyms. In order to present a common system of nomenclature for the evaluation in the dossier a list of metabolites observed in environmental fate testing is included





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